



Supporting Information

for

Glycerol photoelectrochemical oxidation reaction at carbon nitrides/BiVO₄ materials

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Beilstein J. Nanotechnol. **2026**, *17*, 806–817. [doi:10.3762/bjnano.17.57](https://doi.org/10.3762/bjnano.17.57)

Additional figures

1 – Bandgap energy estimation

The Tauc equation is given by:

$$(\alpha h\nu)^{1/n} = (h\nu - E_g) \quad (\text{S1})$$

in which α is the absorption coefficient (proportional to the absorbance), E_g is the optical bandgap energy, $h\nu$ is the energy of the incident photons, and n represents the type of electronic transition. Herein, $n = 1/2$ was used for direct allowed transitions and $n = 2$ for indirect allowed transitions.

2 – Experimental Setup

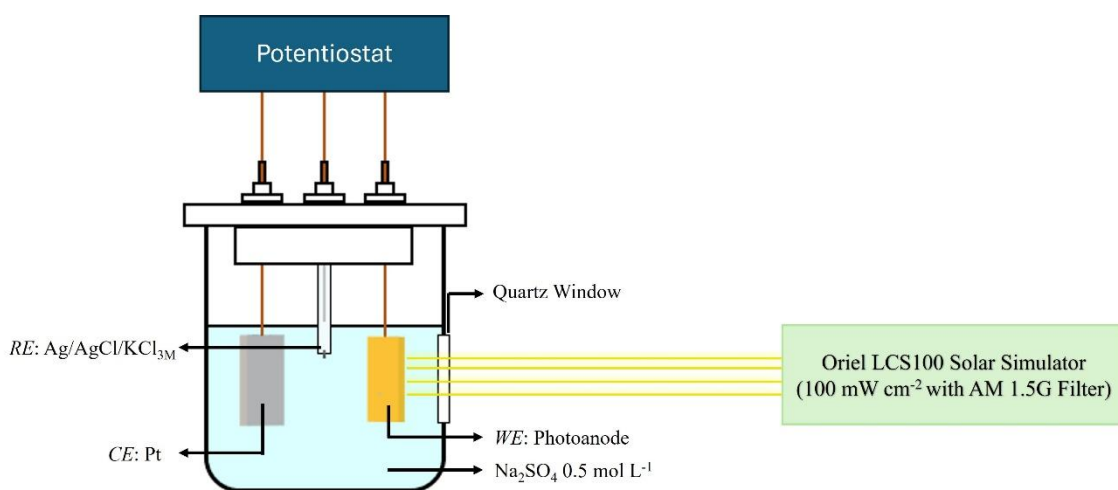


Figure S1: Experimental setup employed for photoelectrochemical experiments.

3 – Physical Characterization

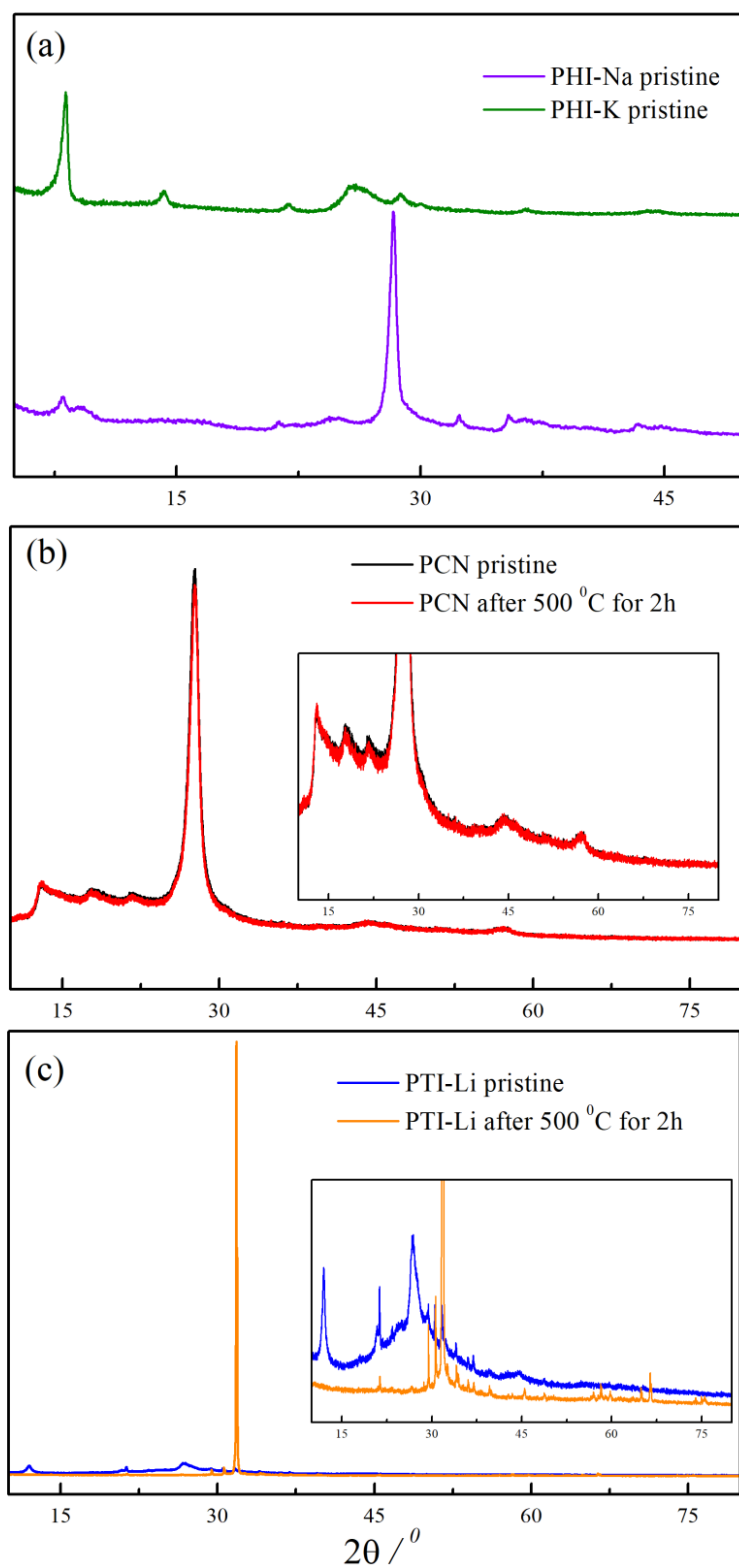


Figure S2: X-ray diffraction patterns for PHI-Na and PHI-K (a), PCN (b), and PTI-Li (c). PCN and PTI-Li XRD were analyzed for both pristine materials and after thermal treatment at 500 °C for 2 h.

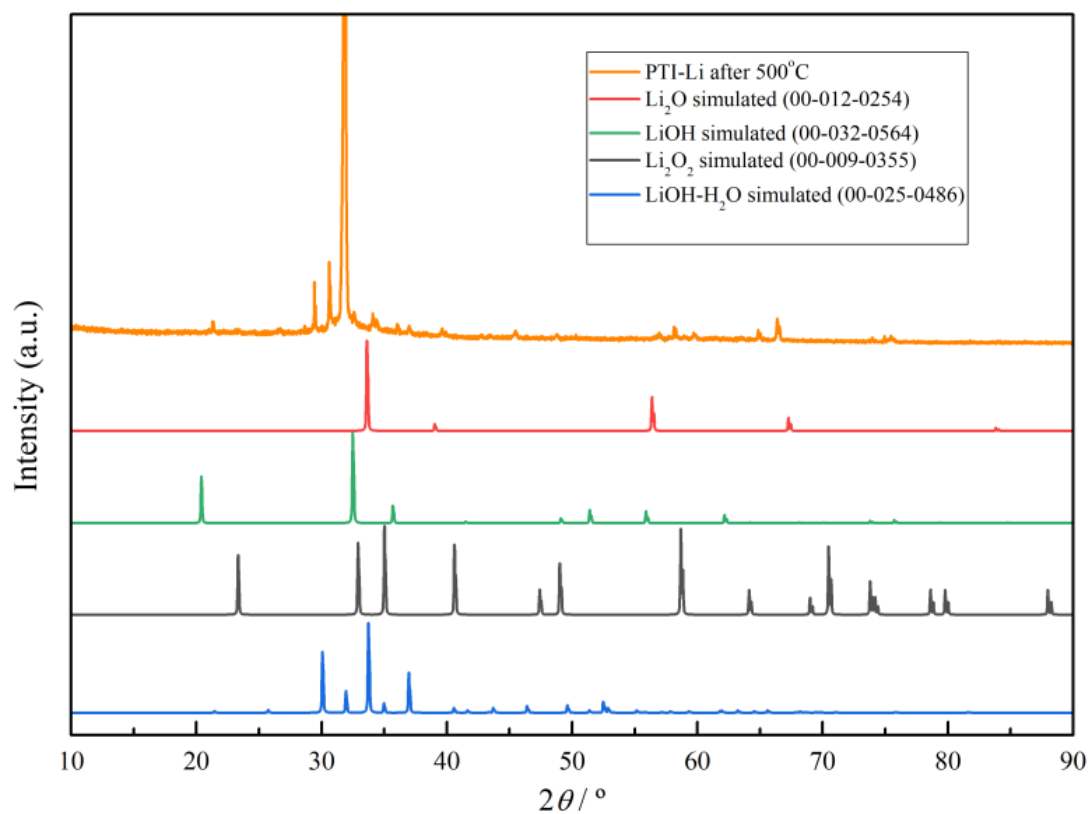


Figure S3: X-ray diffraction patterns for PTI-Li after thermal treatment at 500 °C for 2 h and for some lithium (hydro)oxides.

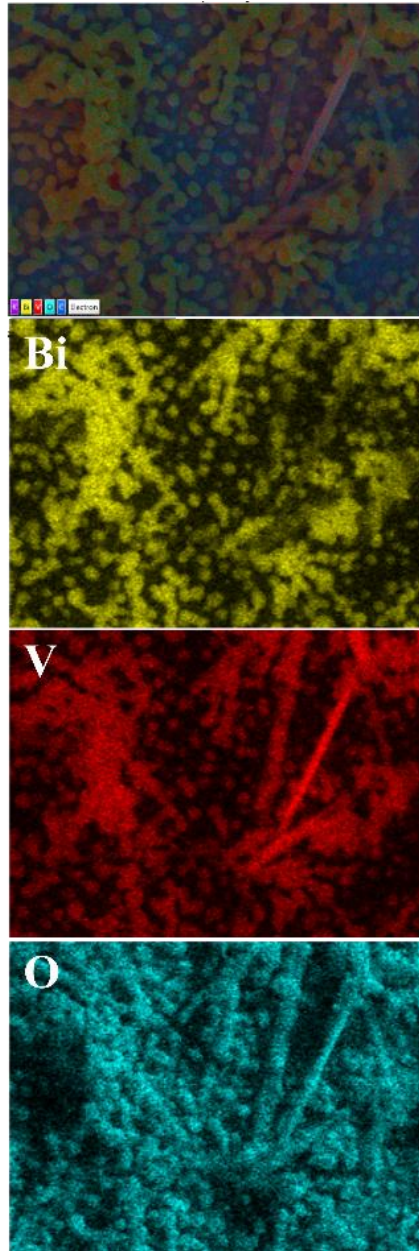


Figure S4: SEM-EDS for PTI-Li/BiVO₄.

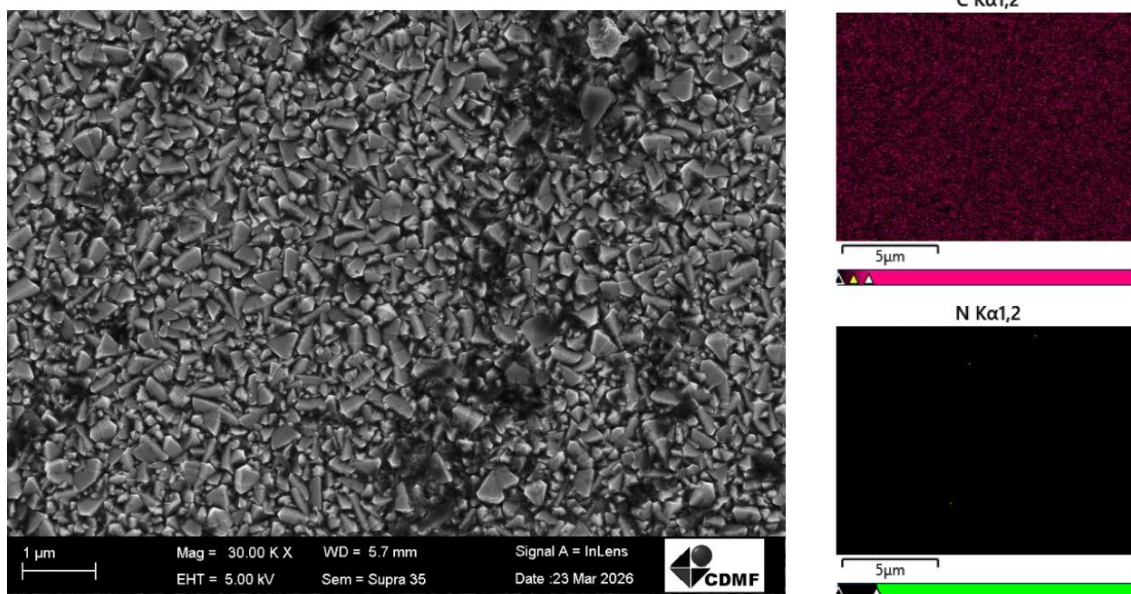


Figure S5: SEM images of PCN onto FTO (left) and carbon and nitrogen elemental mapping obtained by SEM/EDS (right).

4 – Electrochemical Characterization

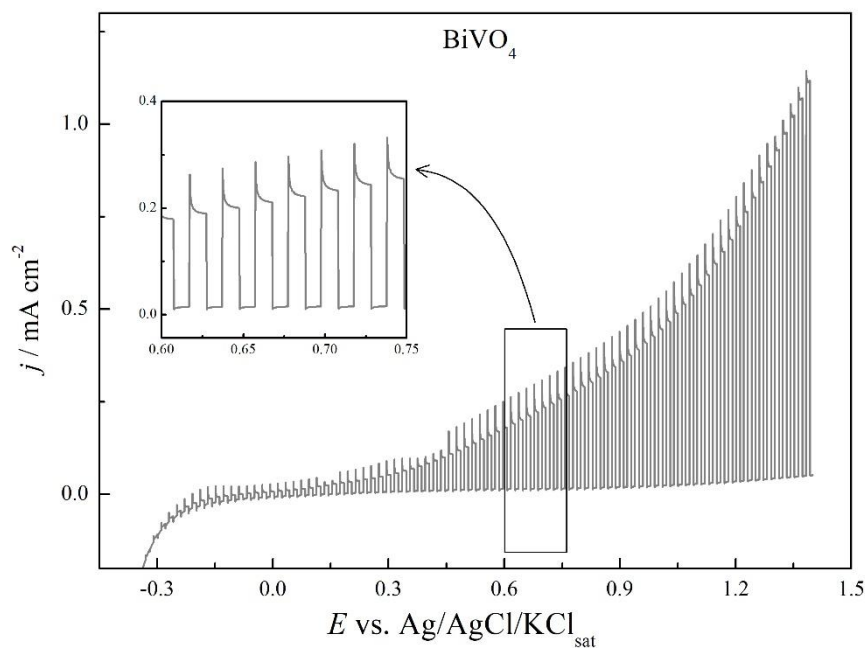


Figure S6: Linear sweep voltammetry (LSV) at 0.05 V·s⁻¹ on chopped illumination for the BiVO₄ film. Solution condition: Na₂SO₄ (0.5 mol·L⁻¹, pH 6.8).

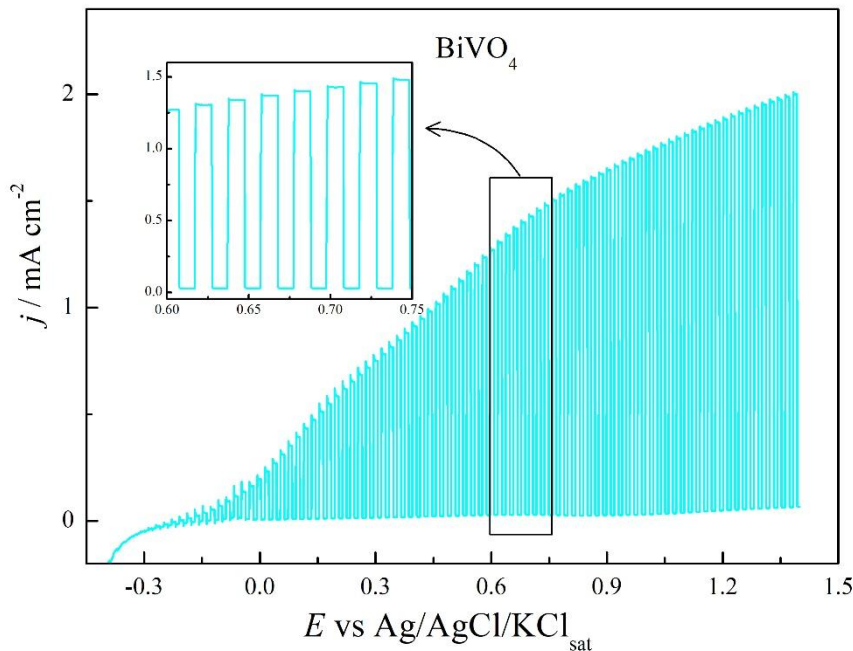


Figure S7: LSV at 0.05 V·s⁻¹ on chopped illumination for the BiVO₄ film. Solution condition: Na₂SO₄ (0.5 mol·L⁻¹, pH 6.8) + glycerol (1.0 mol·L⁻¹).

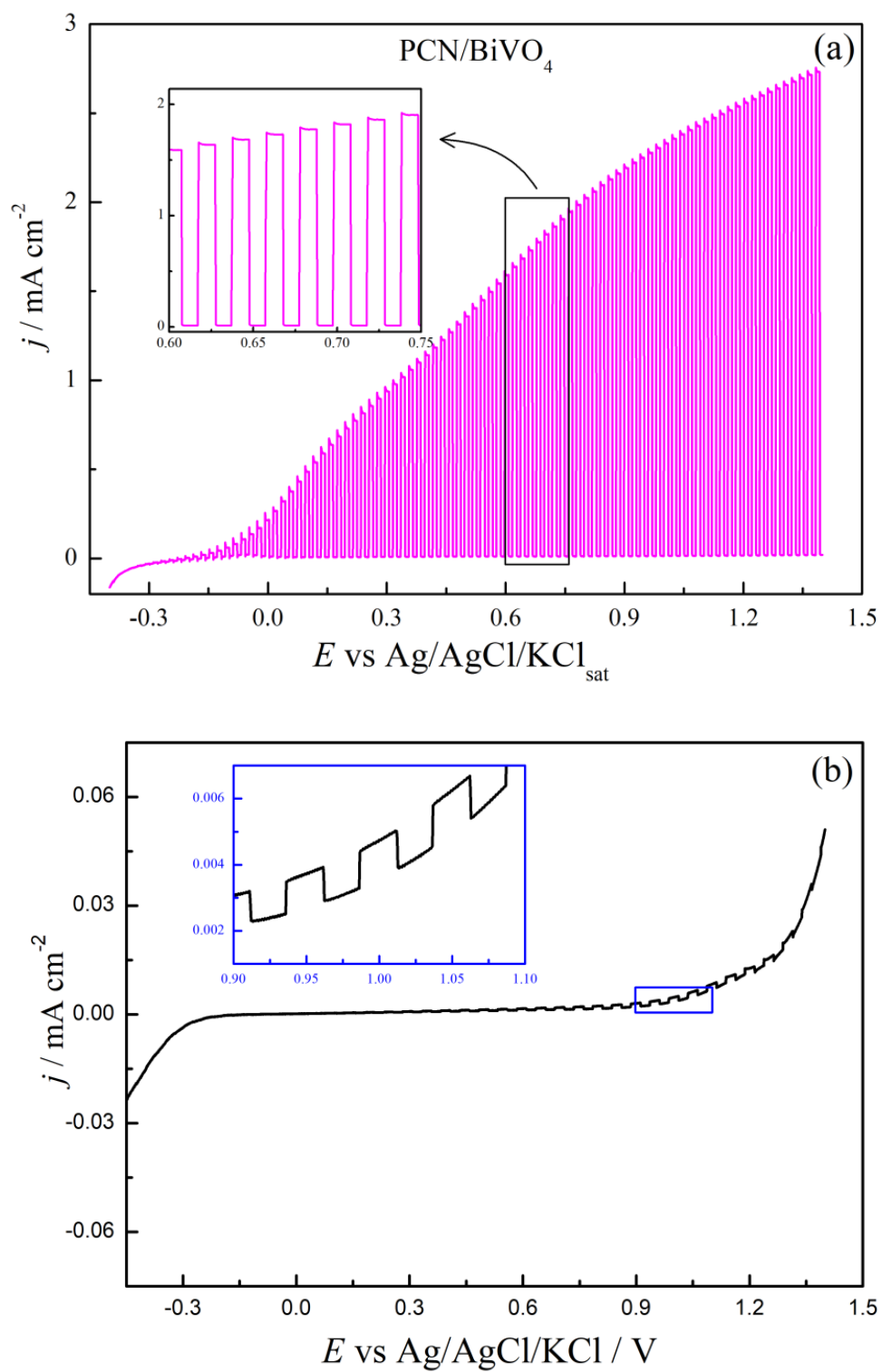


Figure S8: LSV at $0.05 \text{ V}\cdot\text{s}^{-1}$ on chopped illumination for PCN/BiVO₄ (a) and for PCN films (b). Solution condition: Na₂SO₄ ($0.5 \text{ mol}\cdot\text{L}^{-1}$, pH 6.8) + glycerol ($1.0 \text{ mol}\cdot\text{L}^{-1}$).

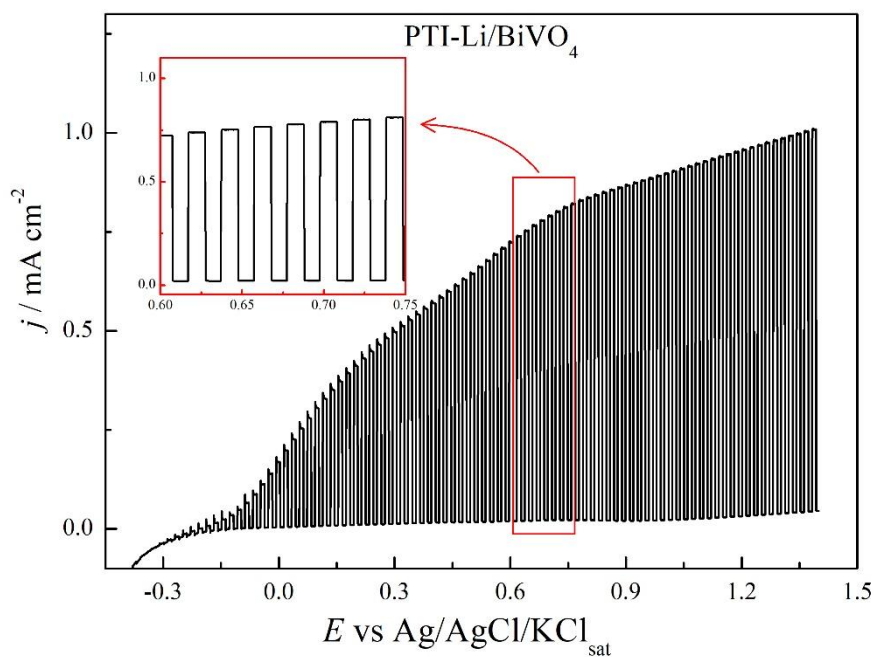


Figure S9: LSV at $0.05 \text{ V}\cdot\text{s}^{-1}$ on chopped illumination for the PTI-Li/BiVO₄ film. Solution condition: Na₂SO₄ ($0.5 \text{ mol}\cdot\text{L}^{-1}$, pH 6.8) + glycerol ($1.0 \text{ mol}\cdot\text{L}^{-1}$).

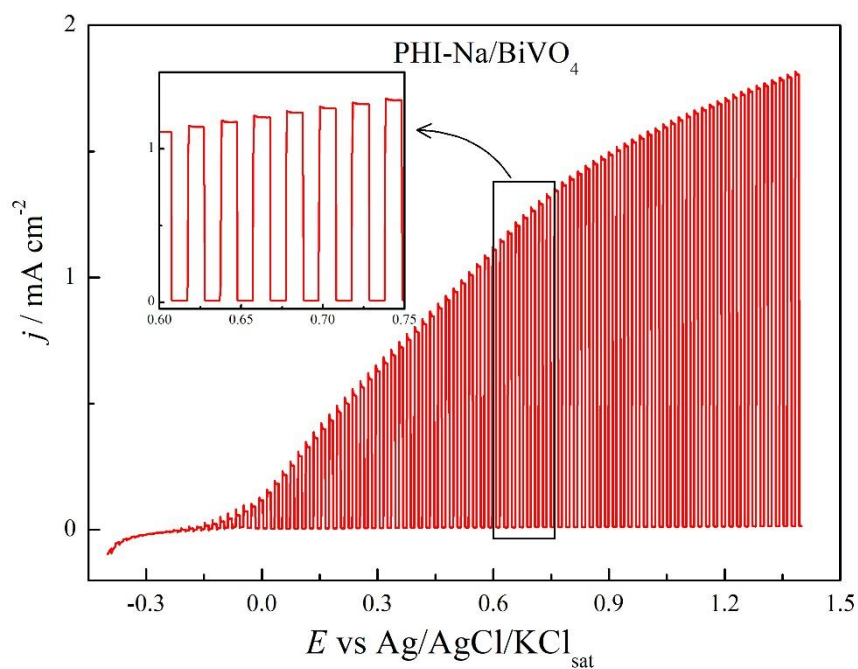


Figure S10: LSV at $0.05 \text{ V}\cdot\text{s}^{-1}$ on chopped illumination for the PHI-Na/BiVO₄ film. Solution condition: Na₂SO₄ ($0.5 \text{ mol}\cdot\text{L}^{-1}$, pH 6.8) + glycerol ($1.0 \text{ mol}\cdot\text{L}^{-1}$).

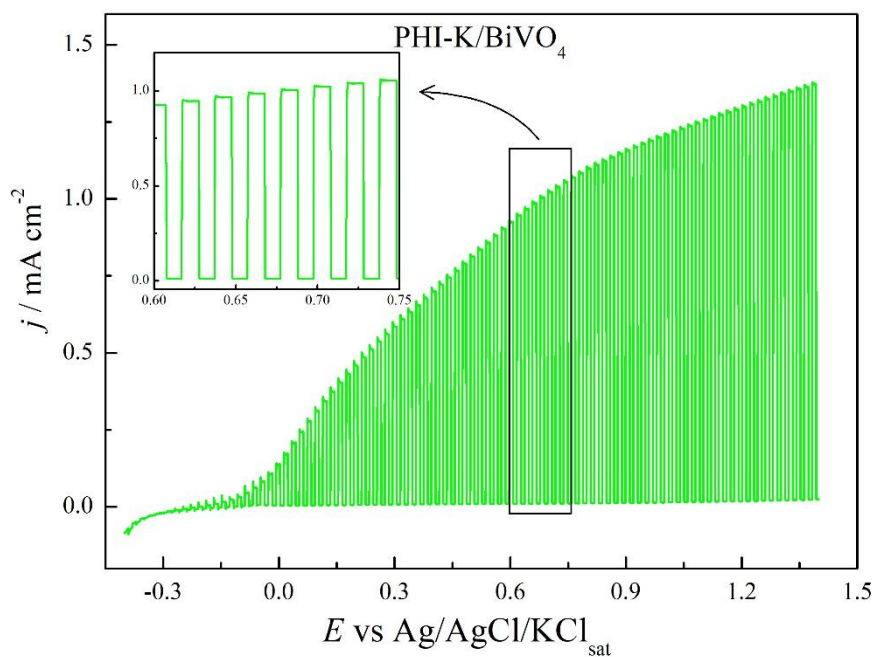


Figure S11: LSV at $0.05 \text{ V}\cdot\text{s}^{-1}$ on chopped illumination for the PHI-K/BiVO₄ film. Solution condition: Na₂SO₄ ($0.5 \text{ mol}\cdot\text{L}^{-1}$, pH 6.8) + glycerol ($1.0 \text{ mol}\cdot\text{L}^{-1}$).

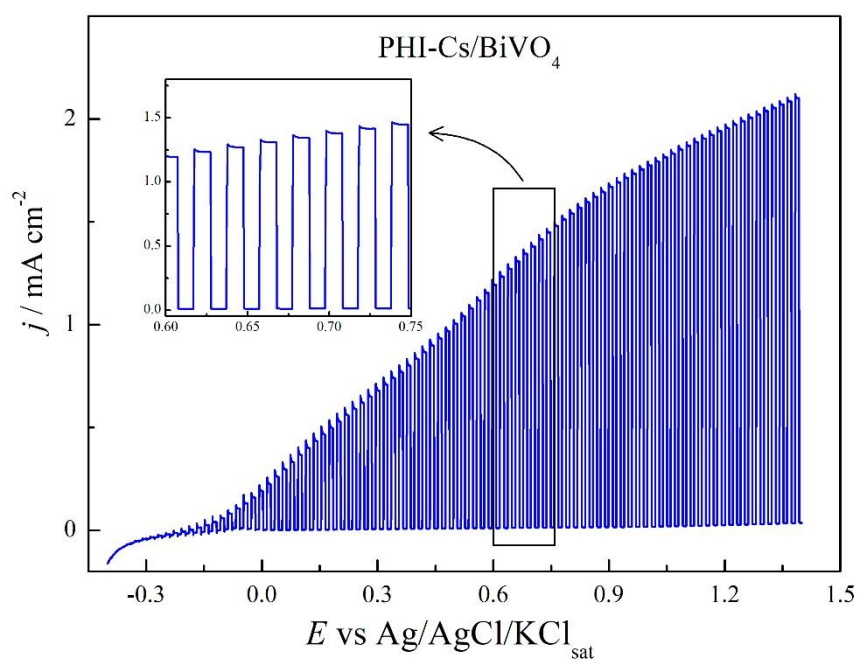


Figure S12: LSV at $0.05 \text{ V}\cdot\text{s}^{-1}$ on chopped illumination for the PHI-Cs/BiVO₄ film. Solution condition: Na₂SO₄ ($0.5 \text{ mol}\cdot\text{L}^{-1}$, pH 6.8) + glycerol ($1.0 \text{ mol}\cdot\text{L}^{-1}$).

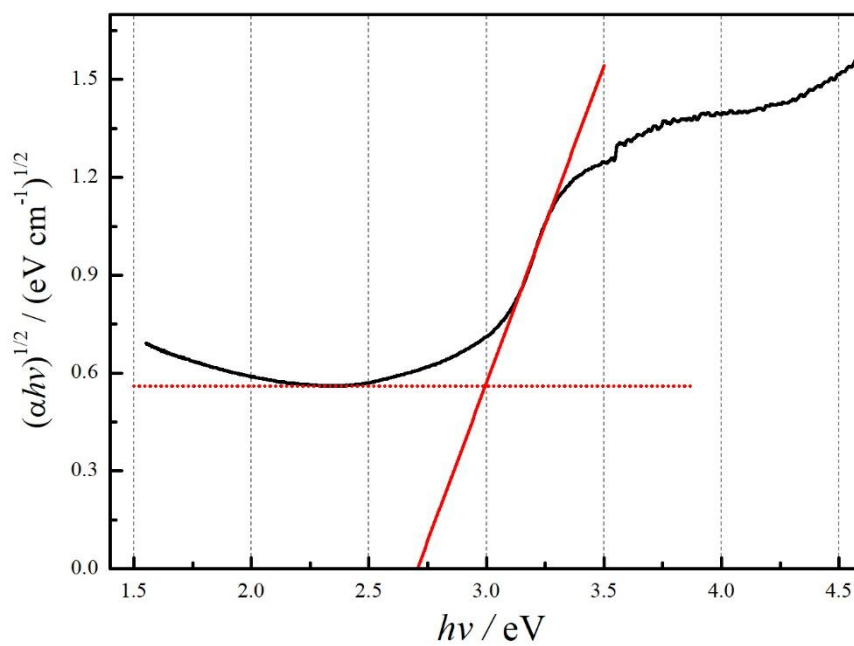


Figure S13: Tauc plot from UV-vis spectrum for PTI-Li after thermal treatment at 500 °C for 2 h.