



Transformation of the cyclohexane ring to the cyclopentane fragment of biologically active compounds

Natalya Akhmetdinova¹, Ilgiz Biktagirov² and Liliya Kh. Faizullina^{*1}

Review

Open Access

Address:

¹Ufa Institute of Chemistry, UfRC RAS, 450054, Ufa, 71, Oktyabrya Ave. Russian Federation and ²«Bazis-Terra» LLC, 20a, Golubyatnikova St., Kazan, 420094, Russian Federation

Email:

Liliya Kh. Faizullina^{*} - sinvmet@anrb.ru

^{*} Corresponding author

Keywords:

biological activity; cyclopentane/enone; rearrangements; ring/cycle contractions; total synthesis

Beilstein J. Org. Chem. **2025**, *21*, 2416–2446.

<https://doi.org/10.3762/bjoc.21.185>

Received: 13 February 2025

Accepted: 16 October 2025

Published: 06 November 2025

Associate Editor: K. N. Allen



© 2025 Akhmetdinova et al.; licensee Beilstein-Institut.

License and terms: see end of document.

Abstract

The review is devoted to strategies for contraction of six-membered cycles in the synthesis of functionalized cyclopentane/enones, which are biologically active compounds. The main synthetic methods of ring contraction (ozonolysis–aldol condensation, ozonolysis–Dieckmann reaction, Baeyer–Villiger cleavage–Dieckmann reaction) and rearrangements (benzil, semipinacol, with the participation of thallium- and iodine-based oxidants, photochemical, Wolff, Meinwald, Wagner–Meerwein and Favorskii) are presented. The review summarizes literature data covering the last 12 years, with some exceptions of earlier works due to the importance of the published information.

Introduction

The functionalized cyclopentane/enone fragment is part of the structure of a large number of natural and synthetic biologically active compounds, including prostaglandins, terpenoids, alkaloids, steroids and carbanucleosides. The spectrum of biological activities of the listed classes is wide and unique in efficacy. Natural prostaglandins are polyoxygenated derivatives of a hypothetical twenty-atom prostanoid acid, the C8–C12 atoms of which are contained in a cyclopentane ring with two attached side chains – of seven (α -chain) and eight (β - or ω -chain) carbon atoms. Their main physiological role is to maintain the homeostatic harmony of the organism [1,2]. Among the terpenoids containing cyclopentane in their structure, we should mention jatrophane or latirane diterpenoids, in which the five-

membered cycle is annelated with a functionally saturated dodecyl fragment [3,4], as well as iridoids – bicyclic compounds with a central cyclopentane unit [5]. Estrone (folliculin, estra-1,3,5(10)-trien-3-ol-17-one), a steroid containing a cyclopentane annelated with octahydrophenanthrene, is a natural follicular hormone essential for normal development of the female body [6]. An example of an alkaloid whose structure includes cyclopentane is lappaconitine, which is noteworthy because its hydrobromide is the active ingredient in the highly effective antiarrhythmic drug allapinine [7].

Syntheses of cyclopentanoids usually are based on cyclopentannulation reactions:

- transformations accompanied by cyclization [8];
- transformation of compounds containing a cyclopentane ring [9];
- contraction of large cycles to cyclopentane derivatives.

Ring contraction reactions are among the most useful strategic transformations for the construction of carbocyclic and heterocyclic cyclopentanes. Six-membered carbocycles are convenient starting compounds for the synthesis of cyclopentanoids due to their widespread occurrence in nature and their synthetic availability. Undoubtedly, the development of practical approaches to the synthesis of known compounds and the synthesis of new biologically active functionalized compounds containing a cyclopentane/enone fragment through transformation of a cyclohexane/ene ring is an urgent task for synthetic chemists.

This review summarizes information on cyclohexane/ene ring contraction. The structure of the review includes examples of simple transformations (ozonolysis–aldol condensation, ozonolysis–Dieckmann reaction, and Baeyer–Villiger cleavage–Dieckmann reaction) and rearrangements (photochemical, benzil, semi-pinacol, Wolff, Meinwald, Wagner–Meerwein and Favorskii reaction), using oxidants based on thallium and iodine, with a focus on recent works published in the period from 2014 to 2024.

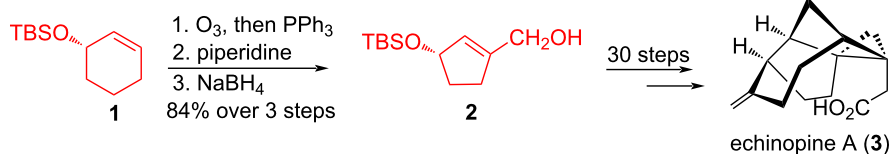
Review

1 Recyclization

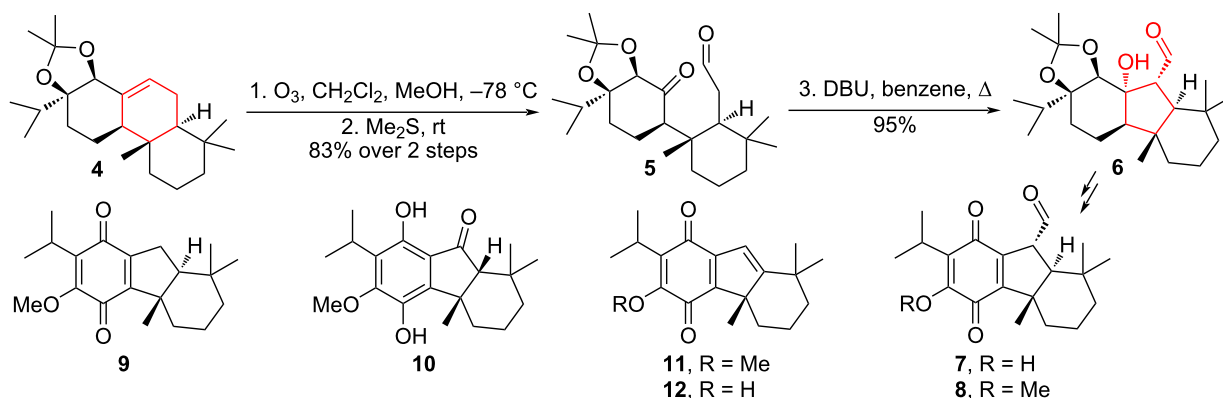
A common method for converting cyclohexene **1** into cyclopentene **2** is the ozonolytic cleavage of the double bond followed by intramolecular aldol condensation of the resulting dialdehyde. This strategy is widely used at the key stage in the formation of a cyclopentane ring in the synthesis of various building blocks, including echinopine A (**3**) [10–12] (Scheme 1).

The ozonolysis–cyclization sequence was used by Alvarez-Manzaneda et al. as a method for ring contraction of the cyclohexene ring to produce cytostatic taiwaniaquinoids [13–15]. Compound **4**, derived from (–)-abietic acid, was subjected to an oxidative cleavage reaction to produce ketoaldehyde **5**. Subsequent intramolecular aldol cyclization using 1,8-diazobicyclo[5.4.0]undec-7-ene (DBU) in benzene resulted in high yield of β -hydroxyaldehyde **6**. Through a series of synthetic transformations, the target products (–)-taiwaniaquinones A (**7**), F (**8**), G (**9**), and H (**11**), (–)-taiwaniaquinol B (**10**) and (–)-dichroanone (**12**) were obtained from intermediate **6** (Scheme 2).

An interesting transformation of a cyclohexene fragment into the cyclopentane fragment in Diels–Alder adducts of levoglucosenone (LG) and 1,3-dienes is presented in [16]. Diels–Alder adducts of LG and 1,3-dienes, containing in their structure frag-



Scheme 1: Ozonolysis–cyclization sequence in the synthesis of echinopine A (**3**).



Scheme 2: Ozonolysis–cyclization sequence in the synthesis of taiwaniaquinoids **7**–**12**.

ments of cyclohexene and a carbohydrate residue, are attractive objects for investigating the possibility of using them in the synthesis of compounds of iridoid topology. Iridoids belong to the class of terpenoids and are secondary metabolites of plants, as well as multi-tasking objects [5,17,18]. Studies on the pharmacological properties of iridoids have shown their high potential for biological activity, such as anti-inflammatory, antimicrobial, immunomodulatory, neuroprotective, hepatoprotective, hypoglycemic, hypolipidemic, antioxidant, antispasmodic, antitumoral, antiviral, and antiallergic. The breadth of the biological action of these cyclopentanoids is explained to some extent by their structural diversity. Figure 1 shows the general formula of iridoids.

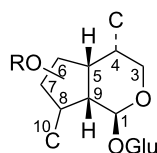


Figure 1: Iridoid skeleton.

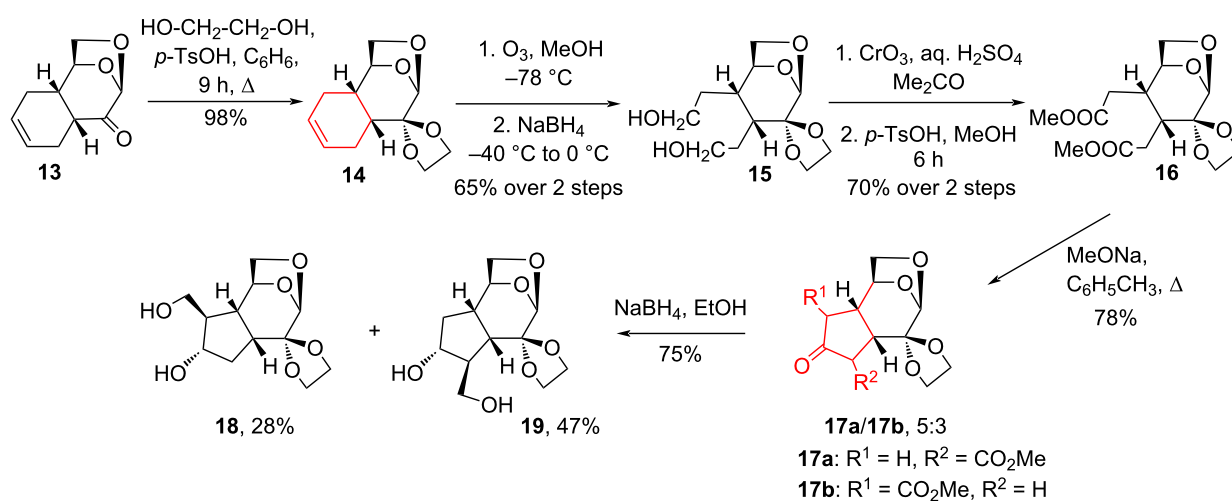
The authors [16] have established regiocontrol of the effect of the 1,6-anhydro bridge on the Dieckmann condensation of dicarboxylic acid diester **16**, obtained in four steps from **13**, the Diels–Alder adduct of LG and 1,3-butadiene (Scheme 3). The ratio of the resulting regioisomers was 5:3 in favor of the 5-methoxycarbonyl derivative **17a**. The synthesized cyclopentane derivatives **17a,b**, **18**, and **19** may be useful for the preparation of iridoids and other biologically active cyclopentanoids.

An alternative method for the synthesis of cyclopentane derivatives through oxidative cleavage of the double bond in

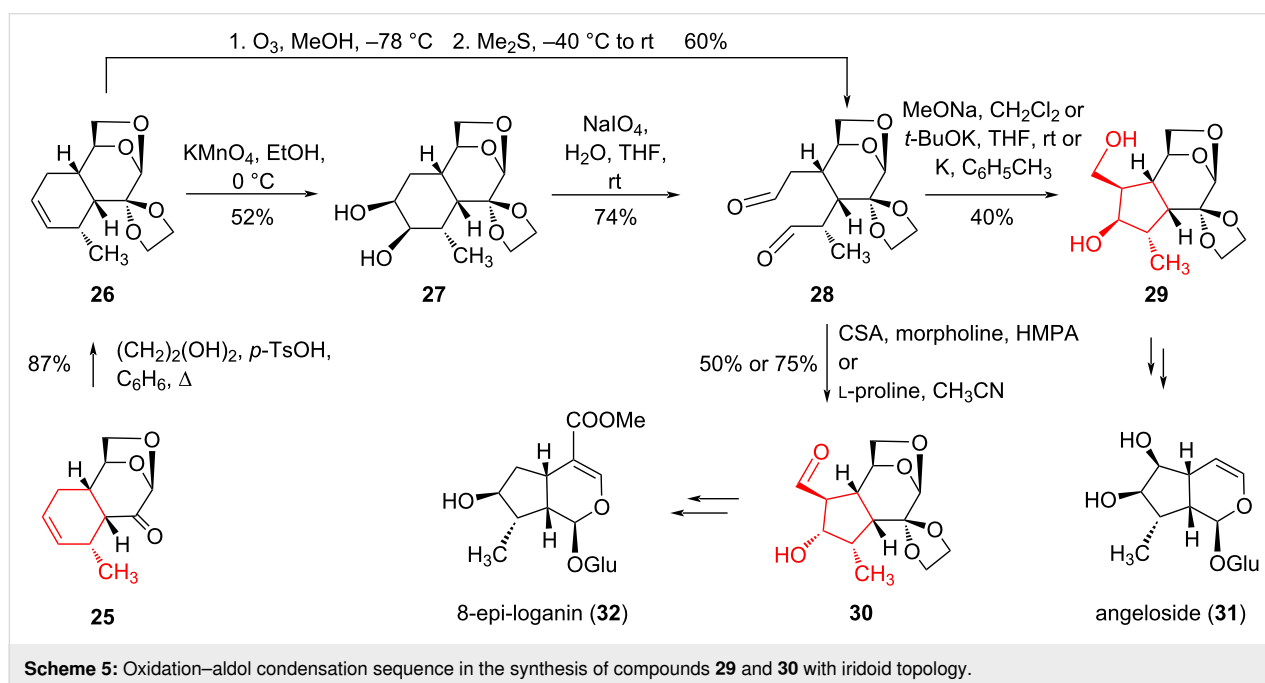
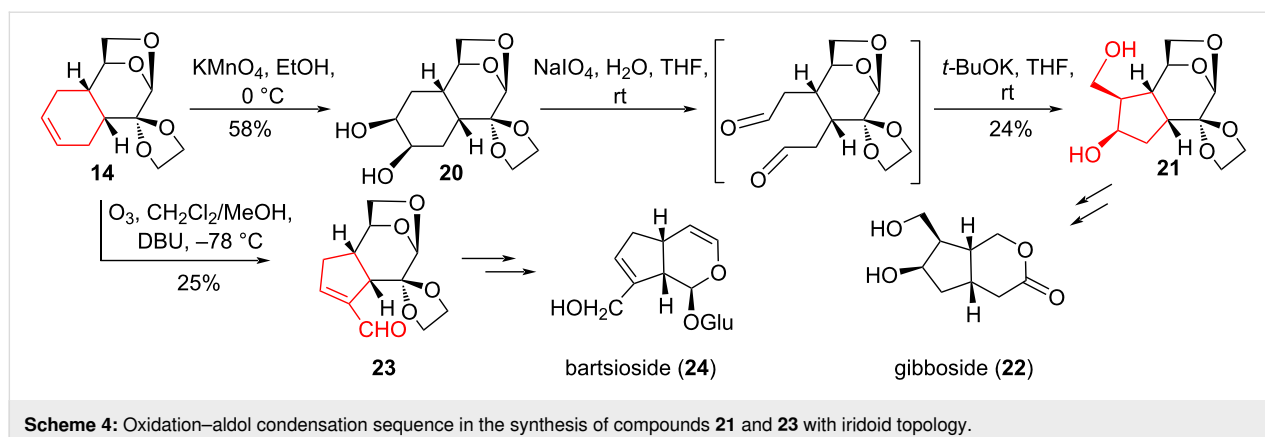
Diels–Alder adducts of LG with 1,3-butadiene and piperylene (3*E*-penta-1,3-diene), followed by an intramolecular aldol condensation of the resulting dialdehydes was described in [19]. The dicarbonyl compound was synthesized from the Diels–Alder adduct **14** between LG and 1,3-butadiene by two methods – vicinal hydroxylation of the double bond followed by periodate cleavage of the *vic*-diols and ozonolysis of the double bond. Alternatively, Wagner oxidation of the double bond in adduct **14** by treatment with KMnO_4 in EtOH at 0 °C afforded vicinal diol **20** with a yield of 58%. The periodate cleavage of the latter under the action of NaIO_4 led to the formation of a labile dialdehyde, which was treated with *t*-BuOK in THF without isolation. As a result of intramolecular aldol condensation, accompanied by the destruction of the main amount of dialdehyde, cyclopentanediol **21** was isolated with a yield of 24%. This synthetic block is promising for the synthesis of gibboside (**22**) [20].

Following the second direction, ozonolysis of the double bond in compound **14** and subsequent treatment of the ozonide with DBU resulted in a stable aldehyde **23**. This aldehyde is a product of intramolecular aldol–croton condensation and can be used in the synthesis of bartsioside (**24**) or its analogues [21,22] (Scheme 4).

The authors [19] performed similar oxidative transformations with Diels–Alder adduct **25** obtained from LG and piperylene, having previously protected its keto group as dioxolane (Scheme 5). Dioxolane **26** was converted to glycol **27** through *vic*-dihydroxylation in moderate yield. Then, dialdehyde **28** was obtained by treatment with NaIO_4 with a yield of 74%. Subsequent intramolecular aldol condensation of dialdehyde **28** with MeONa or *t*-BuOK as base led to the formation of diol **29** with



Scheme 3: Ozonolysis–cyclization sequence in the synthesis of compounds **17a,b**, **18** and **19** with iridoid topologies.

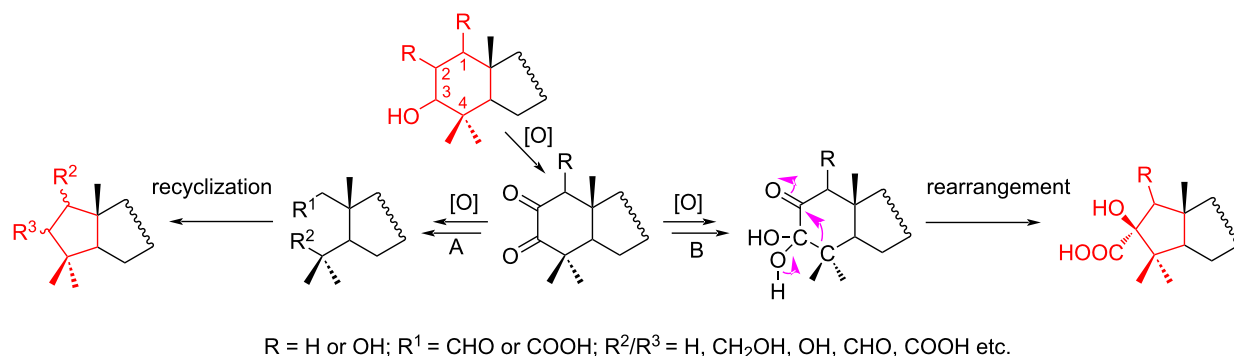


a yield of 40%, which is interesting for the synthesis of angeloside (**31**) [23].

An alternative preparation of diol **29** involved ozonolysis of the double bond in dioxolane **26** at -78°C in methanol, followed by reduction of the ozonide with Me_2S to give dialdehyde **28** and reaction of compound **28** with colloidal potassium in toluene. During aldol condensation in the presence of morpholine–camphorsulfonic acid (CSA) or L-proline, a stable aldehyde **30** was isolated in yields of 50% and 75%, respectively. Decarboxylation and corresponding modifications of the carbohydrate residue in compound **30** resulted in 8-*epi*-loganin (**32**) [24,25]. Thus, as a result of intriguing transformations based on Diels–Alder adducts of LG with butadiene and piperylene, cyclopentane derivatives were obtained, which are promising for the synthesis of natural compounds, particularly iridoids.

One of the promising methods for ring contraction in the absence of multiple bonds in the six-membered ring of terpenoids was proposed by Grishko et al. [26] (Scheme 6). The key step of 2,3-fragmentation was the cleavage of an enolized C–C bond at a keto group, followed by an intramolecular aldol reaction (recyclization or rearrangement). This method is applicable to triterpenoids, such as tirucallane, ursane, oleanane and dammarane types (pathway A), as well as for triterpenoids of the ursane and dammarane types (pathway B) [27].

In [28], a convenient method for ring contraction is described, which is widely used in the synthesis of triterpenoids. In continuation of studies on skeletal and oxidative transformations of triterpenoids from licorice roots (*Glycyrrhiza glabra* L. and *G. uralensis* Fisher) (leguminosae) [29–31], the authors synthesized a derivative with an altered pentacyclic structure, 2,11-



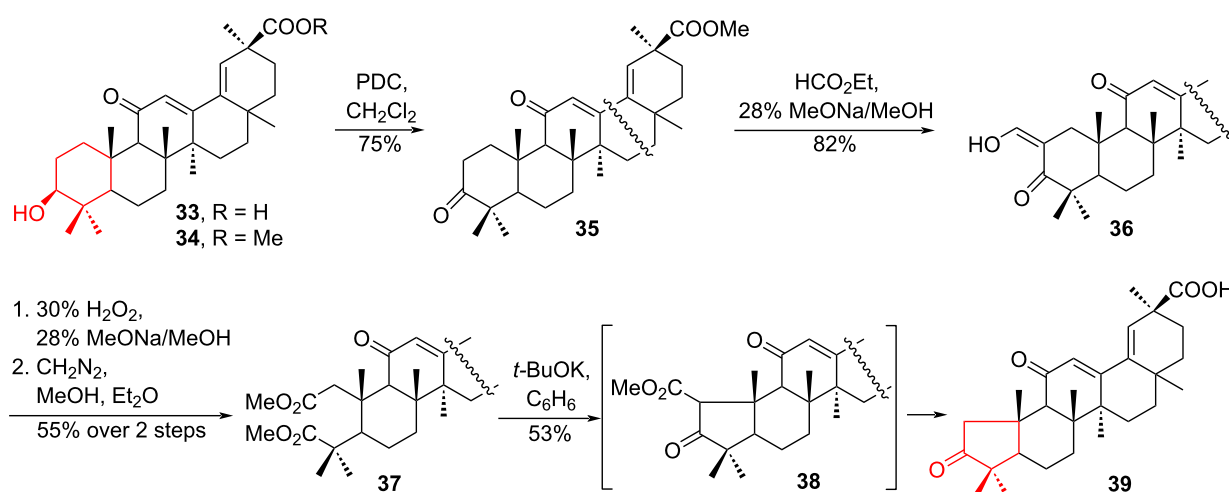
Scheme 6: Method for ring contraction in the absence of a double bond in a six-membered ring of triterpenoids.

dioxo-A-norolean-12,18(19)-dien-30-oic acid **39**, from pharmaceutical 18,19-dehydroglycyrrhetic acid (18,19-dehydro-GLA) **33**. Compound **33** was converted to its methyl ester **34** and oxidized by pyridinium dichromate (PDC) in CH₂Cl₂ to 3-oxo-18,19-dehydro-GLA **35** with a 75% yield. α -Ethylformylation of compound **35** and subsequent oxidation of ketoenol **36** with 30% aqueous H₂O₂ in the presence of a 28% MeONa/MeOH solution resulted in the formation of a diacid, which was then converted into dimethyl ester **37** with a yield of 55%. Refluxing ester **37** with an excess of *t*-BuOK in benzene gave acid **39** with a yield of 53% yield. The excess of base (*t*-BuOK) caused simultaneous decarboxylation and hydrolysis of ester **37**, forming the new nortriterpenoid **39** with a pentacyclic ring A (Scheme 7).

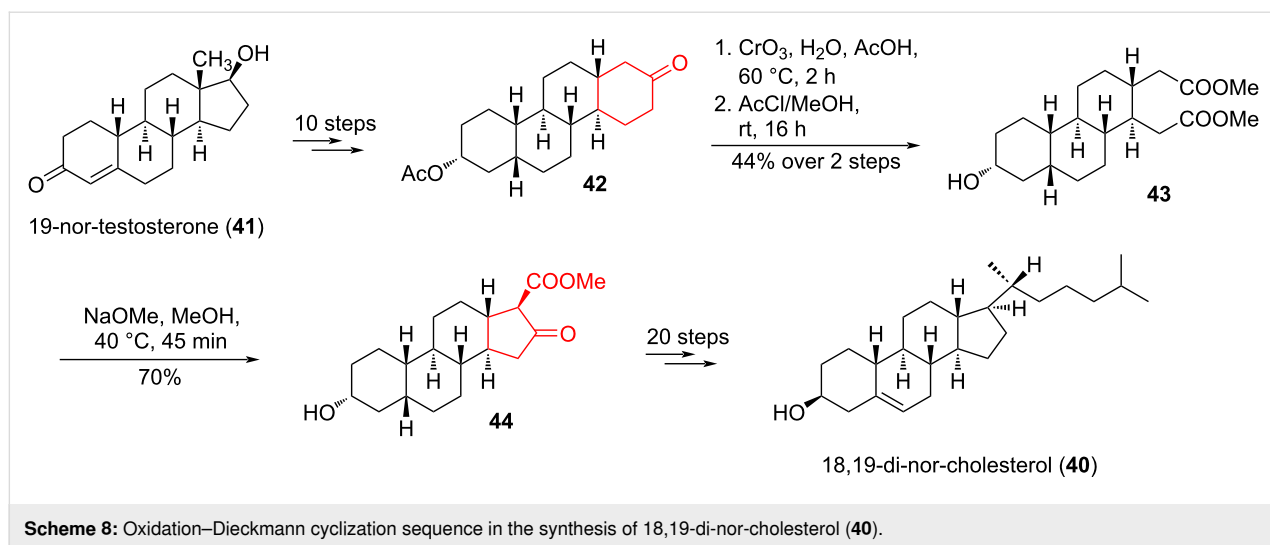
In the work of Rath et al. [32], a similar method was proposed for the interconversion of cycles in the synthesis of the demethylated form of cholesterol, 18,19-di-nor-cholesterol (**40**) from commercially available 19-nortestosterone (**41**)

(Scheme 8). Molecular modeling of 18,19-di-nor-cholesterol showed that cholesterol's opposing rough β -face and smooth α -face play crucial roles in cholesterol's membrane condensing abilities. Specific facial preferences are displayed as cholesterol interacts with different neighboring lipids and transmembrane proteins [33,34]. These biochemical interactions are poorly understood, so the synthesis of a «smoothed» cholesterol analogue provided an opportunity to experimentally measure the significance of the β -face methyl groups.

The attractive synthesis of 3-ethyl-substituted betulinic acid derivatives with high cytotoxic and anti-inflammatory activity was described by Grishko et al. [35]. The 2,3-secotriterpenic 3-ethyl-3-ketone **46** was obtained in two steps from lupane α -ketoxime **45** by Grignard reduction alkylation, followed by a Beckmann fragmentation of the C2–C3 bond of the intermediate 3-ethyl-substituted hydroxyimino ketone in the SOCl₂-CH₂Cl₂ system. The introduction of a carbonyl substituent into the isopropylidene fragment of ketone **46** was achieved either

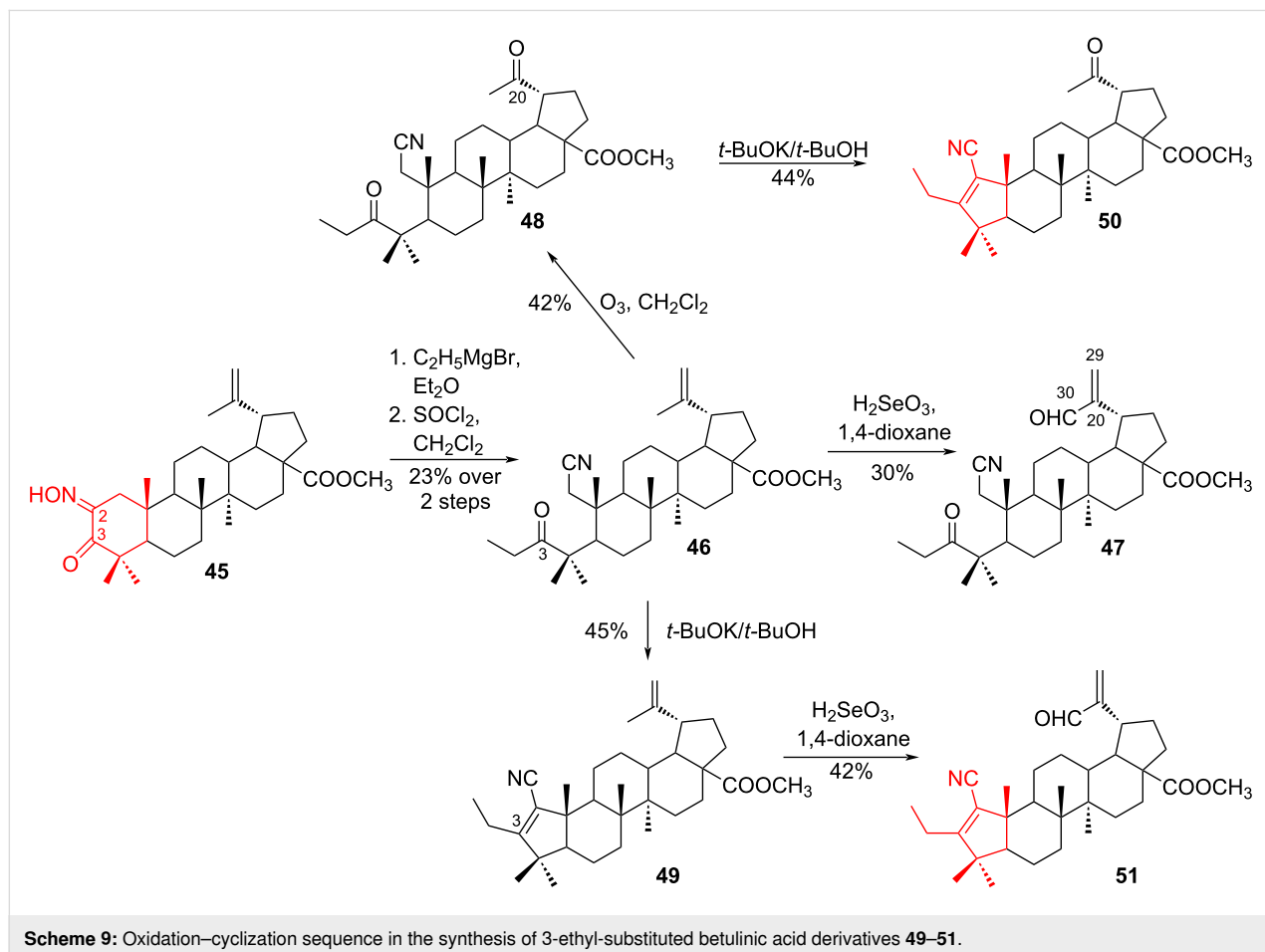


Scheme 7: Oxidation–Dieckmann cyclization sequence in the synthesis of a new nortriterpenoid **39**.



by allylic oxidation using H_2SeO_3 -dioxane system to form the C30 aldehyde **47**, or by the ozonolytic cleavage of the double bond between C20 and C29 to produce 20-methyl-3-ethylidiketone **48** [36]. Intramolecular nitrile–anionic cyclization of ketone **46** or diketone **48** under conditions of basic catalysis

proceeded via the oxo-nitrile mechanism, with formation of the corresponding A-pentacyclic 3-ethyl α,β -alkenenitriles **49** and **50** with yields of 44% to 45% (Scheme 9). Cytotoxic screening was performed using the MTT assay with a panel of 7 human tumor cell lines, and showed that the synthesized



triterpenoids **47** and **51** showed high cytotoxic activity (IC₅₀ 1.38–15.91 μM).

2 Rearrangements with 1,2-carbon migration

Rearrangements with 1,2-carbon migration are widely used in the synthesis of natural compounds [37]. These include benzylic acid and semipinacol-type rearrangements, reactions promoted by thallium(III) and iodine(III), and Wolff rearrangement.

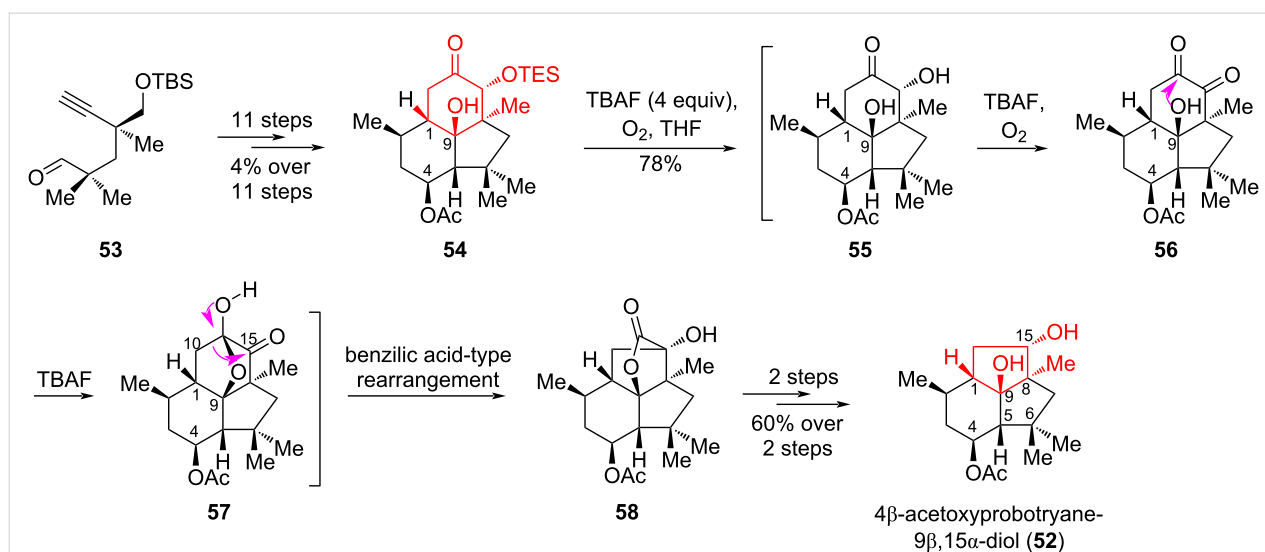
2.1 Benzylic acid and semipinacol-type rearrangements

The strategy of ring contraction using the benzylic acid-type rearrangement was used by Zhang et al. [38] for the asymmetric synthesis of 4β-acetoxypobotryane-9β,15α-diol (**52**). This compound contains a sterically compact [6-5-5] tricyclic skeleton, as well as a highly strained *trans*-fused bicyclo[3.3.0]octane ring system and seven contiguous stereo-centers. A structurally interesting diol **52** was diastereoselectively obtained using a linear sequence of 14 steps starting from the readily available aldehyde **53** [39] (Scheme 10). The key steps in this synthesis are based on an asymmetric rhodium-catalyzed [4 + 2] cycloaddition reaction [40], followed by a unique benzylic acid-type rearrangement under very mild conditions [41]. A step-by-step mechanism for the benzylic acid-type rearrangement of compound **54** was proposed by the authors [38]. It involves deprotection of the TES group in **54** using tetrabutylammonium fluoride (TBAF) to give **55**. In the presence of O₂ and TBAF, **55** spontaneously oxidized to **56**. Intramolecular hemiketalization of **56** resulted in the formation of hemiketal **57**, followed by rearrangement to produce lactone **58** with a yield of 78%. The last stage of the synthesis proceeds through a chemo- and diastereoselective reduction of lactone **58**, contain-

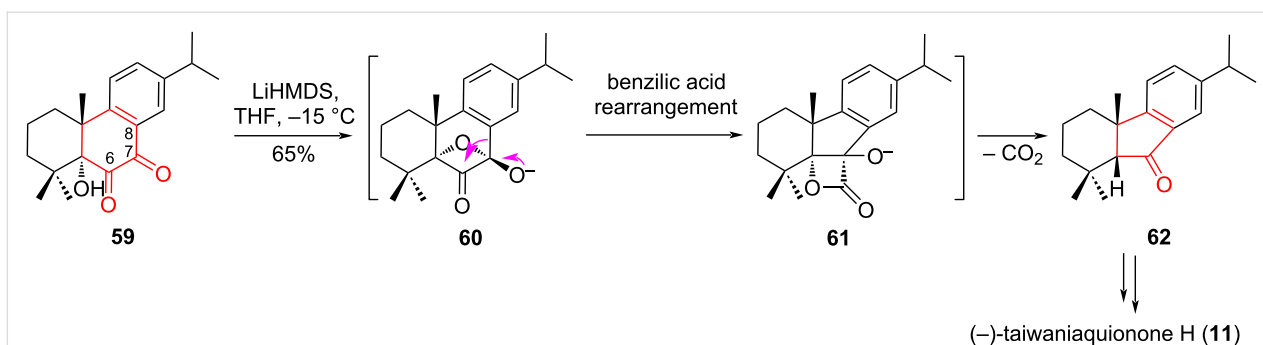
ing the desired *trans*-fused bicyclo[3.3.0]octane ring system, leading to the target 4β-acetoxypobotryane-9β,15α-diol (**52**).

An alternative approach to the synthesis of taiwaniaquinoids was applied by Gademann et al. [13,42] for the transition from an abietane structure to a five-membered system using a rearrangement of benzylic acid. The strong base promoted the intramolecular attack of the α-hydroxy group in compound **59** on the keto group at C7, forming synthon **60**. Destruction of this tetrahedral intermediate with migration of the aryl group promoted the formation of intermediate **61**, decarboxylation of which led to the *cis*-substituted product **62**. The resulting ketone **62** was the key synthon in the synthesis of (–)-taiwaniaquinone H (**11**) (Scheme 11).

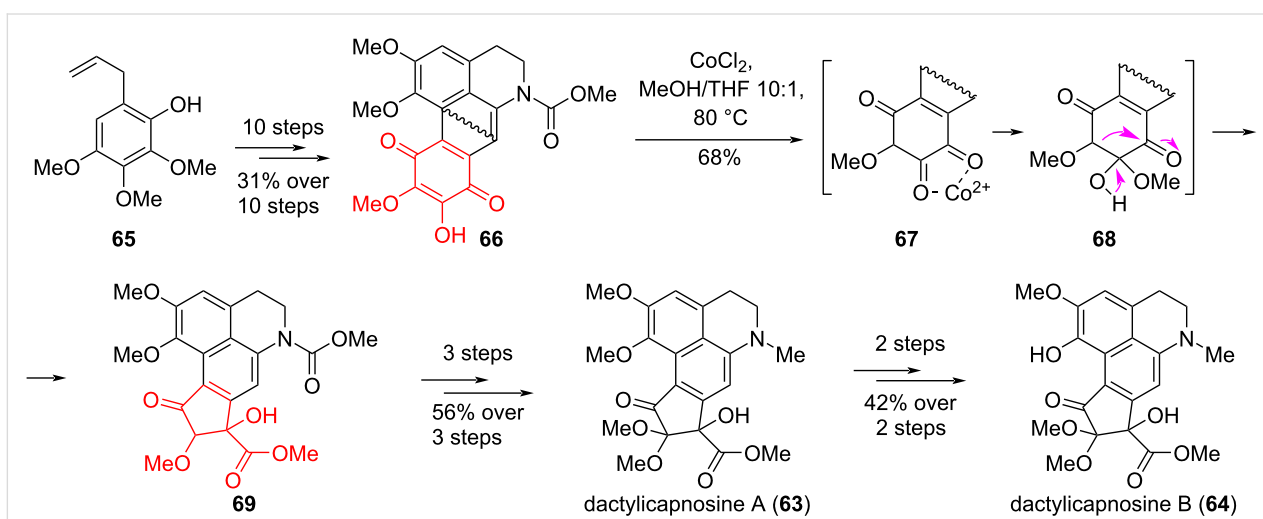
The technique of regioselective oxidative contraction of a six-membered ring was described in the work by Luo, Zhang and co-workers [43] during their synthesis of the natural alkaloids, dactylicapnosines A (**63**) and B (**64**), which contain a 9,10-seco-7-dehydroaporphinoid skeleton with an unprecedentedly highly oxygenated five-membered D-ring. These compounds exhibit potent anti-inflammatory and analgesic activity both in vitro and in vivo [44]. The synthetic method for the preparation of dactylicapnosines A (**63**) and B (**64**) was based on the known phenol **65** and involved the ring contraction of *p*-quinone **66** through a Co-mediated benzylic acid rearrangement to give cyclopentanone **69** with a 68% yield. The target products **63** and **64** were obtained in 14 and 16 steps with a total yield of 12% and 5%, respectively, from compound **65** (Scheme 12). The presented improved method opens up the possibility of obtaining medically interesting dactylicapnosine-like analogues for detailed study of their biological activity.



Scheme 10: Benzylic acid-type rearrangement in the synthesis of 4β-acetoxypobotryane-9β,15α-diol (**52**).



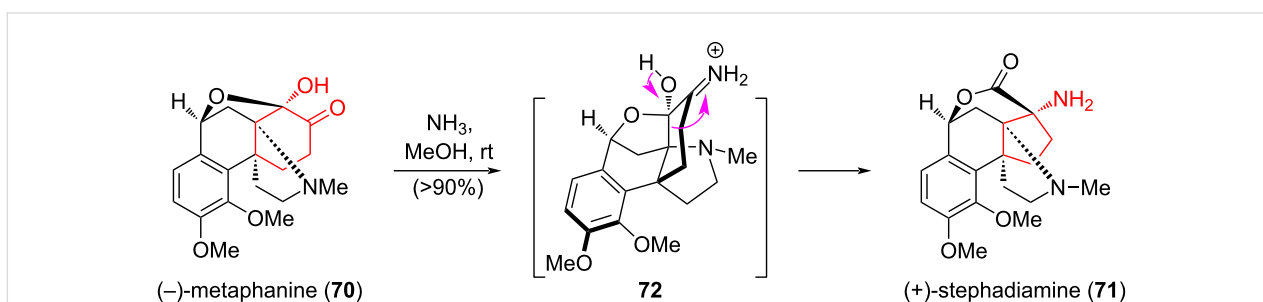
Scheme 11: Benzilic acid-type rearrangement in the synthesis of (-)-taiwaniaquinone H (11).



Scheme 12: Benzilic acid-type rearrangement in the synthesis of dactylicapnosines A (63) and B (64).

Matoba et al. [45] reported the first enantioselective synthesis of the hasubanane alkaloid (-)-metaphanine (70) and the norhasubanane alkaloid (+)-stephadiamine (71) using a cyclohexane ring contraction through aza-benzilic acid-type rearrangement (Scheme 13). Hasubanan-type alkaloids exhibit a wide range of biological activities, including antiviral, antimicrobial, and cytotoxic activities [46], while the biological activities and synthesis of structurally attractive norhasubanan-type alkaloids

have been little studied, and are of interest to synthetic chemists. Thus, the interaction of (-)-metaphanine (70) with ammonia in CH₃OH at room temperature led to the in situ formation of imine 72, which, as a result of an intramolecular aza-benzilic acid-type rearrangement, was converted in more than 90% yield to (+)-stephadiamine (71). The authors noted that the identification of the structure of (+)-stephadiamine (71) was carried out without purification of the reaction mixture due to



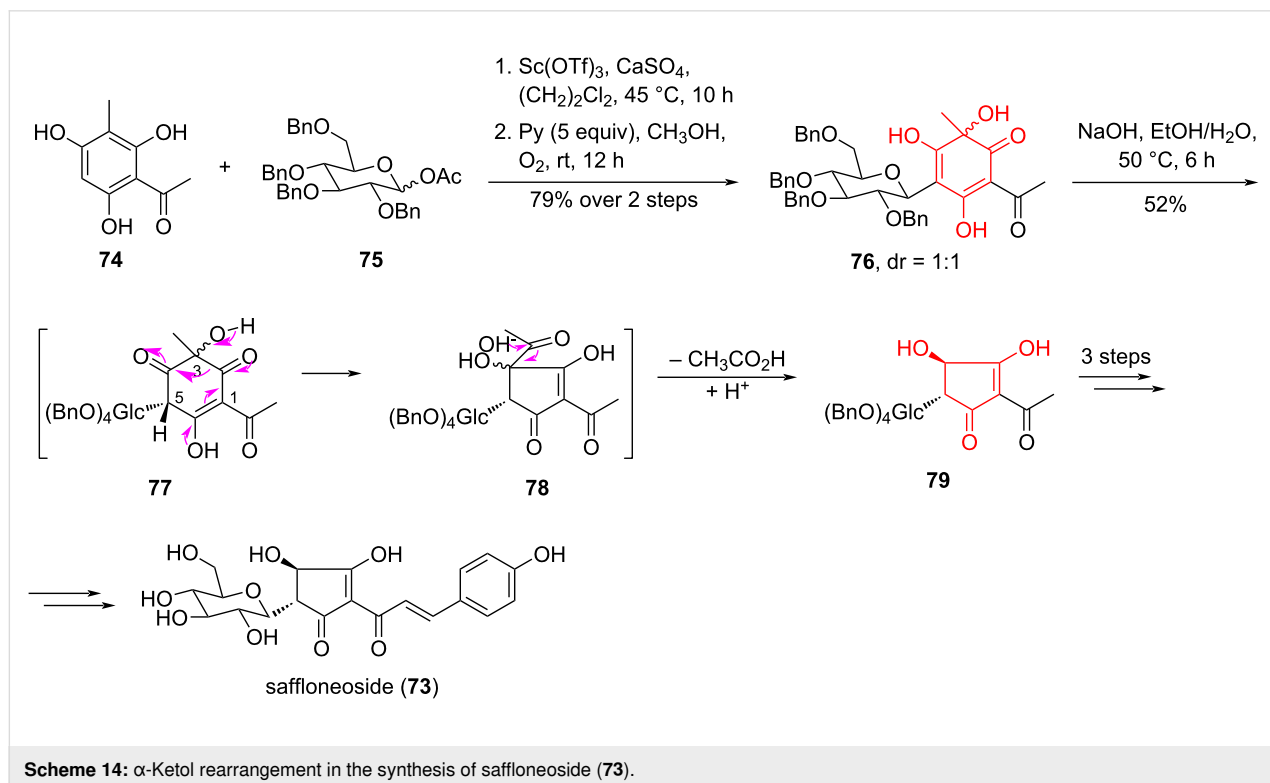
Scheme 13: Aza-benzilic acid-type rearrangement in the synthesis of (+)-stephadiamine (71).

its instability on a silica gel column under acidic or basic conditions.

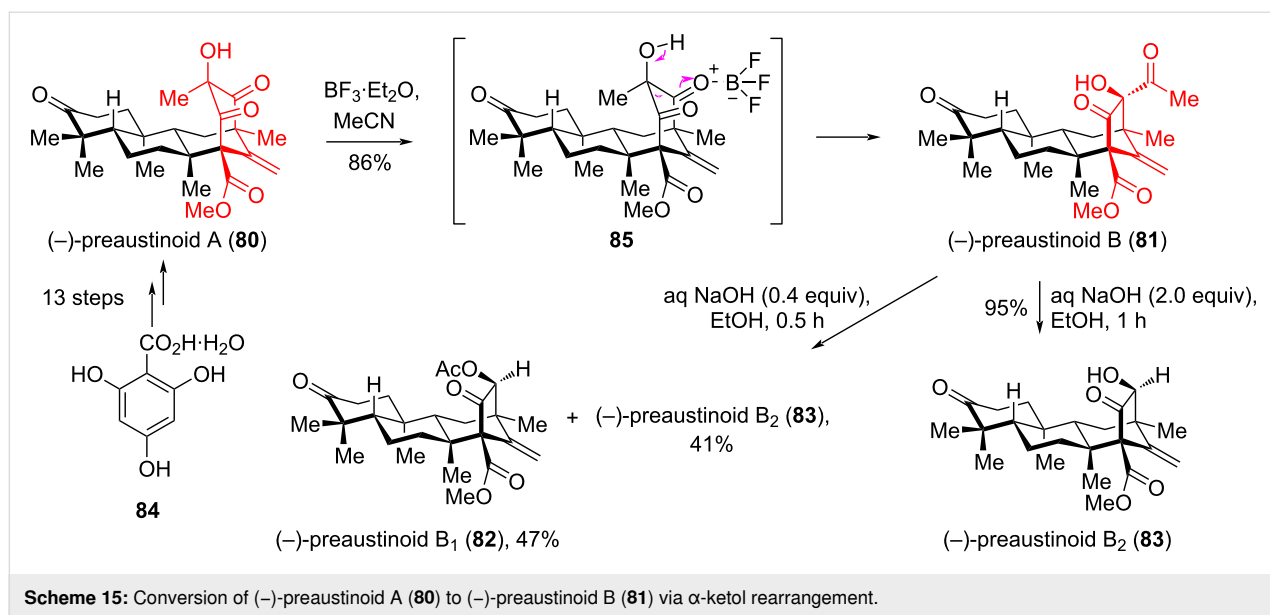
Zhang et al. [47] proposed an efficient seven-step synthesis of the new, structurally interesting *p*-hydroxycinnamylcyclopentenone *C*-glucoside saffloneoside (**73**), isolated from the inflorescences of *carthamus tinctorius*. *Carthamus tinctorius* L. is a medicinal plant that can be used to treat ischemic stroke [48]. In their study [47] they reported a stereospecific reduction of an acyloin ring controlled by the chirality of a glucose fragment at position C5, and suggested a possible mechanism for hydrogen migration during the conversion of C5 (sp²) to C5 (sp³). The key 3,4,6-trihydroxycyclohexadienone **76** was obtained by a sequence of Friedel–Crafts-type glucosylation reactions of 2,4,6-trihydroxy-3-methylacetophenone (**74**) with glucosyl acetate **75** in the presence of Sc(OTf)₃ (0.4 equiv) as a Lewis acid promoter and oxidative dearomatization of the *C*-glucoside precursor by O₂ in the presence of pyridine in CH₃OH. The key step in the acyloin ring contraction of compound **76** was the stereospecific formation of the cyclopentenone-containing intermediate **78**, by α -ketol rearrangement under the action of NaOH solution. After cleavage of the acyl moiety in synthon **78** under basic conditions, diastereomerically pure cyclopentenone **79** was obtained with 52% yield (Scheme 14). The authors [47] provided convincing evidence that the acyloin ring contraction is stereospecific, by replacing the benzil *D*-glucose in compound **76** at the C5 atom with an achiral prenyl group.

In [49], Zhang and co-workers carried out the first enantioselective synthesis of fungal meroterpenoids ((–)-preaustinoids A (**80**), B (**81**), B₁ (**82**), B₂ (**83**)) from commercially available 2,4,6-trihydroxybenzoic acid hydrate (**84**) using a cyclohexanone ring contraction strategy – α -ketol rearrangement. Fungal meroterpenoids derived from a simple aromatic polyketide, 3,5-dimethylorcellinic acid (DMOA), are a large series of hybrid natural products with a huge structural diversity and impressive biological activity [50]. Attempts by the authors to use acidic (TiCl₄, TMSI, SnCl₄), basic (NaOMe, *t*-BuONa, Et₃N, DBU, DMAP) and thermal conditions (PhMe, 110 °C) failed to give the desired rearrangement product. However, treatment of (–)-preaustinoid A (**80**) with BF₃·Et₂O in MeCN led to an α -ketol rearrangement affording (–)-preaustinoid B (**81**) as the sole product in 86% yield. Presumably, this reaction proceeded through intermediate **85**. Subsequently, the deacylation of the product **81** with 2.0 equivalents of aqueous NaOH in EtOH, afforded (–)-preaustinoid B₂ (**83**) with 95% yield. Using 0.4 equivalents of aqueous NaOH, a mixture of (–)-preaustinoid B₁ (**82**) and (–)-preaustinoid (**83**) was obtained (Scheme 15).

Nagaraju and Prasad used the α -ketol rearrangement mediated by Lewis acids to carry out the ring contraction reaction on simple cyclohexanones to synthesize functionalized di- and triquinanes [51]. Treatment of 6-oxabicyclo[3.2.1]octan-8-one **86** with BF₃·Et₂O (2 equiv) promoted the activation of the car-



Scheme 14: α -Ketol rearrangement in the synthesis of saffloneoside (**73**).



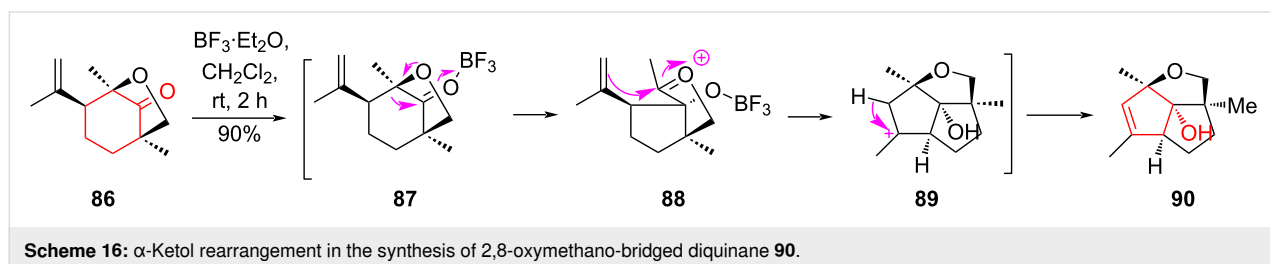
bonyl group in **87** by the Lewis acid, leading to the formation of an oxocarbenium ion **88**. Next, ion **88** reacted with the isopropenyl group, forming intermediate **89**, which was then converted into the desired 2,8-oxymethano-bridged diquinane **90** in 90% yield (Scheme 16).

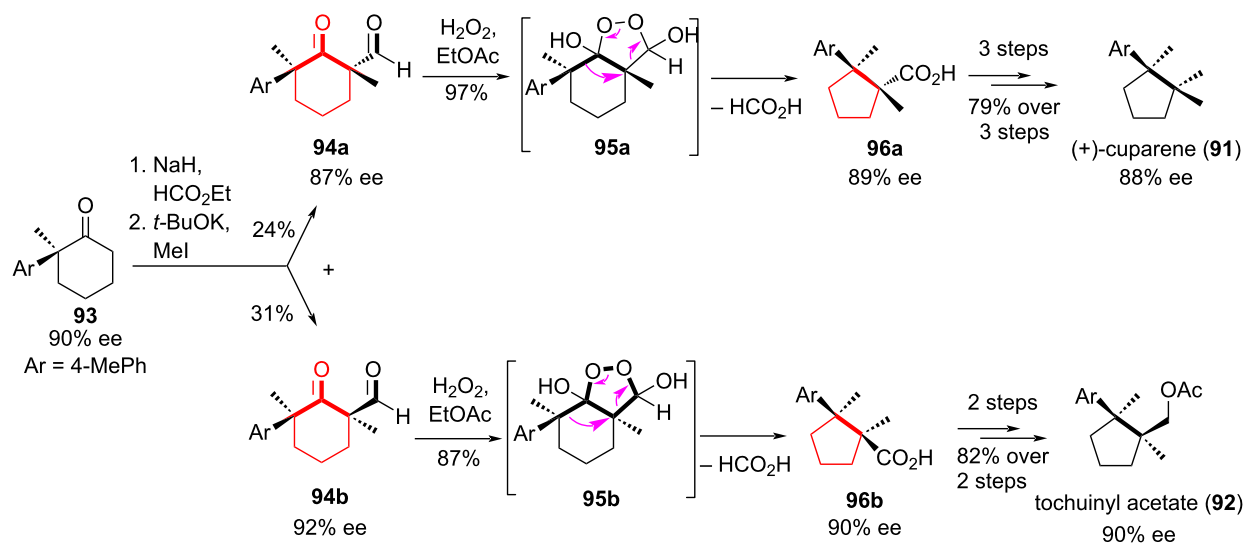
In the enantioselective synthesis of sesquiterpenoids (+)-cuparene (**91**) and (+)-tochuinyl acetate (**92**), Xie and co-workers [52] achieved high regioselectivity and stereospecific construction of contiguous all-carbon quaternary centers through an oxidative ring contraction of cyclic α -formyl ketones by the action of H_2O_2 . This reaction is easy to perform, environmentally friendly, and does not require expensive catalysts.

Formylation of enantioenriched hexanone **93** (90% ee) with $\text{NaH}/\text{HCO}_2\text{Et}$, followed by methylation produced two separable isomers, **94a** and **94b**, in yields of 24% and 31%, respectively. Treatment of these isomers with H_2O_2 smoothly delivered the carboxylic acids **96a** and **96b** stereospecifically with excellent regioselectivity (>99:1) and complete preservation of the stereochemistry at both quaternary carbon centers (Scheme 17). Through a series of synthetic transformations, the

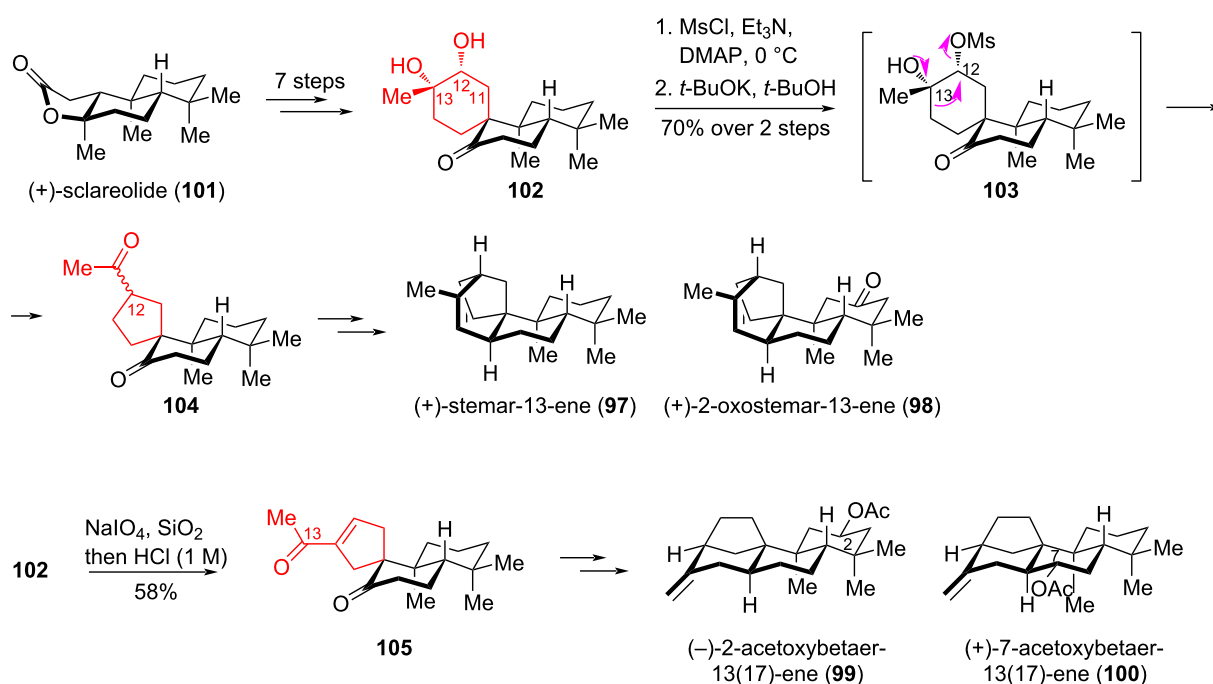
target products (+)-cuparene (**91**) and (+)-tochuinylacetate (**92**) were synthesized from acids **96a** and **96b** with high regioselectivity. In [53], the mechanism and factors controlling the selectivity of the oxidative ring contraction in cyclic α -formyl ketones under the action of H_2O_2 were studied in detail.

Due to their complex architecture and diverse biological activities, including antiviral, antitumor, antimalarial, and antifungal activity [54], stemarenes **97**, **98** and betaerene **99**, **100** diterpenoids attracted the attention of synthetic chemists. In [55], Chen et al. described the elegant synthesis of these tetracyclic diterpenoids using a semipinacol rearrangement as a ring contraction technique (Scheme 18). The authors used commercially available (+)-sclareolide (**101**) as starting compound, from which key alcohol **102** was synthesized in 7 steps. Selective mesylation of the secondary alcohol in **102** followed by semipinacol rearrangement catalyzed by *t*-BuOK afforded diketone **104** as a pair of epimers at C12 in a 1:1 ratio, with 70% overall yield. Cleavage of diol **102** with silica-supported NaIO_4 generated an unstable tricarbonyl intermediate, which without purification was rapidly treated with 1 M HCl to produce enone **105** with a yield of 58%. Through a series of synthetic transformations from the obtained diketones **104** and **105**, the target





Scheme 17: Oxidative ring contraction during the synthesis of (+)-cuparene (**91**) and (+)-tochuinylacetate (**92**).



Scheme 18: Semipinacol rearrangement in the synthesis of diterpenoids **97**–**100**.

diterpenoids **97**–**100** containing pseudoenantiomeric bridged C/D rings were obtained.

Andrastins and terretonins represent a large subset of 3,5-dimethylorcellinic acid (DMOA)-derived members with interesting biological profiles and unique carbocyclic frameworks. In 2017, Newhouse, Maimone and colleagues [56] described the

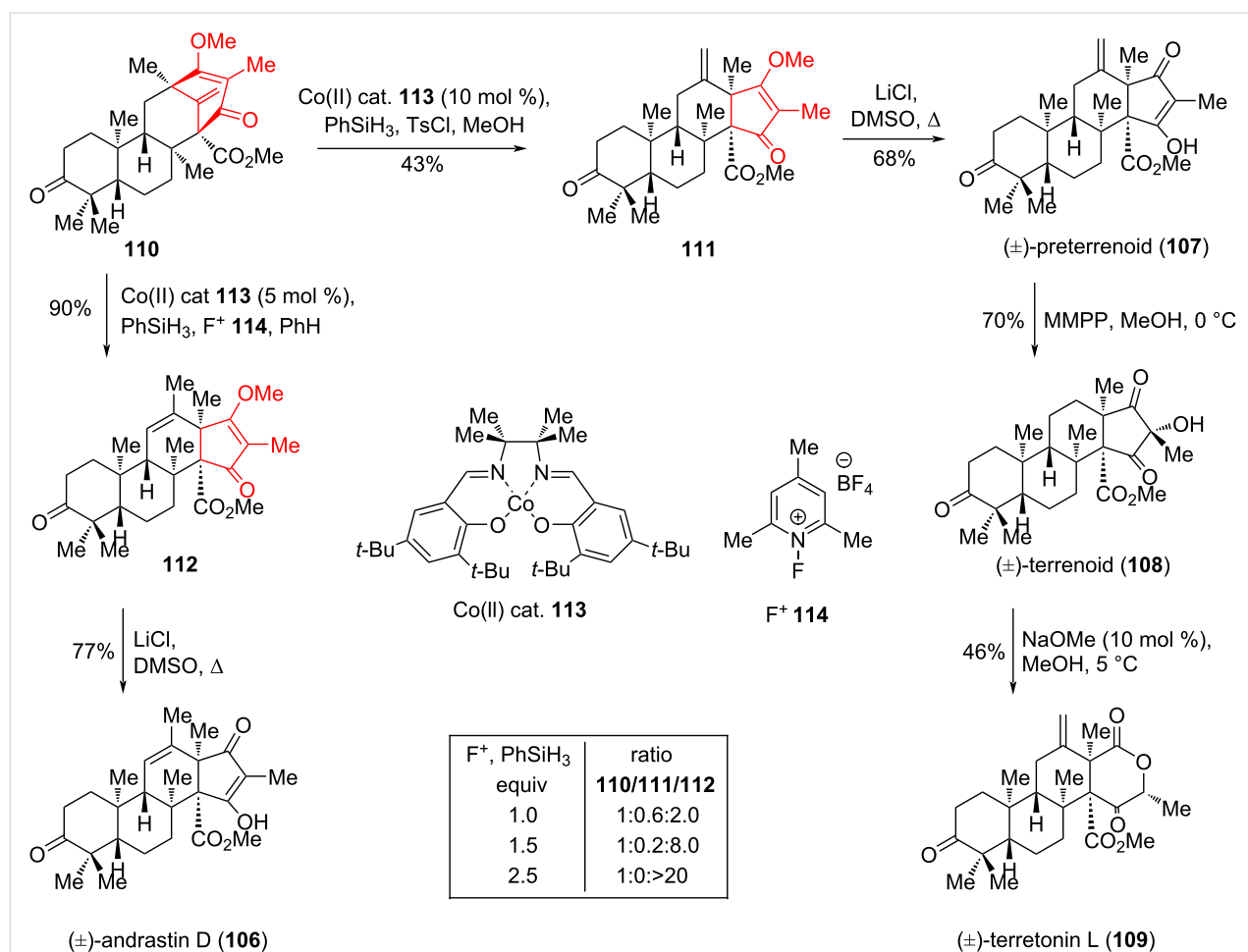
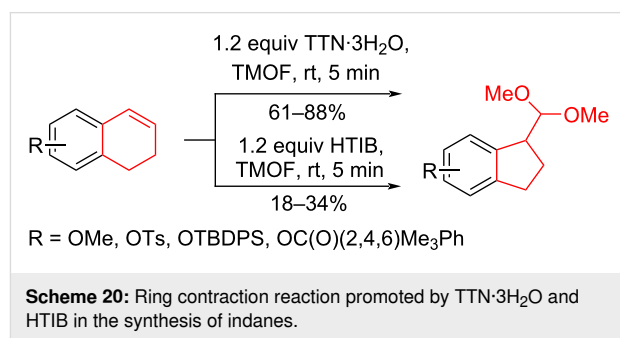
first total syntheses of racemic meroterpenes (±)-andrastin D (**106**), (±)-preterrenoid (**107**), (±)-terrenoid (**108**), and (±)-terretonin L (**109**) using a Co-catalyzed homoallyl-type rearrangement/hydrogen atom transfer (HAT) as ring contraction strategy. Radical hydrochlorination of olefin **110** [57] using Co(II) catalyst **113**, in the presence of PhSiH₃, TsCl in MeOH resulted in the formation of a cyclopentenone derivative **111**

with a yield of 43% and a small amount (<5%) of cyclopentenone derivative **112**. The remaining mass balance consisted mainly of the recovered starting material. Demethylation of **111** led to (±)-preterrenoid (**107**) with 68% yield, which was subjected to stereoselective oxidation with magnesium monoperoxyphthalate (MMPP) to give (±)-terrenoid (**108**) as a single diastereomer. Subsequent treatment of the key precursor **108** with catalytic amounts of NaOMe in MeOH resulted in (±)-terretonin L (**109**) as the main product with a yield of 46%. Radical hydrochlorination of olefin **110** under more powerful oxidizing conditions (Co(II) catalyst **113**, *N*-fluoropyridinium salt (F⁺) **114**), resulted in the formation of cyclopentenone derivative **112** with a yield of 90%. Demethylation of the latter led to (±)-andrastin D (**106**) at a yield of 77% (Scheme 19).

2.2 Ring-contraction reactions promoted by thallium(III) and iodine(III)

Silva et al. [58,59] described a universal and efficient strategy for the synthesis of indanes using a ring contraction reaction mediated by iodine(III) and thallium(III) (Scheme 20). Indanes

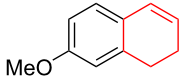
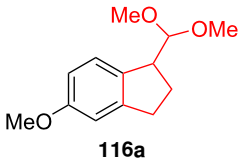
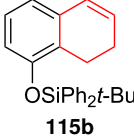
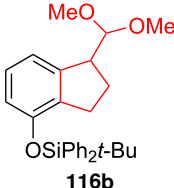
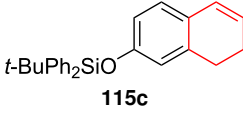
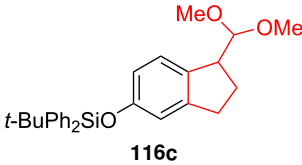
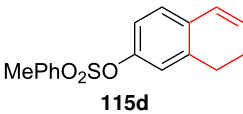
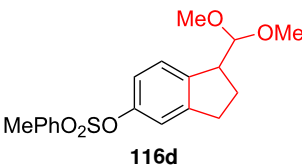
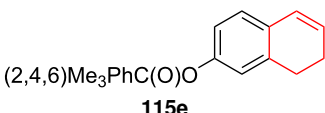
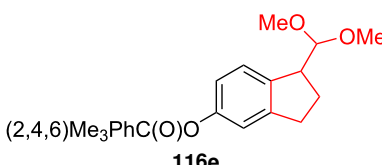
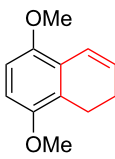
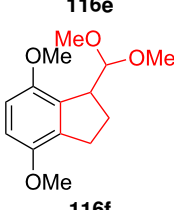
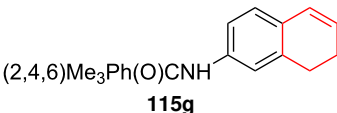
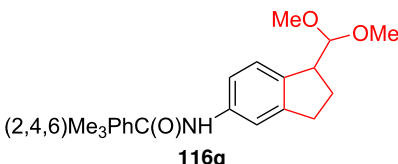
(benzocyclopentanes) and their derivatives are widely used in materials science and nanotechnology and have important biological properties such as anti-allergic, antitumor, anticonvulsive, antihypercholesteremic, herbicidal, fungicidal and antimicrobial activity [60]. In [58], a comparative study of two effective and practical oxidizing agents, hydroxy(tosyloxy)iodobenzene (HTIB) and thallium nitrate trihydrate (TTN·3H₂O), was presented in the ring contraction reaction of protected 1,2-dihydronaphthalenes using trimethylorthoformate (TMOF). In



addition, various protective groups were introduced to substrates, aiming to study their tolerance for both oxidizing agents (Table 1). The yields of the ring contraction products (indanes) **116a–g** ranged from 61 to 88% when the reactions were carried out with TTN·3H₂O in TMOF. When HTIB was used in TMOF, the yields were significantly lower, ranging from 18%

to 34% (Scheme 20). Analysis of literature data [58] indicated that the thallium(III) salt is a much better oxidizing agent than iodine(III) for indane syntheses. Similarly, the TTN-mediated reactions are less prone to form addition by-products compared to the HTIB-mediated reactions of 1,2-dihydronaphthalenes.

Table 1: Oxidative contraction of 1,2-dihydronaphthalene derivatives **115a–g** under the action of TTN·3H₂O and HTIB during the synthesis of indanes **116a–g**.

Entry	Substrate	Acetal (indane)	Indane yield under the action of	
			HTIB in TMOF	TTN·3H ₂ O in TMOF
1	 115a	 116a	29%	88%
2	 115b	 116b	28%	74%
3	 115c	 116c	34%	79%
4	 115d	 116d	29%	69%
5	 115e	 116e	31%	61%
6	 115f	 116f	18%	72%
7	 115g	 116g	32%	68%

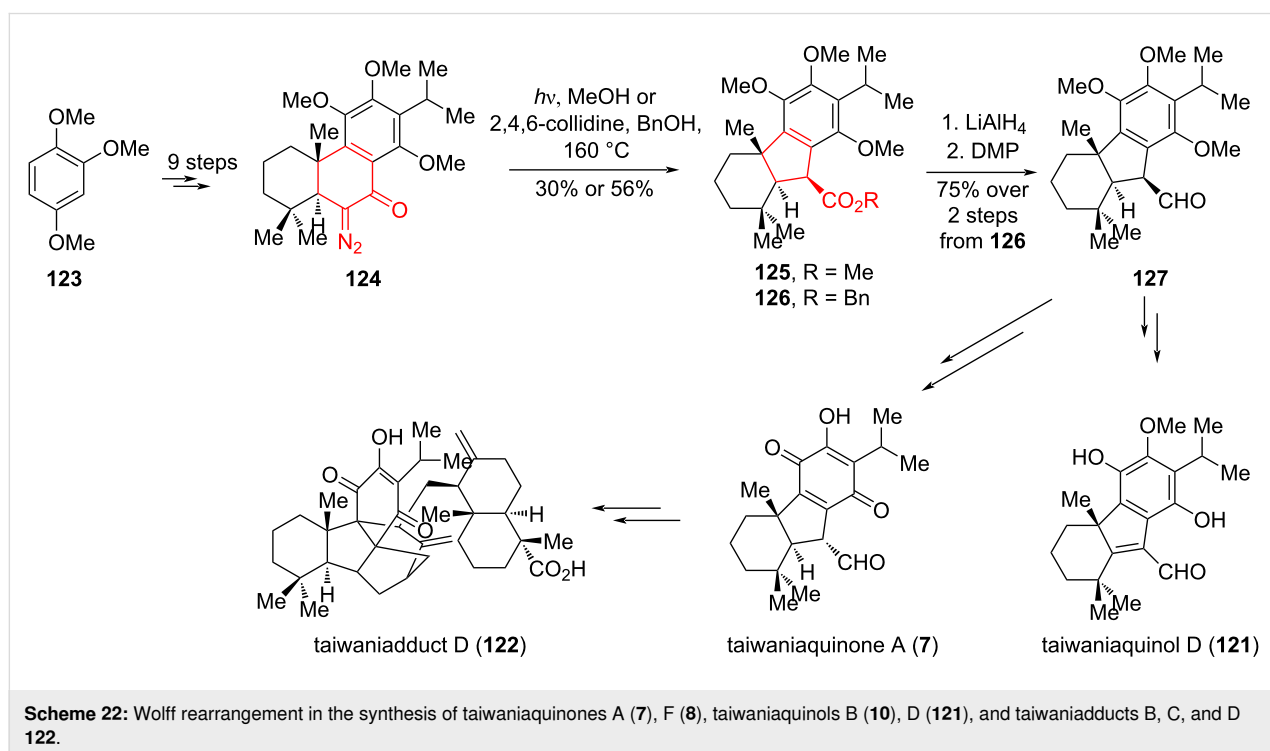
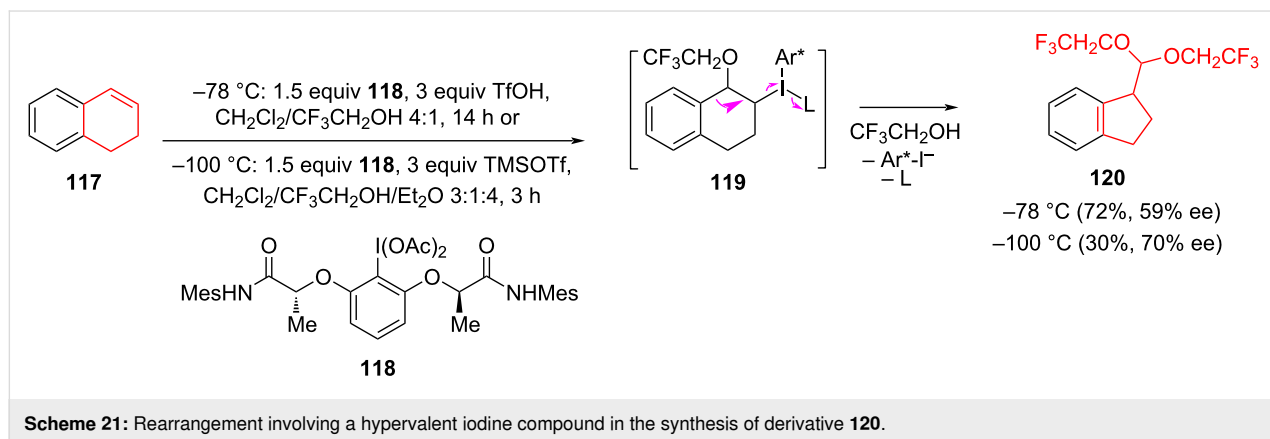
Silva et al. have also developed an effective method for the stereoselective oxidation of tetralone derivatives using chiral hypervalent iodine reagents [61,62]. Hypervalent iodine compounds are widely used in organic synthesis as selective oxidants and enantiomerically pure reagents. In terms of reactivity, they are environmentally sustainable alternatives to heavy metals. The advantage of hypervalent catalytic systems based on iodine lies in their reusability [63]. The oxidative contraction of the tetralone cycle **117** depended on the stoichiometric amounts of iodine(III)-based chiral reagent **118** using trifluoromethanesulfonic acid (TfOH) or TMSOTf as Lewis acids. The authors suggested that the reaction proceeded via the formation of an intermediate **119** followed by reaction with an alcohol nucleophile and a 1,2-carbon migration to produce a

cyclopentane derivative **120** with a yield of 30% and a selectivity of 70% at $-100\text{ }^{\circ}\text{C}$ (Scheme 21).

2.3 Wolff rearrangement

The Wolff rearrangement is the transformation of α -diazoketones into acids or their derivatives through heating, catalysis, or UV irradiation in the presence of water, alcohols, ammonia, amines, etc.

The Wolff rearrangement is a convenient method for forming a cyclopentane ring in the synthesis of racemic forms of taiwaniaquinones A (**7**), F (**8**), and taiwaniaquinols B (**10**), D (**121**) [64] and taiwaniaadducts B, C and D **122** [65], proposed by Deng et al. (Scheme 22). Taiwaniaquinoids are a class of



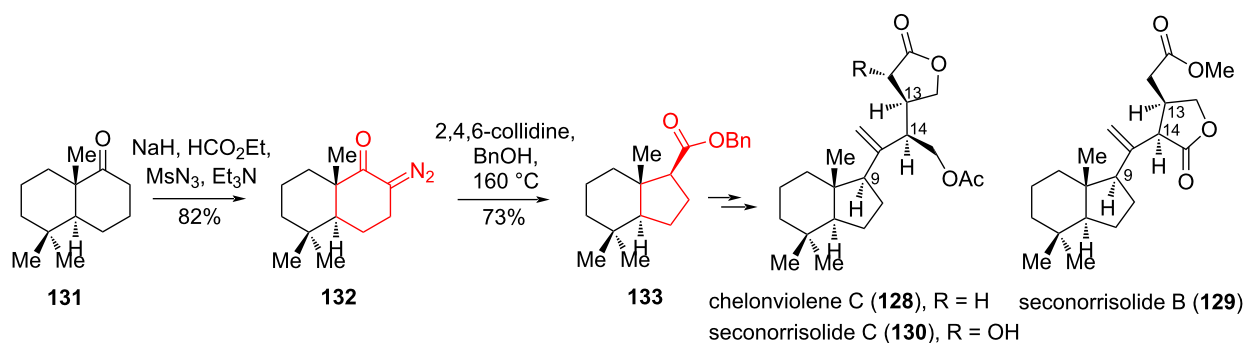
terpenoids with an unusual 6,5,6-abeoabietane scaffold and impressive antitumor activity. α -Diazoketone **124** was synthesized from commercially available 1,2,4-trimethoxybenzene (**123**) [64] by a series of synthetic transformations, and converted by Wolff rearrangement to esters **125** and **126** with a *trans*-fused 6,5,6-ring system. Irradiation of a solution of **124** in CH₃OH using a medium-pressure Hg lamp resulted in the formation of methyl ester **125** as a single diastereomer with a yield of 30%. Due to the decrease in the efficiency of this reaction at higher loads (>50 mg), the thermal conditions for the Wolff rearrangement were tested (BnOH, 2,4,6-collidine, 160 °C). The desired benzyl ester **126** was isolated as single diastereomer with a yield of 56% yield. Reduction of **126** with LiAlH₄ followed by oxidation with Dess–Martin periodinane (DMP) led to the synthesis of a key aldehyde **127**, with a yield of 75% in two steps. The target products were synthesized through a series of synthetic transformations starting with **127**.

In [66], it was shown that the Wolff rearrangement is an effective method for obtaining a transhydrindane system in norrisolide-type rearranged spongian diterpenes chelviolene C (**128**), seconorrisolide B (**129**), and seconorrisolide C (**130**) from Wieland–Miescher ketone derivative **131**. The spongian diterpenes are a large family of marine natural products that have pronounced effects on the Golgi apparatus [67]. The introduction of the diazo group into ketone **131** was achieved under standard conditions through an α -formyl intermediate. The Wolff rearrangement was carried out by heating **132** in collidine and BnOH to obtain *trans*-hydrindane **133** as a single diastereomer with a yield of 73% (Scheme 23).

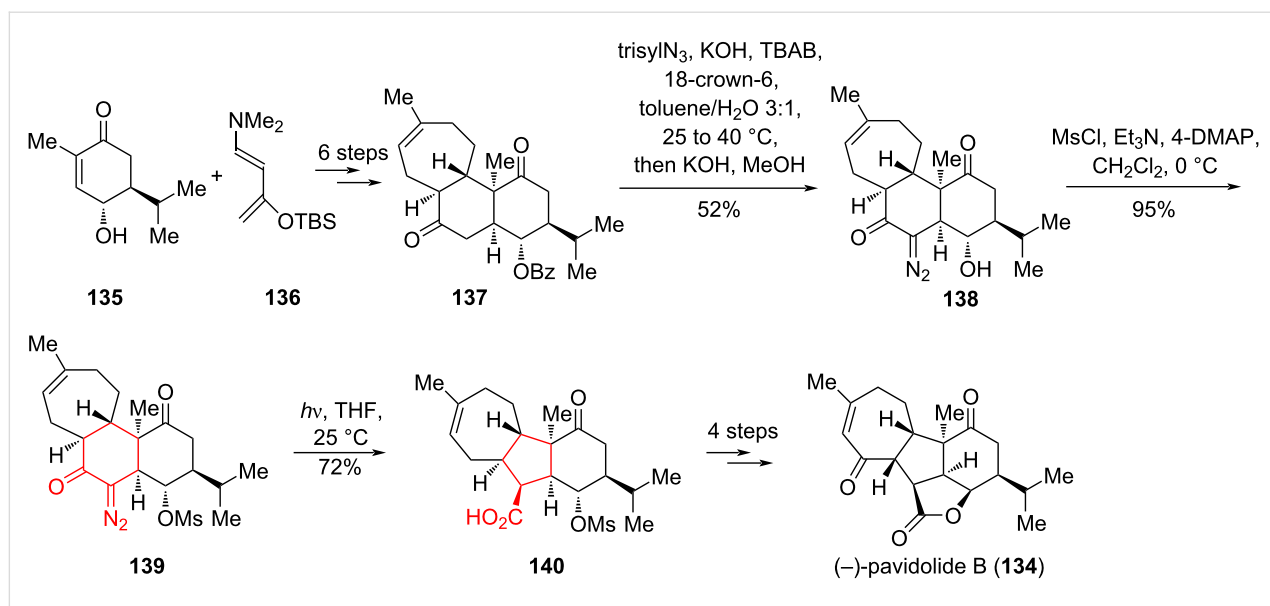
In the work of Ding et al. [68], an elegant synthesis of (–)-pavidolide B (**134**) was described using the Wolff rearrangement. In 2012, Lin and co-workers [69] isolated the membrane diterpenoid (–)-**134** from the marine soft coral *Simularia pavida*. The study showed that (–)-**134** exhibits highly selective

inhibition against the human promyelocytic leukemia cell line HL-60 with an IC₅₀ of 2.7 μ g/mL. Structurally, (–)-**134** contains a challenging [6-5-5-7]-tetracyclic ring system embedded within a fully functionalized cyclopentane, as well as seven contiguous stereogenic centers. In 2017, (–)-pavidolide B (**134**) was prepared in 13 steps from the known chiral alcohol **135**, which in turn was synthesized from commercially available (+)-carvone in two steps [70]. The intermediate tricyclic ketone **137** was converted to the corresponding α -diazoketone through treatment with 2,4,6-triisopropylbenzenesulfonyl azide (trisylN₃), KOH, tetrabutylammonium bromide (TBAB), and 18-crown-6. Concurrent cleavage of the benzoyl group under basic conditions led to the isolation of alcohol **138** with a yield of 52%. The Wolff rearrangement was carried out upon irradiation of a solution of mesylate **139** in THF at 25 °C with a medium-pressure Hg lamp for 1 hour. The ring contraction proceeded diastereoselectively and the desired acid **140** was obtained as a single diastereomer with a yield of 72%. A subsequent series of synthetic transformations on acid **140** resulted in the target (–)-pavidolide B (**134**) (Scheme 24).

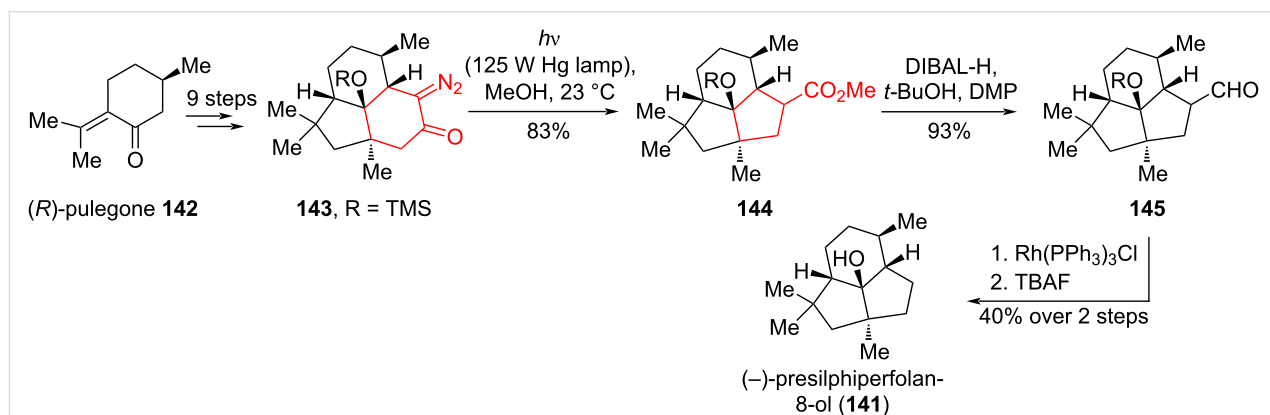
Hu and Snyder [71] carried out the first enantioselective synthesis of the highly strained natural product (–)-presilphiperfolan-8-ol (**141**) in 13 steps from commercially available (*R*)-pulegone (**142**) using a ring contraction strategy based on the Wolff rearrangement. By irradiating diazo ketone **143** for 20 minutes at 23 °C in MeOH, the desired product **144** was obtained with a yield of 83%. The synthesis of (–)-presilphiperfolan-8-ol (**141**) was completed by converting the exocyclic methyl ester **144** to an aldehyde **145** using a one-pot DIBAL-H/Dess–Martin procedure followed by the removal of TMS protecting group using a stoichiometric Wilkinson's catalyst (Scheme 25). The authors noted that, in the final stages of reduction and oxidation, yields were lower when **145** was obtained (70% vs 93%). At the same time, the reaction with (PPh₃)₃Cl in small amounts (\approx 15 mg) proceeded smoothly.



Scheme 23: Wolff rearrangement in the synthesis of chelviolene C (**128**), seconorrisolide B (**129**), and seconorrisolide C (**130**).



Scheme 24: Wolff rearrangement in the synthesis of (-)-pavidolide B (134).

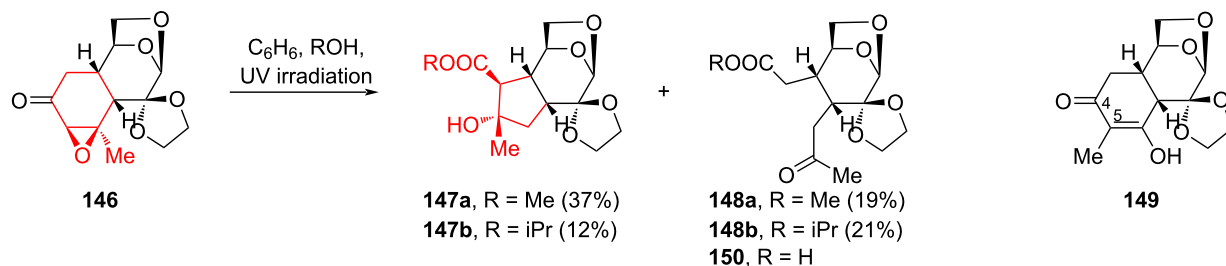


Scheme 25: Wolff rearrangement in the synthesis of presilphiperfolan-8-ol (141).

3 Photochemical rearrangements

Photochemical reactions are a powerful tool for quickly accessing a variety of structural and stereochemical compounds from relatively simple building blocks [72]. The main feature of light-induced transformations is the involvement of electronically excited states that occur during photon absorption, leading to the formation of reactive intermediates causing significant changes in chemical compound reactivity. As a result, light energy can be used to synthesize complex and unique compounds that cannot be produced by thermal reactions, making photochemical transformations preferable to classical ones, especially when considering the concept of «green chemistry» [73]. Therefore, photochemical rearrangements have great theoretical and practical importance when considering various transformations including the transformation of hexanes into cyclopentanes.

An interesting photochemical rearrangement of 5,6-epoxy derivatives of the Diels–Alder adduct of LG and piperylene was described in reference [74]. This reaction leads to the formation of cyclopentane derivatives, promising for the synthesis of iridoids and compounds, that stimulate the formation of receptors that prevent platelet aggregation [75]. The derivatives of the Diels–Alder adduct of LG and piperylene, keto epoxide **146** and its analogue **152** lacking the spiro-fused ethylenedioxy fragment (see below) are attractive substrates for studying the possibility of a photochemically induced contraction of the cyclohexane ring. Photoirradiation of keto oxirane **146** [76] in benzene containing 1% methanol afforded cyclopentane annulated derivative **147a** along with keto ester **148a**, a product of cyclohexane ring opening (Scheme 26). A decrease in the methanol content of the mixture to 0.5% was accompanied by a slowdown in the process and the appearance of keto enol **149** in

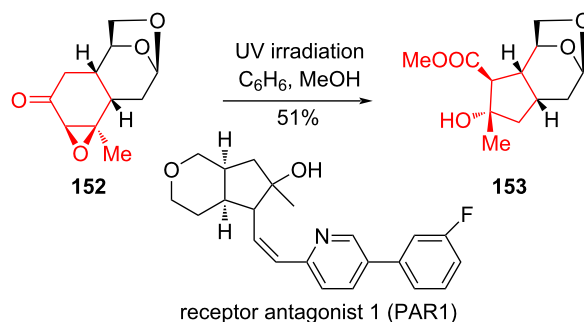


Scheme 26: Photochemical rearrangement in the synthesis of cyclopentane derivatives **147a,b**.

solution. Replacing methanol with isopropanol under these conditions resulted in the formation of isopropyl esters **147b** and **148b**. The authors of [74] suggested that keto enol **149** was the intermediate from which the cyclopentane derivative **147** and fragmentation products **148** and **150** were formed. Irradiation of a solution of keto enol **149** in benzene containing 1% methanol resulted in the formation of the proposed products **147a** and **148a**. The authors of [74] believe that methyl ketones **148a,b** were obtained as a result of breaking the C4–C5 bond, and cyclopentanes **147a,b** were formed from methyl ketones through intramolecular aldol condensation.

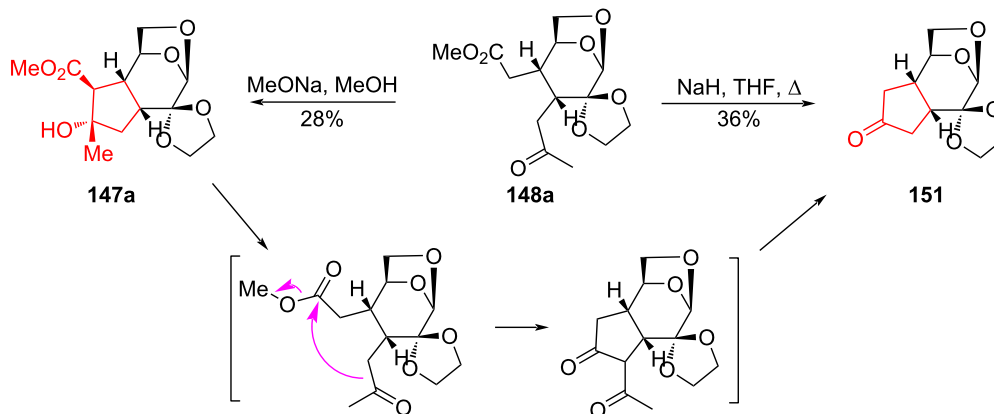
Treatment of keto ester **148a** with NaH led to a complex mixture, from which cyclopentanone annulated derivative **151** was isolated with a yield of 36%. Performing the reaction under milder conditions (MeONa/MeOH) yielded cyclopentane derivative **147a** with a yield of 28%. The latter compound, when boiled in THF solution in the presence of NaH produced cyclopentanone derivative **151** (Scheme 27). According to the authors, this transformation proceeded through stages of retro-Claisen condensation, intramolecular aldol-type condensation, and retro-Claisen deacetylation [77].

Photolysis of compound **152** proceeded more efficiently in $\text{C}_6\text{H}_6/\text{MeOH}$ (1%) resulting in cyclopentanone derivative **153** with a yield of 51% (Scheme 28).



Scheme 28: Photochemical rearrangement in the synthesis of cyclopentane derivative **153**.

As a result of the photolysis of 5,6-epoxy ketones **146** and **152**, valuable derivatives of cyclopentane, **147a,b**, **151**, and **153** were obtained.

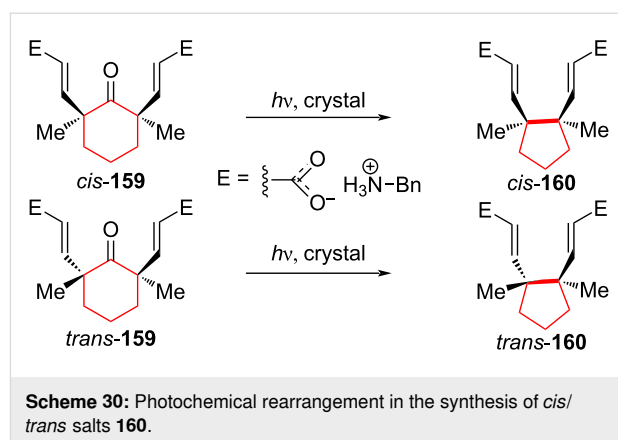


Scheme 27: Synthesis of cyclopentane derivatives **147a** and **151**.

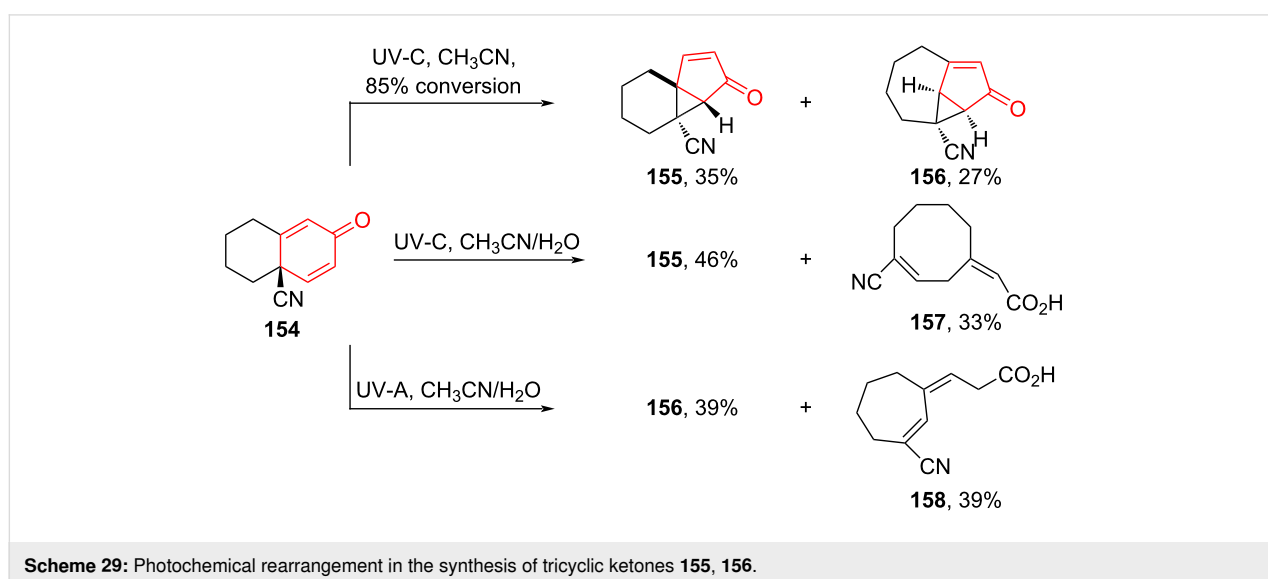
Photochemistry is an important tool in the synthesis of natural and synthetic biologically active substances based on polycyclic compounds [78–82]. The work of Davis, Derksen and co-workers [83] presents interesting results on the use of UV light to study the reactivity of bicyclic divinyl ketones and the dependence of photochemical reaction products on wavelength. Under anhydrous conditions, irradiation of divinyl ketone **154** with UV-C (200–280 nm) light yielded approximately equal amounts of products **155** and **156** (1.25:1 after 85% conversion). In the presence of water (CH₃CN/H₂O 10:1), a crystalline cyclooctene compound **157** was obtained by irradiation of **154**. However, compound **156** was not detected. When irradiating **154** with UV-A (315–400 nm) in the presence of water (CH₃CN/H₂O 10:1), cycloheptene compound **158** was obtained, but compound **155** could not be detected. These results are consistent with the selective conversion of ketone **155** to acid **158** by UV-A irradiation and the selective transformation of ketone **156** to acid **157** by UV-C, which occurred only in the presence of water (Scheme 29). In the work [83], it is shown that a vinylogous Norrish Type II cascade reaction can occur for unsaturated ketones substituted with cyclopropanes.

Interesting results of photochemical ring contraction were described by Dotson et al. [84] for *cis/trans* salts **159** (Scheme 30). The authors noted that photochemical experiments carried out in benzene solutions with benzylic esters *cis*-**159** or *trans*-**159** gave complex product mixtures. To achieve higher conversion, a photochemical reaction was carried out in the crystalline phase. Benzyl esters were cleaved, and the resulting carboxylic acids were converted into corresponding benzylammonium salts by addition of two equivalents of benzylamine. Subsequent irradiation of these salts in the form of a crystalline

suspension in hexane led to ring-contraction products with a 95% yield.



Jin et al. [85] reported an efficient photoinduced carboborative ring contraction of monounsaturated six-membered carbo- and heterocycles, allowing the regio- and stereoselective synthesis of functionalized cyclopentanes at gram scales. This method leads to the formation of compounds with multiple stereocenters, including contiguous quaternary carbons. As a result, the carboborative ring contraction reaction is conveniently used for structural modification of natural products containing a cyclohexene ring and in the synthesis of new compounds to increase their activity, metabolic stability, and target specificity [86]. Impressive results were obtained by testing the photoinduced carboborative ring contraction reaction of terpenoids and steroids. The reactions proceeded with high regio- and stereoselectivity, leading to the formation of substituted cyclopentanes **162**, **164a,b**, **166**, **168**, **170**, derivatives of cholesterol **171a,b,c,d**, diosgenin **172a,b**, pregnenolone **173–175**, dehy-



droeppiandrosterone **176**, **177a,b**, and azasteroids **178a,b**, **179**, in the form of single diastereomers (Table 2 and Figure 2) [85].

In the five-step synthesis of artalbic acid (**180**), the authors [85] used a photoinduced carboborative ring contraction of (*S,S*)-carveol-derived TBS ether **163b** with triethylborane (Scheme 31). This reaction proceeded with isomerization and formation of the intermediate carbocation **181**, which underwent 1,2-carbon migration to form the cyclopentane product **182**. A sequence of oxidations of borane **182** with trimethylamine *N*-oxide and Dess–Martin periodinane, α -selenylation, and hydrogen peroxide-induced selenoxide elimination led to unsaturated ketone **183**, from which artalbic acid (**180**) was obtained through a series of chemical transformations.

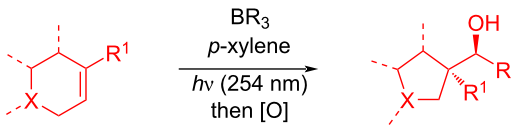
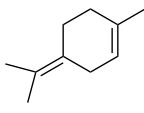
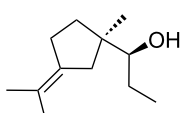
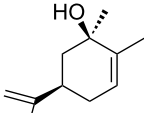
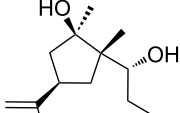
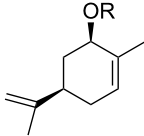
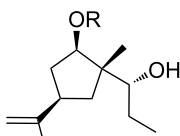
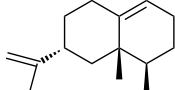
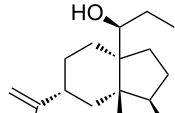
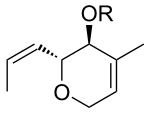
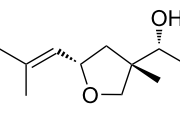
The synthetic versatility of the photoinduced carboborative ring contraction is demonstrated by the example of converting organoborane intermediates into alcohols, amines, and *E*-alkenes (Scheme 32).

4 Meinwald rearrangements of epoxides

The rearrangements of oxiranes annelated to cyclohexane occupy an important place in the synthesis of cyclopentanes due

to the stereospecificity of the reactions. Lewis acids usually initiate this transformation, and the reaction is most efficient when there are electron-accepting groups present. Sometimes, the rearrangement of epoxides into cyclopentane is accompanied by competing reactions – the formation of cyclohexanone/ols – which are products of epoxide ring opening. It should be noted that the position of the epoxy ring in the 6-membered ring has a major influence on the composition of reaction products. Catalysts only affect the yield of the reaction. A striking confirmation of this is the work of Berteina-Raboin and co-workers [87], which describes the acid-catalyzed rearrangement of epoxides **189** and **190** derived from sesquiterpene α -isocostic acid **188** under the action of various Lewis acids: $\text{BF}_3 \cdot \text{Et}_2\text{O}$, InCl_3 , TfOH , ZnBr_2 , $\text{Bi}(\text{OTf})_3$, *p*-TSA, or TFA. Under the action of the listed acids, the epoxy ring in compound **189** was opened to form three possible products: ketone **191**, olefin **192**, and cyclopentane **193** (Scheme 33). At the first stage, the epoxy ring was opened to form intermediate **A**. According to the authors [87], there are three possible ways of carbocation stabilization. The 1,2-shift of a proton at C3 to C4 led to the formation of ketone **191**. This was observed when $\text{BF}_3 \cdot \text{Et}_2\text{O}$, InCl_3 , TfOH , or ZnBr_2 were used as catalysts. Elimination of a proton from C15 led to formation of alcohol **192** with a second exocyclic double

Table 2: Scope of the photoinduced carboborative ring contraction of terpenoids.

			
Terpenoid	Product, yield, dr	Terpenoid	Product, yield, dr
 terpinolene 161	 162 , 91%, >20:1	 167	 168 , 60%
 (<i>R,R</i>)-carveol 163a , R = H 163b , R = TBS	 164a , 86%, 10:1 164b , 63%, >15:1	 valencene 169	 170 , 74%
 nerol oxide 165	 166 , 52%		

^aReaction conditions: terpenoid (1 mmol), trialkyl borane (1–1.5 mmol), EtOH (5 mL), *p*-xylene (2 mL), UV (254 nm), then H_2O_2 , NaOH, or $\text{Na}_2\text{CO}_3 \cdot 1.5\text{H}_2\text{O}$.

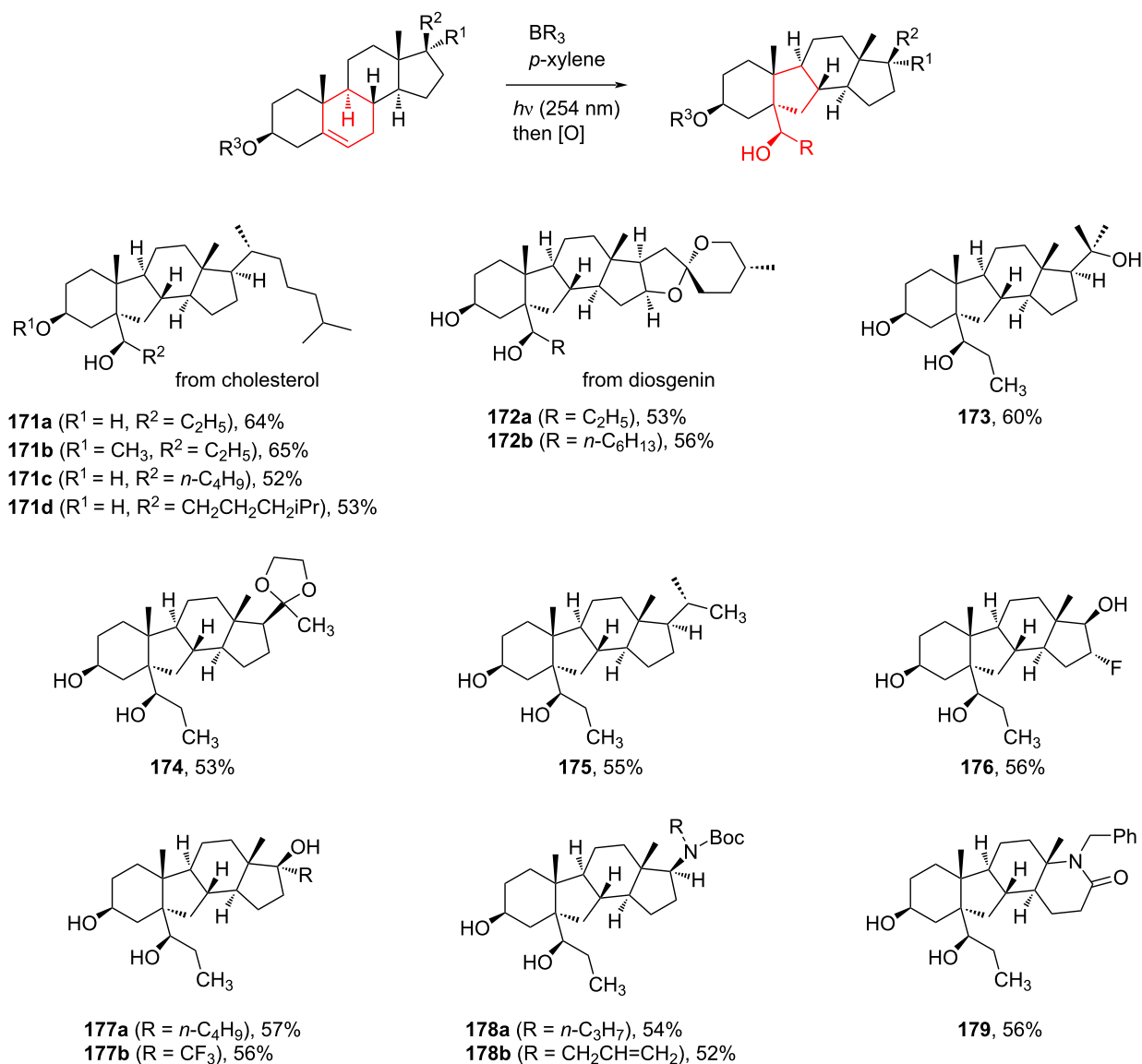
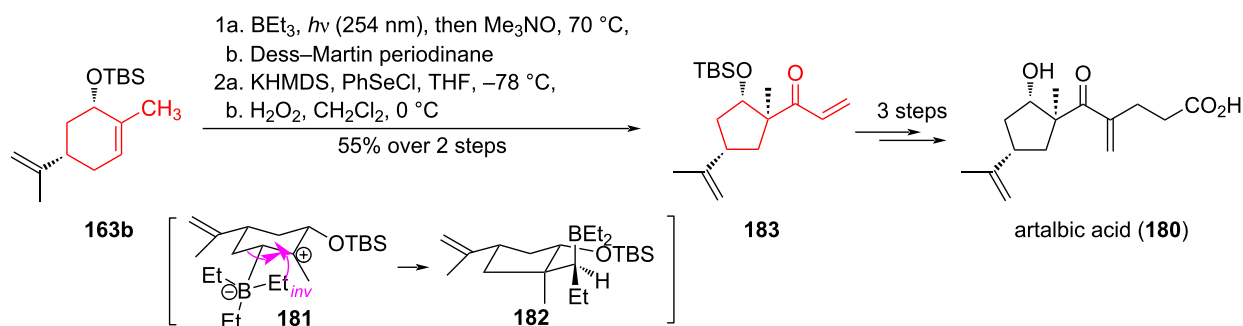
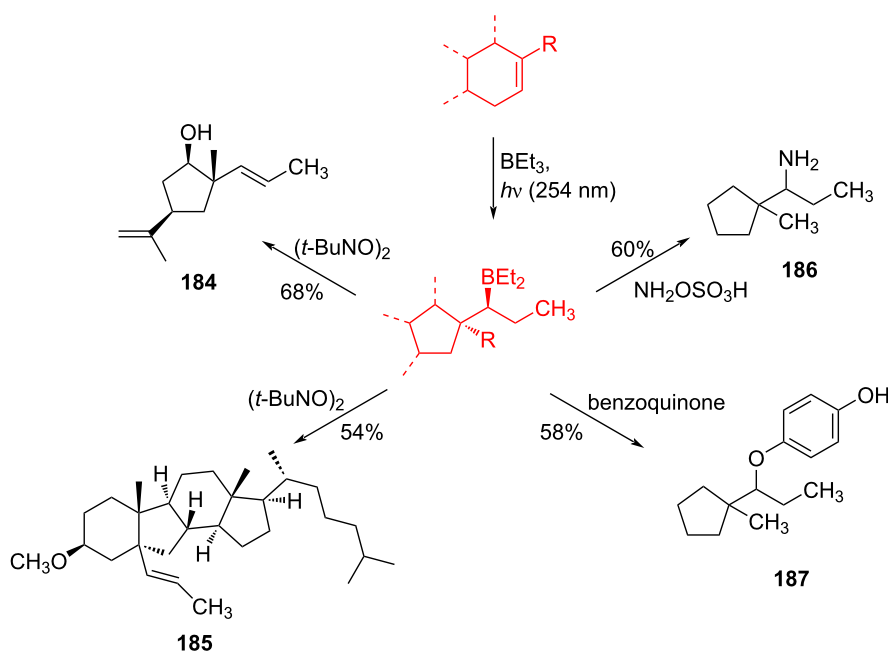


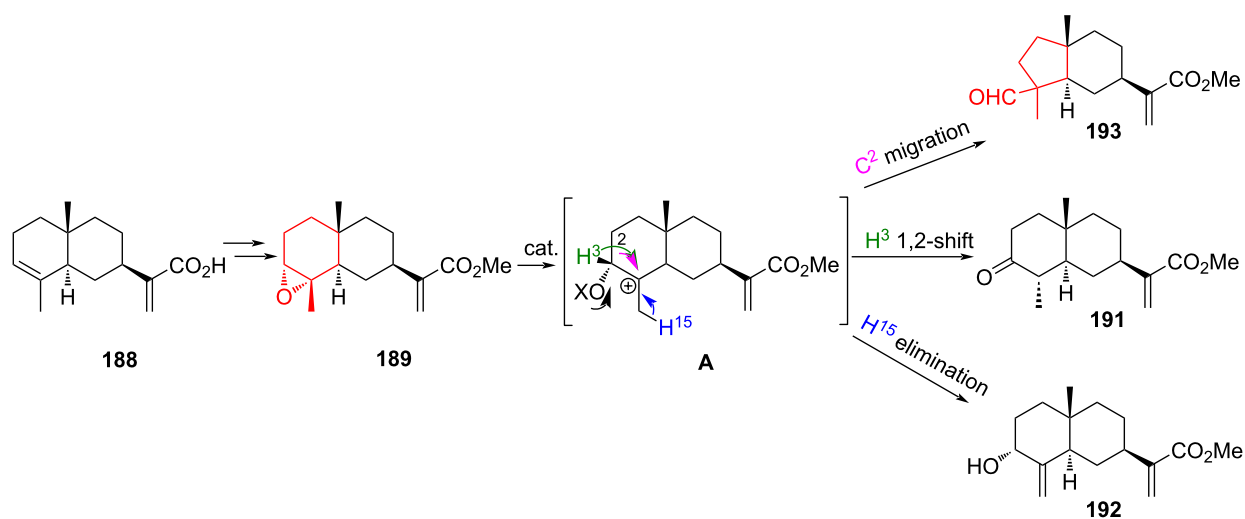
Figure 2: Scope of the photoinduced carboborative ring contraction of steroids. Reaction conditions: steroid derivative (0.23–0.5 mmol), trialkylborane (0.3–0.6 mmol), EtOH or THF (2–4 mL), *p*-xylene (0.5–1 mL), UV (254 nm), then H₂O₂, NaOH.



Scheme 31: Photoinduced carboborative ring contraction in the synthesis of artalbic acid (**180**).



Scheme 32: Synthetic versatility of the photoinduced carboborative ring contraction.

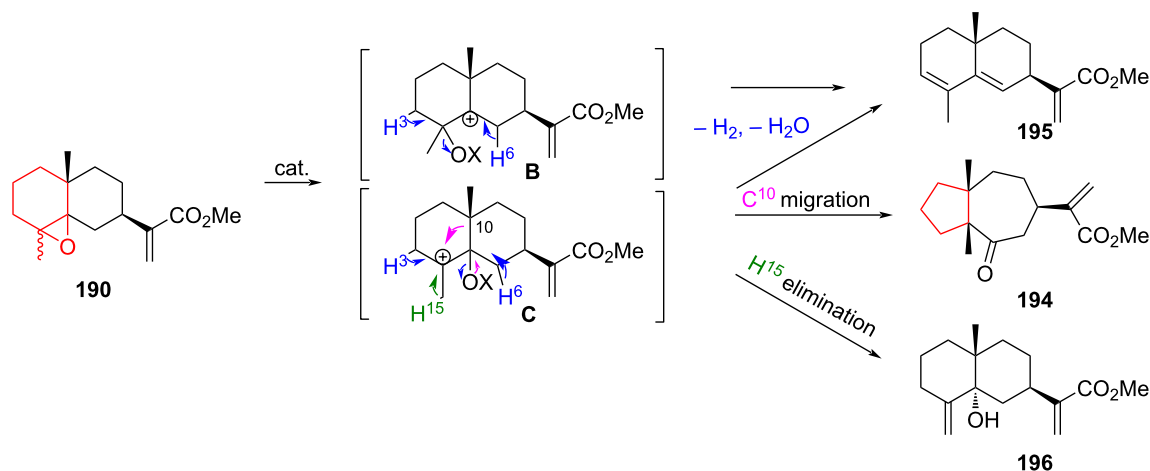


Scheme 33: Methods of disclosure of epoxide **189**.

bond in reaction mediated by ZnBr_2 , $\text{Bi}(\text{OTf})_3$, p -TSA or TFA. A 1,2-rearrangement or migration of C2 results in the formation of a cyclopentane derivative **193** in the presence of InCl_3 or $\text{Bi}(\text{OTf})_3$.

The action of the same Lewis acids on the angular epoxide **190**, a regioisomer of epoxide **189**, led to other products (Scheme 34). According to the authors [87], depending on the nature of the Lewis acid, the reaction proceeded through the

formation of intermediate carbocations **B** and **C**. The cyclopentane derivative **194**, annelated with cycloheptanone, was obtained from carbocation **C**. As a result of the reaction, the C10 atom underwent migration, which led to the expansion of the cyclopentane ring. Compound **195** was the product of the double dehydration of carbocations **B** and **C**. The regioisomer of alcohol **192**, a tertiary alcohol **196**, was formed as a result of the elimination of a proton from the C15 atom. It is noteworthy that the reaction of 1,2-oxirane **190** with InCl_3 , ZnBr_2 , TfOH ,

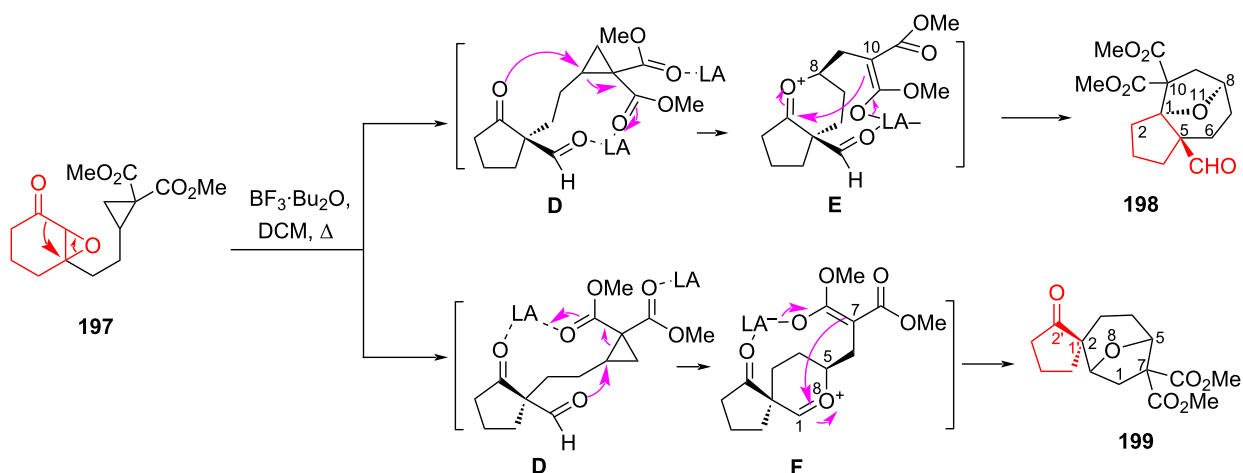
Scheme 34: Methods of disclosure of epoxide **190**.

or $\text{Bi}(\text{OTf})_3$ resulted in the formation of a mixture of ketone **194** and cyclic diene **195**. Regio- and stereoselectivity was observed in the case of the epoxide **189**. The cyclopentane derivative **194** was obtained stereoselectively and in high yield during the reaction with $\text{BF}_3 \cdot \text{Et}_2\text{O}$.

An alternative example of the tandem epoxide rearrangement of α,β -epoxy ketone **197** using Lewis acids ($\text{Sc}(\text{OTf})_3$, TMSOTf , SnCl_4 , $\text{Yb}(\text{OTf})_3$, $\text{BF}_3 \cdot \text{Et}_2\text{O}$, $\text{BF}_3 \cdot \text{CH}_3\text{CO}_2\text{H}$, $\text{BF}_3 \cdot \text{THF}$, $\text{BF}_3 \cdot \text{Bu}_2\text{O}$) as a catalyst and different solvents (DCE, DCM, THF, CHCl_3) has been described in [88]. By varying the temperature, solvent, and catalyst, optimal conditions were found for time, yield, and ratio of reaction products (Scheme 35). The authors [88] suggested that the 1,2-oxirane was opened by the

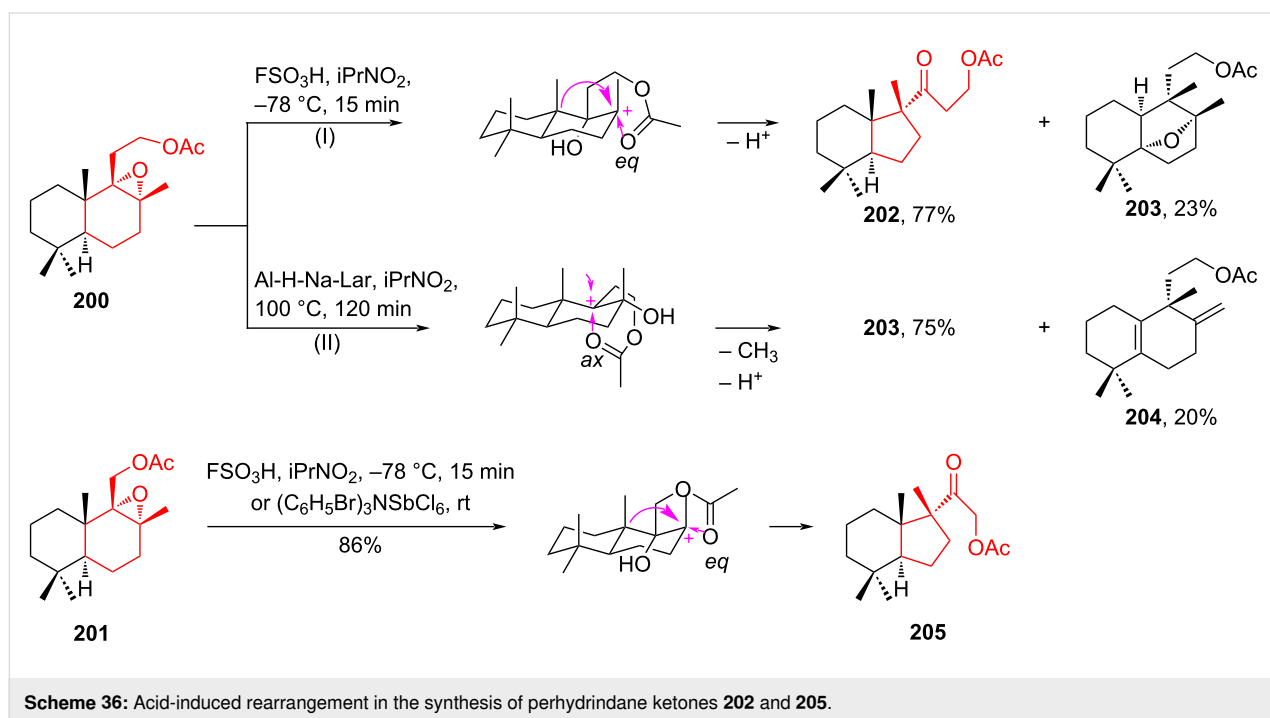
Lewis acid at the first stage. This is accompanied by the contraction of the six-membered ring and the formation of an intermediate **D**. The spatially close reactive centers in molecule **D** promotes tandem transformations with the Lewis acid, resulting in oxygen-containing bicyclic compounds **198** and **199** through the formation of oxonium intermediates **E** and **F**.

In the work of Kulciński et al. [89], a very attractive acid-induced rearrangement of homodrimanic epoxide **200** was described. This reaction proceeded through two alternative directions with the formation of ring-contraction products – perhyrindane **202** (pathway I) and oxide **203**, which has a halimanic bicyclic system (pathway II). The use of fluorosulfonic acid as a promoter at low temperature contributed to ring contraction to

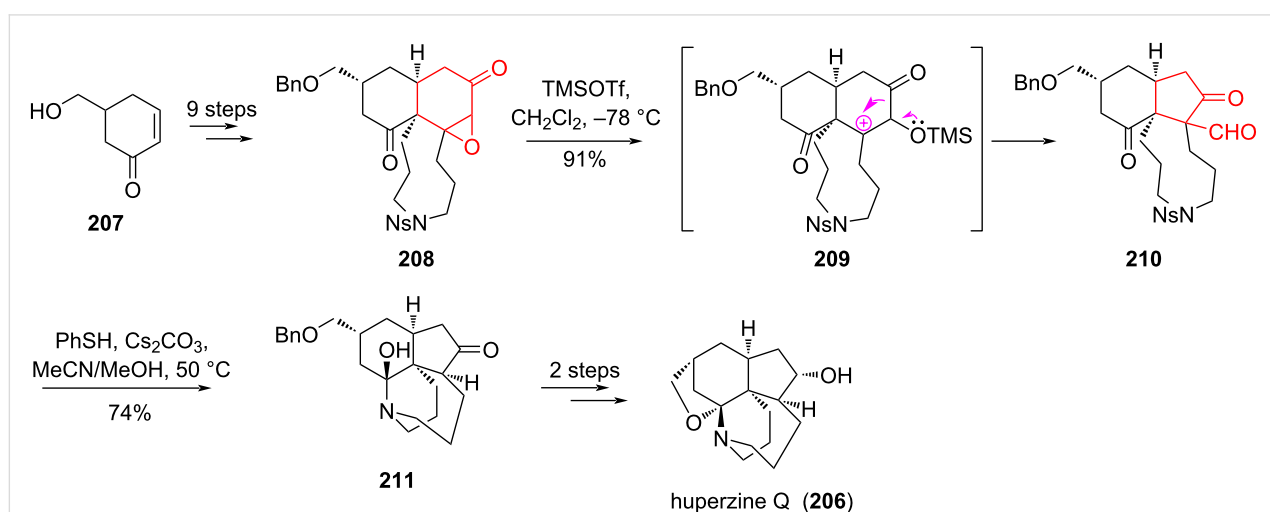
Scheme 35: Rearrangement of α,β -epoxy ketone **197**.

form the perhydrindanic structure. In contrast, the use of Al-H-Na-Lar pillared clay [90] as a heterogeneous catalyst and heating in 2-nitropropane at 100 °C induced angular methyl migration with formation of the halimanic bicyclic system **203** [90]. The acid-induced rearrangement of the drimanic epoxide **201** proceeded exclusively to the ring-contracted perhydrindanic ketone **205** in high to quantitative yields (Scheme 36). The authors suggest that the selectivity of the acid-induced rearrangement of epoxides **200** and **201**, differing only by a single CH₂ group, is influenced by the involvement of the lateral chain in the stabilization of cationic reaction intermediates [91,92].

Yokoshima and co-workers [93,94] developed an elegant approach to the total synthesis of the alkaloid huperzine Q (**206**) in racemic form (Scheme 37). In the process of synthesis, a *cis*-hydrindane core was prepared from the known hydroxyketone **207** using the Diels–Alder reaction and the ring contraction reaction of epoxyketone **208**. After treatment with TMSOTf in dichloromethane at –78 °C, epoxyketone **208** was subjected to selective cleavage of the epoxide, followed by a 1,2-shift of the carbonyl group and the formation of the ring-contracted ketoaldehyde **210** with a yield of 91%. Under the conditions of removal of the nosyl group, cleavage of the formyl group



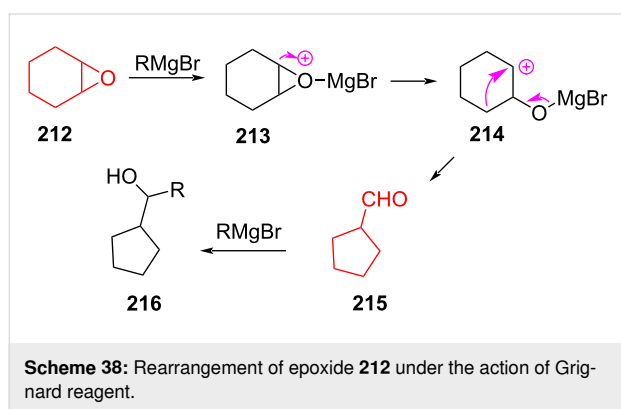
Scheme 36: Acid-induced rearrangement in the synthesis of perhydrindane ketones **202** and **205**.



Scheme 37: Rearrangement of epoxyketone **208** in the synthesis of huperzine Q (**206**).

occurred concomitantly with the addition of methanol to afford hemiaminal **211** with a 74% yield. Subsequent two-step synthetic transformations of **211** yielded huperzine Q (**206**) in 50% yield.

Banerjee et al. [95] demonstrated an alternative method for the synthesis of cyclopentane derivatives. The method included the cleavage of epoxides using Grignard reagents followed by ring contraction. Treatment of epoxide **212** with Grignard reagent through intermediates **213** and **214** produced aldehyde **215**, and further reaction of **215** with the Grignard reagent gave alcohol **216** (Scheme 38).

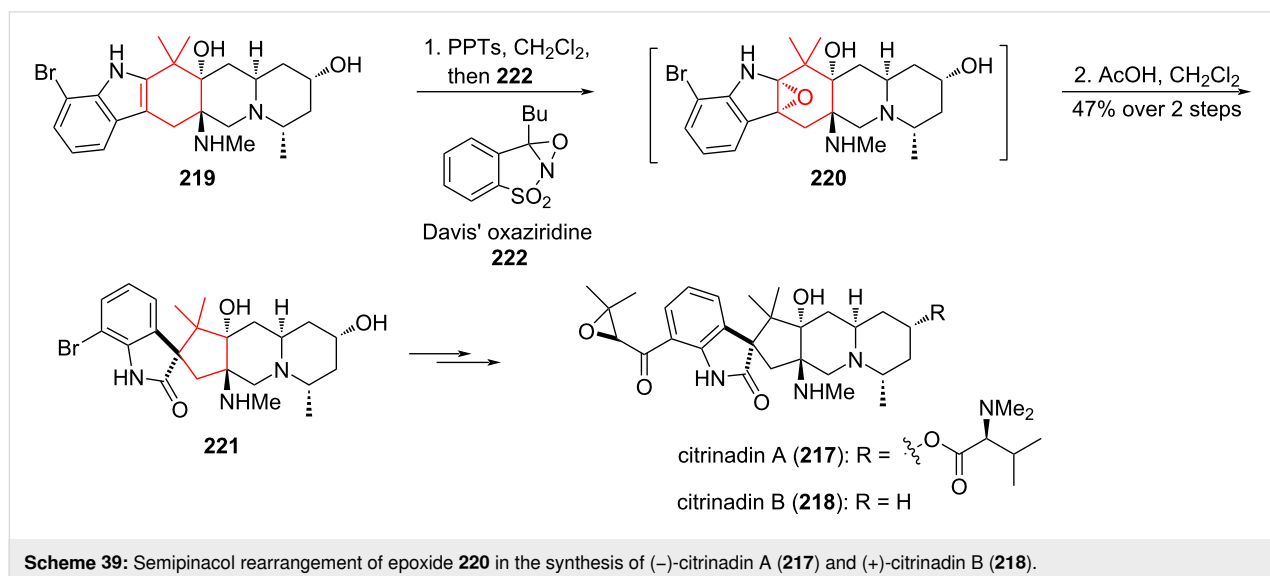


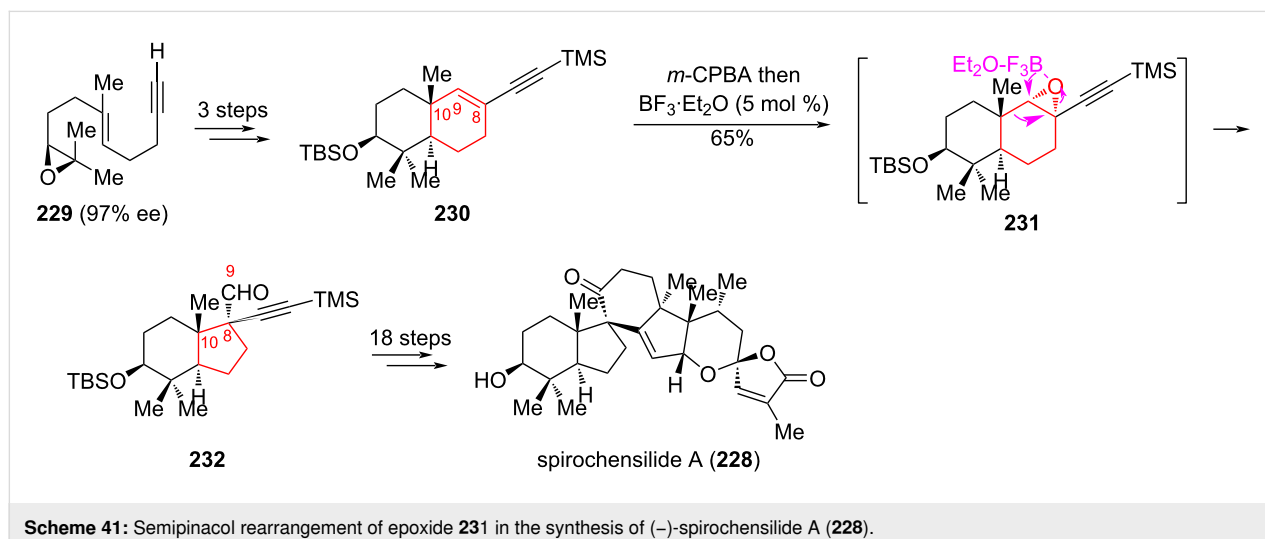
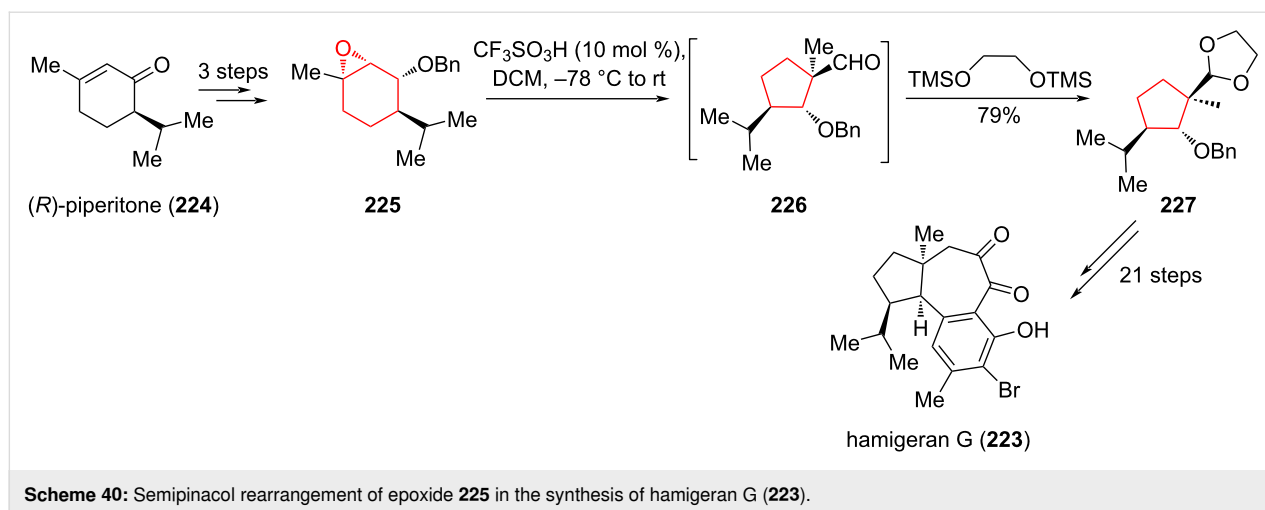
A promising method for the contraction of cyclohexane rings is the semipinacol rearrangement. For the enantioselective total synthesis of (–)-citrinadin A (**217**) and (+)-citrinadin B (**218**), the authors [96] used the semipinacol rearrangement as a key step in converting indole **219** to spirooxindole **221**. The spirooxindole alkaloids, (–)-citrinadin A and (+)-citrinadin B, exhibit notable activity against murine leukemia L1210 (**217**, IC₅₀

6.2 μg/mL; **229**, 10 μg/mL) and human epidermoid KB cells (**218**, IC₅₀ 10 μg/mL). Sequential interaction of indole **219** with pyridinium *p*-toluenesulfonate (PPTS) and excess of Davis' oxaziridine **222**, an effective oxidizing agent for synthesis of spirooxindole alkaloids, led to epoxide **220**. Labile oxirane **220**, without isolation, was subjected to a semipinacol rearrangement with acetic acid. The yield of spirooxindole **221** as a single stereoisomer was 49% (Scheme 39).

In 2016 Gao and co-workers [97], reported the first total synthesis of the halogenated tetraterpenoid hamigeran G (**223**), which inhibited growth of the P388 tumor cell line and the HL-60 promyelocytic leukemia cell line (IC₅₀ 8 μM). The key step in this approach was the acid-promoted semipinacol rearrangement of epoxide **225** under the action of trifluoromethanesulfonic acid to produce aldehyde **226** with a quaternary carbon atom (C9) as a single diastereomer. The addition of 1,2-bis(trimethylsiloxy)ethane to the reaction mixture protected the aldehyde group as a dioxolane, providing cyclopentane **227** with a yield of 79% in two steps (Scheme 40).

In the alternative stereoselective synthesis of (–)-spirochensilide A (**228**), proposed by Liang et al. [98], one of the key steps was the semipinacol rearrangement of alkyne **230** into aldehyde **232**, which contains two vicinal quaternary chiral centers at C8 and C10 (Scheme 41). (–)-Spirochensilide A belongs to an emerging and biologically important class of natural products with a unique spirocyclic core [99–102]. It is also a promising compound for the study of inflammatory diseases and other bioactive properties. Epoxidation of **230** with *m*-chloroperoxybenzoic acid (*m*-CPBA) gave the epoxide **231**. Subsequent addition of excess BF₃·OEt₂ led to a semipinacol rearrangement, affording **232** in 65% yield as a single diastereo-





mer. The aldehyde **232** was then converted by a series of synthetic transformations to (–)-spirochensilide A (**228**) with a total yield of 2.2% in 22 steps starting from acetylenic epoxide **229**.

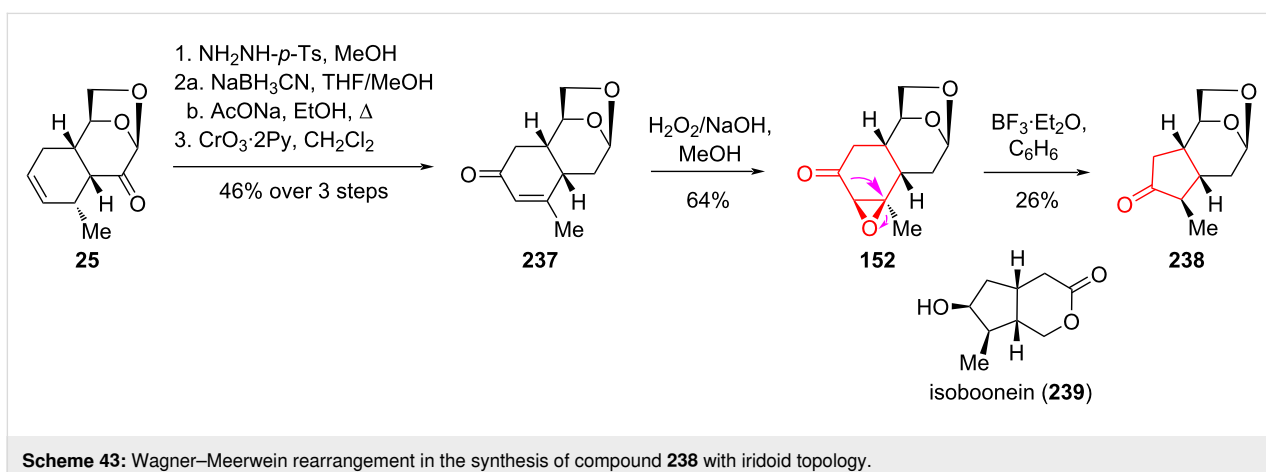
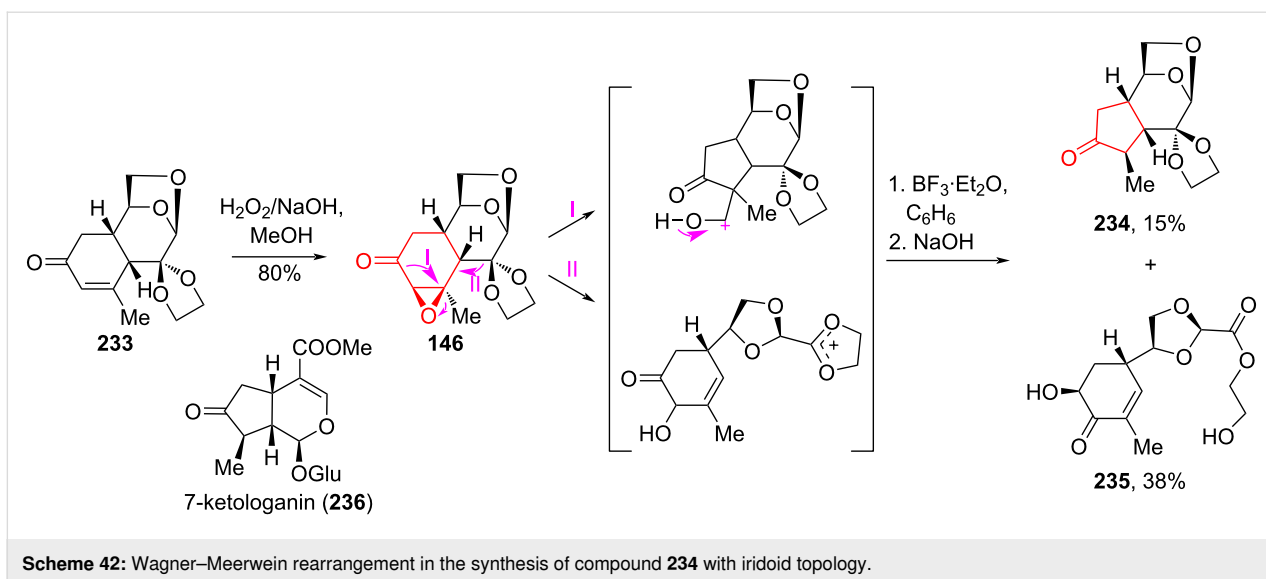
4.1 Wagner–Meerwein rearrangement

The isomerization of terpenes via cleavage, addition or nucleophilic substitution reactions accompanied by a redistribution of C–C bonds in the ring, is called the Wagner–Meerwein rearrangement. Although this rearrangement has been known since the 19th century, only a limited number of articles have been published about it in the last decade.

The Wagner–Meerwein and Favorskii contraction of a cyclohexene ring was studied by Faizullina et al. [76] using the epoxy derivative of enone **233**. The required α -ketooxirane **146** was obtained from enone **233** by oxidation with a mixture of 30% H_2O_2 and NaOH in methanol with a yield of 80% (Scheme 42). After treatment of α -ketooxirane **146** with

$\text{BF}_3 \cdot \text{Et}_2\text{O}$, two compounds, cyclopentanone **234** and the Grob fragmentation product, enone **235**, were isolated from the reaction mixture. Cyclopentanone **234** can be used in the synthesis of 7-ketologanin (**236**) after optimization of the transformation. The product of the Wagner–Meerwein rearrangement, cyclopentanone **234**, was formed according to the classical mechanism, and the process was completed by decarbonylation. Enone **235** was obtained by concurrent Grob fragmentation through an intermediate 1,3-dioxolan-2-ylum ion.

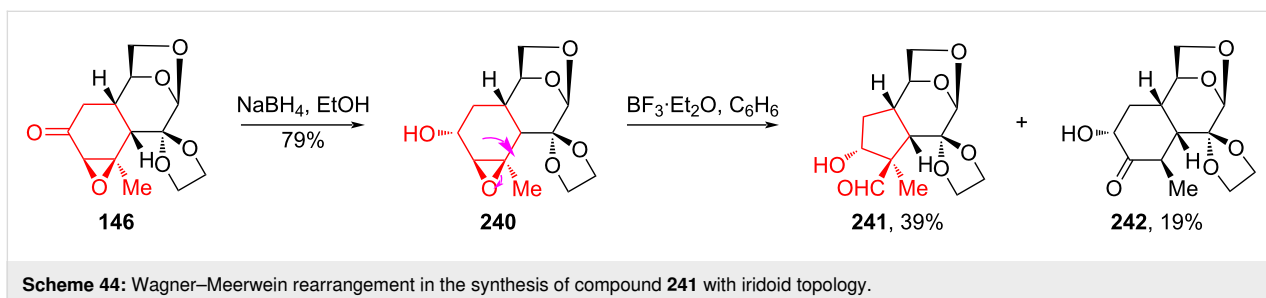
In order to eliminate the influence of the dioxolane fragment on the transformation process, the same authors [76] deoxygenated a ketone **25** by transforming it to the corresponding tosylhydrazone, followed by deamination to give derivative **237** [103]. This derivative was then converted to epoxide **152** using H_2O_2 –NaOH in methanol (Scheme 43). Subsequent treatment of compound **152** with $\text{BF}_3 \cdot \text{Et}_2\text{O}$ in benzene afforded the annelated methylcyclopentanone **238** with a yield of 26%. This



compound can be used in the synthesis of isoboonein (**239**) [104].

In compound **146**, the effect of the keto group is more significant than that of the dioxolane fragment, as is known for this type of compounds. Therefore, changes to this functionality or its removal can affect the course of reactions. To clarify this issue, α -hydroxyoxirane **240** was stereospecifically obtained by

borohydride reduction of α -ketooxirane **146**. Subsequent treatment with $\text{BF}_3 \cdot \text{Et}_2\text{O}$ resulted in the formation of aldehyde **241** with a yield of 39% and hydroxyketone **242** with a 19% yield (Scheme 44). An attempt to perform similar procedures on α -ketooxirane **152** lacking the the dioxolane fragment, led, as expected, to the formation of a difficult-to-separate mixture of products. It is interesting to note that in the studied series of oxiranes, derivatives of the Diels–Alder adduct of LG and



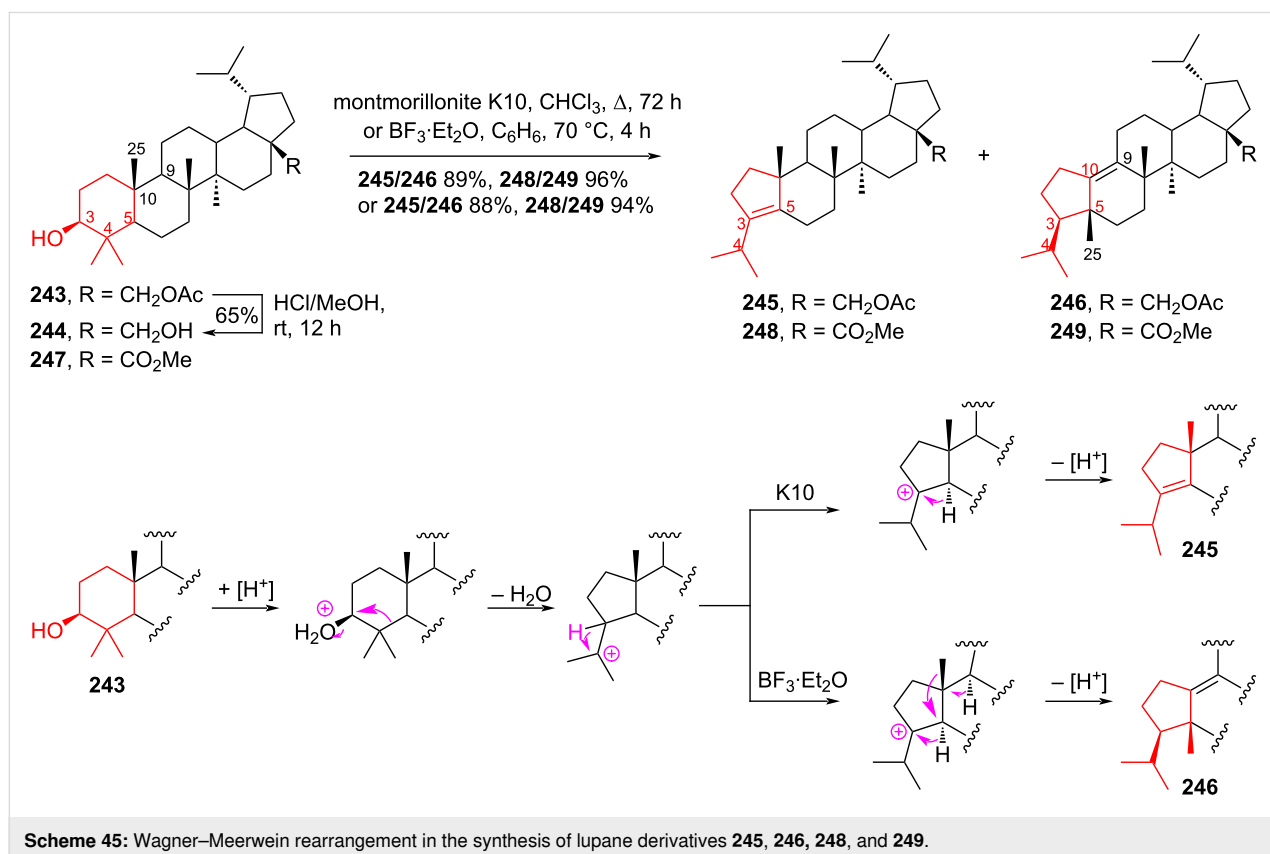
piperylene, the structure of α -hydroxy derivative **240** was found to be most optimal for the Wagner–Meerwein rearrangement under the action of $\text{BF}_3 \cdot \text{Et}_2\text{O}$ to an annelated cyclopentane derivative.

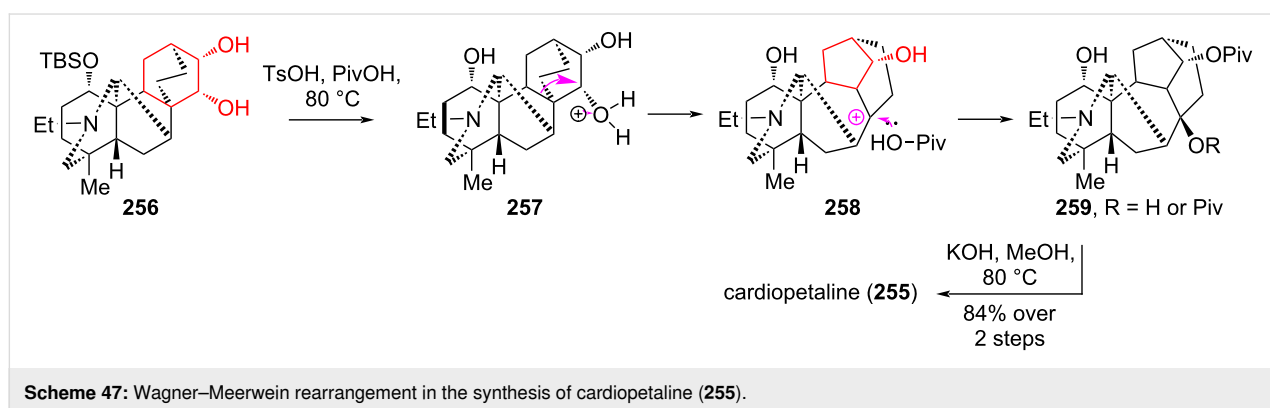
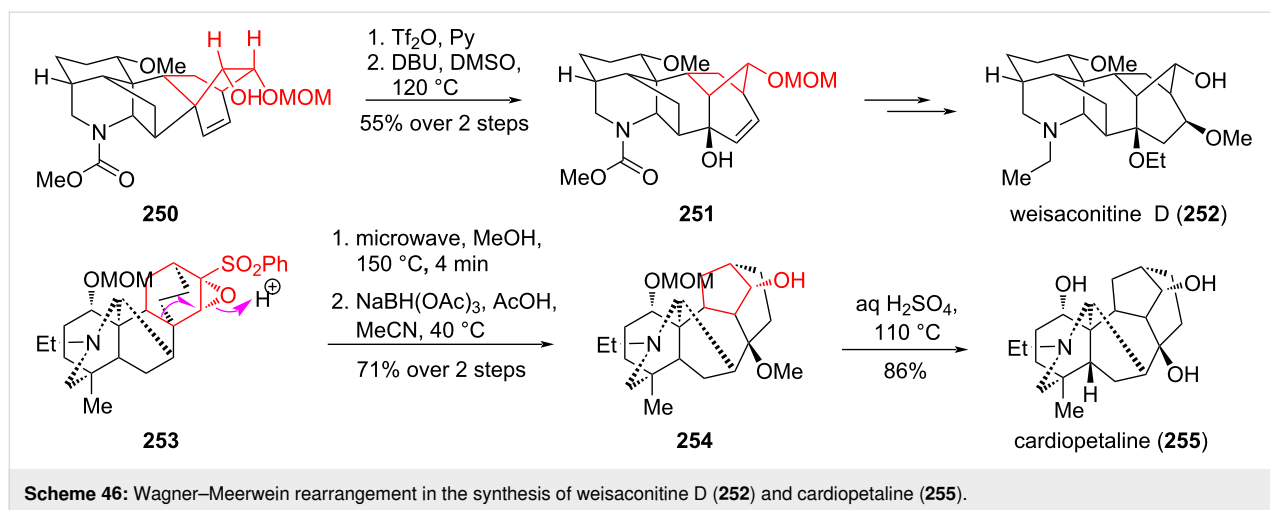
In 2020, Pakulski and co-workers [105] applied a convenient ring-contraction strategy using the Wagner–Meerwein rearrangement to lupane derivatives containing the betulin framework. This reaction was induced by a number of catalysts, including hydrogen chloride, montmorillonite K10, and boron trifluoride etherate. As a result, promising derivatives for the synthesis of ceanothic acid analogues were obtained [106–109]. When treating betulin monoacetate **243** with hydrogen chloride in methanol, the deacetylation of the product to dihydrobetulin **244** was observed with a yield of 65%. Ring contraction was observed when compound **243** was treated with montmorillonite K10 in boiling chloroform, giving an unseparated mixture of abeolupanes **245** and **246**, in a ratio of 1.5–3:1, with a total yield of 89%. The interaction of compound **243** with $\text{BF}_3 \cdot \text{Et}_2\text{O}$ in benzene at 70 °C gave a mixture of olefins **245** and **246** in a 1:2 ratio, with an overall yield of 88%. Similar transformations were carried out with methyl dihydrobetulinate **247**. In the presence of montmorillonite K10, a ring contraction was observed to form an unseparated mixture of olefins **248** and **249** at a ratio of 2–2.5:1, with a total yield of 96%. A highly selec-

tive rearrangement was observed when $\text{BF}_3 \cdot \text{Et}_2\text{O}$ was present to form olefins **249** and **248** in a ratio of >4:1, with a yield of 94% (Scheme 45).

The Wagner–Meerwein rearrangement is widely used in the synthesis of diterpenoid alkaloids of the aconitine-type (Scheme 46). Sarpong et al. [110] used the triflate derived from alcohol **250** as a substrate for the Wagner–Meerwein rearrangement, leading to the synthesis of weisaconitine D (**252**) and lilestrandinin. In 2016, Nishiyama et al. synthesized cardiopetaline (**255**) through the rearrangement of sulfonyloxirane **253** [111].

In 2017, Nishiyama et al. presented an elegant synthesis of cardiopetaline (**255**) by converting the denudatine skeleton to the aconitine skeleton via the Wagner–Meerwein rearrangement of diol **256** [112]. The rearrangement was initiated by heating **256** with *p*-toluenesulfonic acid (TsOH) in pivalic acid (PivOH) to form a mixture of pivalates **259**. The hydrolysis of this mixture allowed cardiopetaline (**255**) to be obtained with a yield of 84% (Scheme 47). This strategy does not require preactivation of the pivotal hydroxy group, the differentiation of several hydroxy groups in the polyoxygenated substrate molecule, and can be applied to the synthesis of highly oxygenated diterpenoid alkaloids of the aconitine-type.





Conclusion

An analysis of literature data over the last 12 years has shown that ring-contraction reactions are widely used in the synthesis of various natural compounds and their structural analogues, which are useful for creating compound libraries. Indeed, ring interconversions play an important role in organic synthesis, allowing the creation of highly functional small carbocycles with high numbers of stereocenters, thereby being a powerful method for the total synthesis of biologically active compounds. The advantages of transformations and ring rearrangements are the reduction of synthetic steps, construction of complex structures of the reaction products, and use of chiral substrates, which solve the problem of controlling optical activity. We hope this review will serve as a stimulus for the further application of well-established ring-contraction reactions and inspire colleagues to develop new, practical and convenient strategies.

Funding

The work was completed on the topic of the state assignment "Development of a strategy and methods for targeted synthesis of practically important substances based on fundamental research into the properties of natural compounds

and organic synthesis products" (state registration No. 125020601627-6).

Author Contributions

Natalya Akhmetdinova: writing – original draft; writing – review & editing. Ilgiz Biktagirov: resources; writing – original draft. Liliya Kh. Faizullina: conceptualization; supervision; writing – review & editing.

ORCID® iDs

Natalya Akhmetdinova - <https://orcid.org/0009-0002-9931-911X>

Ilgiz Biktagirov - <https://orcid.org/0009-0007-2014-1248>

Liliya Kh. Faizullina - <https://orcid.org/0000-0003-2552-1833>

Data Availability Statement

Data sharing is not applicable as no new data was generated or analyzed in this study.

References

- Hajeyah, A. A.; Griffiths, W. J.; Wang, Y.; Finch, A. J.; O'Donnell, V. B. *Front. Endocrinol.* **2020**, *11*, 591819. doi:10.3389/fendo.2020.591819

2. Burstein, S. H. *Prostaglandins Other Lipid Mediators* **2020**, *148*, 106408. doi:10.1016/j.prostaglandins.2020.106408
3. Hu, R.; Gao, J.; Rozimamat, R.; Aisa, H. A. *Eur. J. Med. Chem.* **2018**, *146*, 157–170. doi:10.1016/j.ejmech.2018.01.027
4. Fattahian, M.; Ghanadian, M.; Ali, Z.; Khan, I. A. *Phytochem. Rev.* **2020**, *19*, 265–336. doi:10.1007/s11101-020-09667-8
5. Chawla, P. A.; Grover, P.; Mehta, L.; Malhotra, A.; Kapoor, G.; Nagarajan, K.; Kumar, P.; Chawla, V. *Curr. Top. Med. Chem.* **2023**, *23*, 371–388. doi:10.2174/1568026623666221222142217
6. Segawa, T.; Teramoto, S.; Omi, K.; Miyauchi, O.; Watanabe, Y.; Osada, H. *Reprod. Biol. Endocrinol.* **2015**, *13*, 54. doi:10.1186/s12958-015-0051-y
7. Thawabteh, A. M.; Thawabteh, A.; Lelario, F.; Bufo, S. A.; Scrano, L. *Molecules* **2021**, *26*, 4103. doi:10.3390/molecules26134103
8. Ramaiah, M. *Synthesis* **1984**, 529–570. doi:10.1055/s-1984-30893
9. Loza, V. V.; Gimazetdinov, A. M.; Miftakhov, M. S. *Russ. J. Org. Chem.* **2018**, *54*, 1585–1629. doi:10.1134/s1070428018110015
10. Sanogo, Y.; Allievi, L.; Lecourt, C.; Dhambri, S.; Ardisson, J.; Sorin, G.; Lannou, M.-I. *Tetrahedron* **2016**, *72*, 3369–3377. doi:10.1016/j.tet.2016.04.005
11. Kučera, R.; Ellis, S. R.; Yamazaki, K.; Hayward Cooke, J.; Chekshin, N.; Christensen, K. E.; Hamlin, T. A.; Dixon, D. J. *J. Am. Chem. Soc.* **2023**, *145*, 5422–5430. doi:10.1021/jacs.2c13710
12. Dorel, R.; Echavarren, A. M. *J. Org. Chem.* **2016**, *81*, 8444–8454. doi:10.1021/acs.joc.6b01607
13. Guardia, J. J.; Tapia, R.; Mahdjour, S.; Rodríguez-Serrano, F.; Mut-Salud, N.; Chahboun, R.; Alvarez-Manzaneda, E. *J. Nat. Prod.* **2017**, *80*, 308–318. doi:10.1021/acs.jnatprod.6b00700
14. Graham, M. S.; McErlean, C. S. P. *Asian J. Org. Chem.* **2021**, *10*, 2274–2286. doi:10.1002/ajoc.202100303
15. Mut-Salud, N.; Guardia, J. J.; Fernández, A.; Blancas, I.; Zentar, H.; Garrido, J. M.; Álvarez-Manzaneda, E.; Chahboun, R.; Rodríguez-Serrano, F. *Biomed. Pharmacother.* **2023**, *168*, 115791. doi:10.1016/j.biopha.2023.115791
16. Isakova, M. M.; Biktagirov, I. M.; Faizullina, L. K.; Salikhov, S. M.; Safarov, M. G.; Valeev, F. A. *Russ. J. Org. Chem.* **2014**, *50*, 105–109. doi:10.1134/s1070428014010205
17. Huang, D.-Y.; Luo, Y.-X.; Zheng, W.-D.; Wu, S.-Y.; Huang, P.-Q.; Jin, J.-W.; Wu, P.-P.; Gan, L.-S. *J. Nat. Med.* **2024**, *78*, 1003–1012. doi:10.1007/s11418-024-01820-3
18. Dinda, B.; Dinda, M.; Kulsi, G.; Chakraborty, A.; Dinda, S. *Eur. J. Med. Chem.* **2019**, *169*, 185–199. doi:10.1016/j.ejmech.2019.03.009
19. Biktagirov, I. M.; Faizullina, L. K.; Salikhov, S. M.; Safarov, M. G.; Valeev, F. A. *Russ. J. Org. Chem.* **2015**, *51*, 1457–1463. doi:10.1134/s1070428015100188
20. DaBell, P.; Thomas, S. P. *Synthesis* **2020**, *52*, 949–963. doi:10.1055/s-0039-1690813
21. Bello, O.; Ogbesejana, A.; Osibemhe, M. *J. Appl. Sci. Environ. Manage.* **2018**, *22*, 373–378. doi:10.4314/jasem.v22i3.14
22. Kumar, V.; Chauhan, R. S.; Tandon, C. J. *Plant Biochem. Biotechnol.* **2017**, *26*, 1–13. doi:10.1007/s13562-016-0364-8
23. Worawittayanon, P.; Ruadreo, J.; Disadee, W.; Sahakitpichan, P.; Sitthimonchai, S.; Thasana, N.; Ruchirawat, S.; Kanchanapoom, T. *Biochem. Syst. Ecol.* **2012**, *40*, 38–42. doi:10.1016/j.bse.2011.08.016
24. Ma, W.; Wang, K.-J.; Cheng, C.-S.; Yan, G.-q.; Lu, W.-L.; Ge, J.-F.; Cheng, Y.-X.; Li, N. *J. Ethnopharmacol.* **2014**, *153*, 840–845. doi:10.1016/j.jep.2014.03.051
25. Li, Y.; Zhai, X.; Ma, L.; Zhao, L.; An, N.; Feng, W.; Huang, L.; Zheng, X. *Genes* **2024**, *15*, 155. doi:10.3390/genes15020155
26. Grishko, V. V.; Tolmacheva, I. A.; Pereslavl'tseva, A. V. *Chem. Nat. Compd.* **2015**, *51*, 1–21. doi:10.1007/s10600-015-1193-z
27. Hill, R. A.; Connolly, J. D. *Nat. Prod. Rep.* **2020**, *37*, 962–998. doi:10.1039/c9np00067d
28. Khudobko, M. V.; Mikhailova, L. R.; Baltina, L. A., Jr.; Spirikhin, L. V.; Baltina, L. A. *Chem. Nat. Compd.* **2011**, *47*, 76–78. doi:10.1007/s10600-011-9834-3
29. Baltina, L. A.; Lai, H.-C.; Liu, Y.-C.; Huang, S.-H.; Hour, M.-J.; Baltina, L. A.; Nugumanov, T. R.; Borisevich, S. S.; Khalilov, L. M.; Petrova, S. F.; Khursan, S. L.; Lin, C.-W. *Bioorg. Med. Chem.* **2021**, *41*, 116204. doi:10.1016/j.bmc.2021.116204
30. Baltina, L. A.; Khushnudinova, N. S.; Karimova, E. R.; Saifullina, D. R.; Kondratenko, R. M. *Pharm. Chem. J.* **2023**, *57*, 637–641. doi:10.1007/s11094-023-02931-z
31. Baltina, L. A.; Komissarova, N. G. *Stud. Nat. Prod. Chem.* **2023**, *76*, 331–407. doi:10.1016/b978-0-323-91296-9.00001-0
32. Mydock-McGrane, L.; Rath, N. P.; Covey, D. F. *J. Org. Chem.* **2014**, *79*, 5636–5643. doi:10.1021/jo500813n
33. Fantini, J.; Epand, R. M.; Barrantes, F. J. *Adv. Exp. Med. Biol.* **2019**, *3–25*. doi:10.1007/978-3-030-14265-0_1
34. Fantini, J. *Receptors* **2024**, *3*, 107–121. doi:10.3390/receptors3010006
35. Konyshova, A. V.; Eroshenko, D. V.; Grishko, V. V. *Nat. Prod. Commun.* **2019**, *14*, 1–7. doi:10.1177/1934578x19878919
36. Borkova, L.; Hodon, J.; Urban, M. *Asian J. Org. Chem.* **2018**, *7*, 1542–1560. doi:10.1002/ajoc.201800163
37. Hui, C.; Craggs, L.; Antonchick, A. P. *Chem. Soc. Rev.* **2022**, *51*, 8652–8675. doi:10.1039/d1cs01080h
38. Zhang, W.; Zhou, Z.-X.; Zhu, X.-J.; Sun, Z.-H.; Dai, W.-M.; Li, C.-C. *J. Am. Chem. Soc.* **2020**, *142*, 19868–19873. doi:10.1021/jacs.0c10116
39. Qiao, C.; Zhang, W.; Han, J.-C.; Li, C.-C. *Org. Lett.* **2016**, *18*, 4932–4935. doi:10.1021/acs.orglett.6b02414
40. Liao, W.; Yu, Z.-X. *J. Org. Chem.* **2014**, *79*, 11949–11960. doi:10.1021/jo5017844
41. Wu, H.; Wang, Q.; Zhu, J. *Angew. Chem., Int. Ed.* **2020**, *59*, 7261–7265. doi:10.1002/anie.202001258
42. Jana, C. K.; Scopelliti, R.; Gademann, K. *Chem. – Eur. J.* **2010**, *16*, 7692–7695. doi:10.1002/chem.201001085
43. Zhao, Y.; Li, Y.; Wang, B.; Zhao, J.; Li, L.; Luo, X.-D.; Zhang, H. *J. Org. Chem.* **2020**, *85*, 13772–13778. doi:10.1021/acs.joc.0c01900
44. Wang, B.; Zhao, Y.-J.; Zhao, Y.-L.; Liu, Y.-P.; Li, X.-N.; Zhang, H.; Luo, X.-D. *Org. Lett.* **2020**, *22*, 257–260. doi:10.1021/acs.orglett.9b04252
45. Odagi, M.; Matoba, T.; Hosoya, K.; Nagasawa, K. *J. Am. Chem. Soc.* **2021**, *143*, 2699–2704. doi:10.1021/jacs.1c00047
46. Li, G.; Wang, Q.; Zhu, J. *Nat. Commun.* **2021**, *12*, 36. doi:10.1038/s41467-020-20274-1
47. Gao, W.; Jiang, J.-S.; Chen, Z.; Yang, Y.-N.; Feng, Z.-M.; Zhang, X.; Yuan, X.; Zhang, P.-C. *Org. Chem. Front.* **2019**, *6*, 1858–1862. doi:10.1039/c9qo00279k
48. Yu, L.; Liu, Z.; He, W.; Chen, H.; Lai, Z.; Duan, Y.; Cao, X.; Tao, J.; Xu, C.; Zhang, Q.; Zhao, Z.; Zhang, J. *Cell. Mol. Neurobiol.* **2020**, *40*, 1271–1281. doi:10.1007/s10571-020-00812-7
49. Zhang, Y.; Ji, Y.; Franzoni, I.; Guo, C.; Jia, H.; Hong, B.; Li, H. *Angew. Chem., Int. Ed.* **2021**, *60*, 14869–14874. doi:10.1002/anie.202104014

50. Matsuda, Y.; Abe, I. *Nat. Prod. Rep.* **2016**, *33*, 26–53. doi:10.1039/c5np00090d
51. Nagaraju, C.; Prasad, K. R. *Angew. Chem., Int. Ed.* **2014**, *53*, 10997–11000. doi:10.1002/anie.201407680
52. Yu, X.; Hu, J.; Shen, Z.; Zhang, H.; Gao, J.-M.; Xie, W. *Angew. Chem., Int. Ed.* **2017**, *56*, 350–353. doi:10.1002/anie.201609975
53. Xu, L.-P.; Hou, X.-D.; Wang, H.; Zhou, J.-B.; Ma, J.; Zhuo, S.; Wang, Q. *Chem. – Asian J.* **2019**, *14*, 4262–4267. doi:10.1002/asia.201901263
54. Liu, M.; Wang, W.-G.; Sun, H.-D.; Pu, J.-X. *Nat. Prod. Rep.* **2017**, *34*, 1090–1140. doi:10.1039/c7np00027h
55. Chen, R.; Zhang, F.; Hua, Y.; Shi, D.; Lei, X.; Xiao, H.; Wang, Y.; Ding, S.; Shen, Y.; Zhang, Y. *CCS Chem.* **2022**, *4*, 987–995. doi:10.31635/ccschem.021.202100821
56. Xu, G.; Elkin, M.; Tantillo, D. J.; Newhouse, T. R.; Maimone, T. J. *Angew. Chem., Int. Ed.* **2017**, *56*, 12498–12502. doi:10.1002/anie.201705654
57. Ting, C. P.; Xu, G.; Zeng, X.; Maimone, T. J. *J. Am. Chem. Soc.* **2016**, *138*, 14868–14871. doi:10.1021/jacs.6b10397
58. Khan, A.; Silva, L. F.; Rabnawaz, M. *New J. Chem.* **2021**, *45*, 2078–2084. doi:10.1039/d0nj04700g
59. Silva, L. F., Jr. *Synlett* **2014**, *25*, 466–476. doi:10.1055/s-0033-1340472
60. Gabriele, B.; Mancuso, R.; Veltri, L. *Chem. – Eur. J.* **2016**, *22*, 5056–5094. doi:10.1002/chem.201503933
61. Ahmad, A.; Silva, L. F., Jr. *Synthesis* **2012**, *44*, 3671–3677. doi:10.1055/s-0032-1317497
62. Farid, U.; Malmedy, F.; Claveau, R.; Albers, L.; Wirth, T. *Angew. Chem., Int. Ed.* **2013**, *52*, 7018–7022. doi:10.1002/anie.201302358
63. Yoshimura, A.; Zhdarkin, V. V. *Chem. Rev.* **2016**, *116*, 3328–3435. doi:10.1021/acs.chemrev.5b00547
64. Deng, J.; Li, R.; Luo, Y.; Li, J.; Zhou, S.; Li, Y.; Hu, J.; Li, A. *Org. Lett.* **2013**, *15*, 2022–2025. doi:10.1021/ol400717h
65. Deng, J.; Zhou, S.; Zhang, W.; Li, J.; Li, R.; Li, A. *J. Am. Chem. Soc.* **2014**, *136*, 8185–8188. doi:10.1021/ja503972p
66. Qiao, T.; Wang, Y.; Zheng, S.; Kang, H.; Liang, G. *Angew. Chem., Int. Ed.* **2020**, *59*, 14111–14114. doi:10.1002/anie.202005600
67. Tao, D. J.; Slutskyy, Y.; Muuronen, M.; Le, A.; Kohler, P.; Overman, L. E. *J. Am. Chem. Soc.* **2018**, *140*, 3091–3102. doi:10.1021/jacs.7b13799
68. Rao, P.; Hu, J.; Xuan, J.; Ding, H. *J. Org. Chem.* **2019**, *84*, 9385–9392. doi:10.1021/acs.joc.9b01308
69. Shen, S.; Zhu, H.; Chen, D.; Liu, D.; van Ofwegen, L.; Proksch, P.; Lin, W. *Tetrahedron Lett.* **2012**, *53*, 5759–5762. doi:10.1016/j.tetlet.2012.08.049
70. Zhang, P.-P.; Yan, Z.-M.; Li, Y.-H.; Gong, J.-X.; Yang, Z. *J. Am. Chem. Soc.* **2017**, *139*, 13989–13992. doi:10.1021/jacs.7b07388
71. Hu, P.; Snyder, S. A. *J. Am. Chem. Soc.* **2017**, *139*, 5007–5010. doi:10.1021/jacs.7b01454
72. Kärkäs, M. D.; Porco, J. A., Jr.; Stephenson, C. R. *J. Chem. Rev.* **2016**, *116*, 9683–9747. doi:10.1021/acs.chemrev.5b00760
73. Zhou, Q.-Q.; Zou, Y.-Q.; Lu, L.-Q.; Xiao, W.-J. *Angew. Chem., Int. Ed.* **2019**, *58*, 1586–1604. doi:10.1002/anie.201803102
74. Biktagirov, I. M.; Faizullina, L. K.; Salikhov, S. M.; Galin, F. Z.; Valeev, F. A. *Mendeleev Commun.* **2017**, *27*, 237–239. doi:10.1016/j.mencom.2017.05.006
75. Wadood, A.; Kim, H.; Park, C. M.; Song, J.-H.; Lee, S. *Arch. Pharmacol. Res.* **2015**, *38*, 2029–2041. doi:10.1007/s12272-015-0623-6
76. Biktagirov, I. M.; Faizullina, L. K.; Salikhov, S. M.; Valeev, F. A. *Russ. J. Org. Chem.* **2016**, *52*, 1468–1474. doi:10.1134/s107042801610016x
77. Doronina, O. Y.; Khalilova, Y. A.; Sharipov, B. T.; Spirikhin, L. V.; Valeev, F. A. *Russ. J. Org. Chem.* **2012**, *48*, 1419–1423. doi:10.1134/s1070428012110036
78. Majhi, S. *Photochem. Photobiol. Sci.* **2021**, *20*, 1357–1378. doi:10.1007/s43630-021-00100-3
79. Liu, W.; Hong, B.; Wang, J.; Lei, X. *Acc. Chem. Res.* **2020**, *53*, 2569–2586. doi:10.1021/acs.accounts.0c00531
80. Yu, X.-Y.; Chen, J.-R.; Xiao, W.-J. *Chem. Rev.* **2021**, *121*, 506–561. doi:10.1021/acs.chemrev.0c00030
81. Jurczyk, J.; Lux, M. C.; Adpressa, D.; Kim, S. F.; Lam, Y.-h.; Yeung, C. S.; Sarpong, R. *Science* **2021**, *373*, 1004–1012. doi:10.1126/science.abi7183
82. Rauscher, N.; Næsborg, L.; Jandl, C.; Bach, T. *Angew. Chem., Int. Ed.* **2021**, *60*, 24039–24042. doi:10.1002/anie.202110009
83. Gorobets, E.; Papatzimas, J. W.; Dourado, J.; Yousefalizadeh, G.; Lee, J.; Brownsey, D. K.; Stamplecoskie, K.; Davis, R.; Derksen, D. J. *Chem. Commun.* **2022**, *58*, 2910–2913. doi:10.1039/d2cc00513a
84. Dotson, J. J.; Perez-Estrada, S.; Garcia-Garibay, M. A. *J. Am. Chem. Soc.* **2018**, *140*, 8359–8371. doi:10.1021/jacs.8b03988
85. Jin, S.; Nguyen, V. T.; Dang, H. T.; Nguyen, D. P.; Arman, H. D.; Larionov, O. V. *J. Am. Chem. Soc.* **2017**, *139*, 11365–11368. doi:10.1021/jacs.7b07128
86. Newman, D. J.; Cragg, G. M. *J. Nat. Prod.* **2016**, *79*, 629–661. doi:10.1021/acs.jnatprod.5b01055
87. Zaki, M.; Tebbaa, M.; Hiebel, M.-A.; Benharref, A.; Akssira, M.; Berteina-Raboin, S. *Tetrahedron* **2015**, *71*, 2035–2042. doi:10.1016/j.tet.2014.12.061
88. Wang, L.-F.; Shi, Z.-F.; Cao, X.-P.; Li, B.-S.; An, P. *Chem. Commun.* **2014**, *50*, 8061–8064. doi:10.1039/c4cc02641a
89. Sîrbu, T.; Girbu, V.; Harghel, P.; Rusu, V.; Ungur, N.; Kulcički, V. *Synthesis* **2019**, *51*, 1995–2000. doi:10.1055/s-0037-1610686
90. Rusu, V.; Maftuleac, A.; Petuhov, O.; Kulcički, V.; Grincio, M.; Postolachi, L.; Rusu, M. *Stud. Univ. Mold. Stiinte reale si ale naturii* **2015**, *1*, 178–185.
91. Potter, K. C.; Jia, M.; Hong, Y. J.; Tantillo, D.; Peters, R. *J. Org. Lett.* **2016**, *18*, 1060–1063. doi:10.1021/acs.orglett.6b00181
92. Xu, M.; Jia, M.; Hong, Y. J.; Yin, X.; Tantillo, D. J.; Proteau, P. J.; Peters, R. *J. Org. Lett.* **2018**, *20*, 1200–1202. doi:10.1021/acs.orglett.8b00121
93. Tanimura, S.; Yokoshima, S.; Fukuyama, T. *Org. Lett.* **2017**, *19*, 3684–3686. doi:10.1021/acs.orglett.7b01633
94. Yokoshima, S. *Synlett* **2020**, *31*, 1967–1975. doi:10.1055/s-0040-1707904
95. Banerjee, A. K.; Arrieche, D.; Bedoya, L.; Vera, W. J.; Cabrera, E. V. *J. Pharm. Drug Res.* **2020**, *3*, 233–237.
96. Bian, Z.; Marvin, C. C.; Pettersson, M.; Martin, S. F. *J. Am. Chem. Soc.* **2014**, *136*, 14184–14192. doi:10.1021/ja5074646
97. Li, X.; Xue, D.; Wang, C.; Gao, S. *Angew. Chem., Int. Ed.* **2016**, *55*, 9942–9946. doi:10.1002/anie.201604070
98. Liang, X.-T.; Chen, J.-H.; Yang, Z. *J. Am. Chem. Soc.* **2020**, *142*, 8116–8121. doi:10.1021/jacs.0c02522
99. Hill, R. A.; Sutherland, A. *Nat. Prod. Rep.* **2015**, *32*, 1165–1169. doi:10.1039/c5np90032h

100. Ferreira, R. J.; Kincses, A.; Gajdács, M.; Spengler, G.; dos Santos, D. J. V. A.; Molnár, J.; Ferreira, M.-J. U. *J. Nat. Prod.* **2018**, *81*, 2032–2040. doi:10.1021/acs.jnatprod.8b00326
101. Umeyama, A.; Ohta, C.; Shino, Y.; Okada, M.; Nakamura, Y.; Hamagaki, T.; Imagawa, H.; Tanaka, M.; Ishiyama, A.; Iwatsuki, M.; Otaguro, K.; Ōmura, S.; Hashimoto, T. *Tetrahedron* **2014**, *70*, 8312–8315. doi:10.1016/j.tet.2014.09.013
102. Tang, Y.; Zhao, Z.-Z.; Yao, J.-N.; Feng, T.; Li, Z.-H.; Chen, H.-P.; Liu, J.-K. *J. Nat. Prod.* **2018**, *81*, 2163–2168. doi:10.1021/acs.jnatprod.7b00845
103. Biktagirov, I. M.; Faizullina, L. K.; Salikhov, S. M.; Galin, F. Z.; Safarov, M. G.; Valeev, F. A. *Russ. J. Org. Chem.* **2016**, *52*, 711–720. doi:10.1134/s107042801605016x
104. Kaurav, M. S.; Kumar, V.; Kumar, D.; Khan, T. *ChemistrySelect* **2018**, *3*, 5566–5570. doi:10.1002/slct.201800421
105. Pakulski, Z.; Cmoch, P.; Korda, A.; Luboradzki, R.; Gwardiak, K.; Karczewski, R. *J. Org. Chem.* **2021**, *86*, 1084–1095. doi:10.1021/acs.joc.0c02560
106. Liu, S.-J.; Lv, Y.-P.; Tang, Z.-S.; Zhang, Y.; Xu, H.-B.; Zhang, D.-B.; Cui, C.-L.; Liu, H.-B.; Sun, H.-H.; Song, Z.-X.; Wei, S.-M. *Phytochem. Rev.* **2021**, *20*, 507–541. doi:10.1007/s11101-020-09709-1
107. Hordyjewska, A.; Ostapiuk, A.; Horecka, A.; Kurzepa, J. *Phytochem. Rev.* **2019**, *18*, 929–951. doi:10.1007/s11101-019-09623-1
108. Amiri, S.; Dastghaib, S.; Ahmadi, M.; Mehrbod, P.; Khadem, F.; Behrouj, H.; Aghanoori, M.-R.; Machaj, F.; Ghamsari, M.; Rosik, J.; Hudecki, A.; Afkhami, A.; Hashemi, M.; Los, M. J.; Mokarram, P.; Madrakian, T.; Ghavami, S. *Biotechnol. Adv.* **2020**, *38*, 107409. doi:10.1016/j.biotechadv.2019.06.008
109. Castro, M. J.; Careaga, V. P.; Sacca, P. A.; Faraoni, M. B.; Murray, A. P.; Calvo, J. C. *Anticancer Res.* **2019**, *39*, 3835–3845. doi:10.21873/anticancer.13533
110. Marth, C. J.; Gallego, G. M.; Lee, J. C.; Lebold, T. P.; Kulyk, S.; Kou, K. G. M.; Qin, J.; Lilien, R.; Sarpong, R. *Nature* **2015**, *528*, 493–498. doi:10.1038/nature16440
111. Nishiyama, Y.; Yokoshima, S.; Fukuyama, T. *Org. Lett.* **2016**, *18*, 2359–2362. doi:10.1021/acs.orglett.6b00789
112. Nishiyama, Y.; Yokoshima, S.; Fukuyama, T. *Org. Lett.* **2017**, *19*, 5833–5835. doi:10.1021/acs.orglett.7b02812

License and Terms

This is an open access article licensed under the terms of the Beilstein-Institut Open Access License Agreement (<https://www.beilstein-journals.org/bjoc/terms>), which is identical to the Creative Commons Attribution 4.0 International License (<https://creativecommons.org/licenses/by/4.0>). The reuse of material under this license requires that the author(s), source and license are credited. Third-party material in this article could be subject to other licenses (typically indicated in the credit line), and in this case, users are required to obtain permission from the license holder to reuse the material.

The definitive version of this article is the electronic one which can be found at:
<https://doi.org/10.3762/bjoc.21.185>