



## Pd-Catalyzed asymmetric allylic amination with isatin using a P,olefin-type chiral ligand with C–N bond axial chirality

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### Full Research Paper

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### Abstract

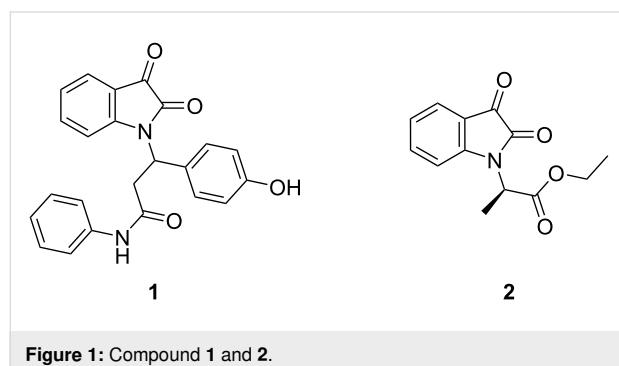
In this study, we implemented the P,olefin-type chiral ligand (*aR*)-**6**, which contains a cyclohexyl group and a cinnamoyl group on the nitrogen atom, in the Pd-catalyzed asymmetric allylic amination of allylic esters with isatin derivatives **11** as nucleophiles. The reaction proceeds efficiently, yielding the products (*S*)-**13** with good-to-high enantioselectivity. A scale-up reaction was also successfully conducted at a 1 mmol scale. Additionally, when malononitrile was added to the resulting product (*S*)-**13a** in the presence of  $\text{FeCl}_3$  as the catalyst, the corresponding malononitrile derivative (*S*)-**16** was obtained without any loss in optical purity.

### Introduction

Isatin is a well-known natural indole derivative. Due to the broad biological activities of its derivatives, extensive research has been conducted on their synthesis. Furthermore, the isatin framework is a versatile starting material for various transformations, including multicomponent reactions and the synthesis of spirocyclic compounds [1-3]. The nucleophilicity of isatin at

the nitrogen atom allows it to participate in reactions such as alkylation [4], arylation [5], and *aza*-Michael addition [6-8]. However, the products obtained from these reactions are primarily achiral or racemic, and only a few studies have reported the use of isatin as a nucleophile in asymmetric reactions [9-11]. On the other hand, it has been revealed that com-

pounds in which the carbon bonded to the nitrogen atom of newly constructed *N*-substituted isatin becomes a chiral center exhibit pharmacological properties in medicinal chemistry. For example, racemic compound **1** (Figure 1) was evaluated for its cytotoxicity against human breast cancer cells (MCF7) in comparison to the standard doxorubicin and exhibited excellent activity against the MCF7 cell line [12]. The optically active compound **2** also showed activity against Huh7.5-FGR-JC1-Rluc2A cells, which carry HCV gt 2a [13].



Therefore, developing asymmetric reactions that simultaneously form a carbon–nitrogen bond and construct a chiral center is of great importance. Although a relatively large number of asymmetric allylic amination reactions using palladium catalysts with amines as nucleophiles have been reported [14–25], there have been only a few reports on the *N*-substitution of isatin using asymmetric methods. Recently, Wolf's group reported a transition-metal-catalyzed (Pd-catalyzed) asymmetric allylic amination of allyl esters using isatin as a nucleophile. In this reaction, bisphosphine-type ligands such as BINAP and SEGPHOS derivatives, as well as P,N-type ligands like oxazoline-type ligands, were utilized as chiral ligands [26]. On the other hand, several groups have recently reported new chiral ligands with axial chirality for Pd-catalyzed asymmetric allylic substitution reactions. For example, the Zhou group reported a P,olefin-type chiral ligand **3** with C–C bond axial chirality for this reaction (Figure 2) [27]. Additionally, we have recently reported chiral ligands with C–N bond axial chirality, such as *N*-alkyl-*N*-cinnamyl-type chiral ligands **4** [28,29] and **5** [30], and a P,olefin-type chiral ligand **6** [31] with a cinnamoyl group instead of a cinnamyl group. In particular, the chiral ligand **6** is effective in the Pd-catalyzed asymmetric allylic substitution reaction of allylic esters with indoles. Here, we describe the Pd-catalyzed asymmetric allylic amination of allylic esters with isatin as a nucleophile using chiral ligand **6** and its derivative **7**. Compared to chiral ligand **6**, which has a secondary alkyl group (cyclohexyl) as a substituent on the nitrogen and has already been reported, compound **7** has a primary alkyl group (*n*-propyl). This difference reduces steric hindrance and lowers

the rotational barrier around the carbon–nitrogen bond, increasing the likelihood of racemization.

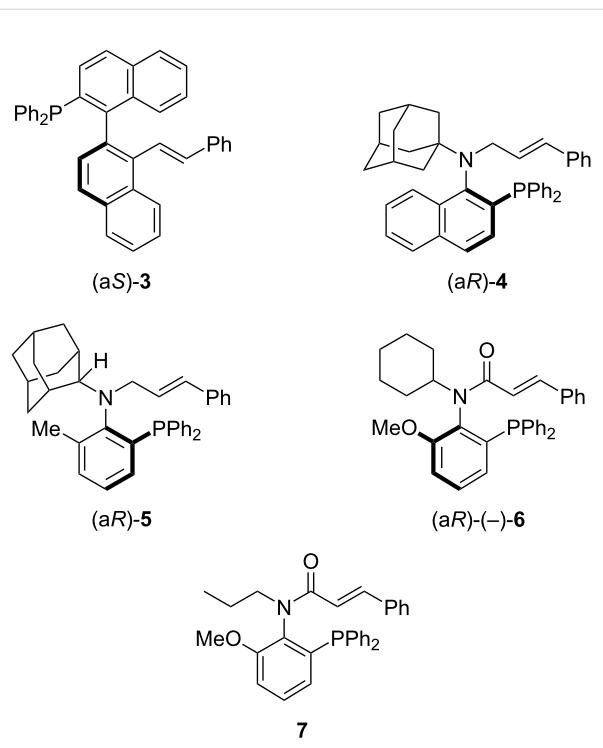
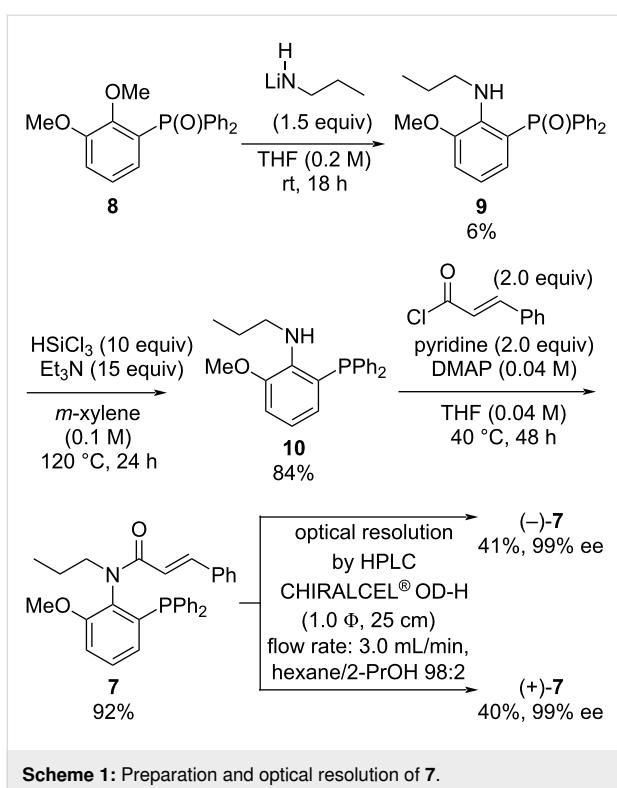


Figure 2: Chiral ligands **3**–**7**.

## Results and Discussion

*N*-Propyl-*N*-cinnamoylamide **7** was prepared from phosphine oxide **8** [32] via an  $S_NAr$  reaction with nucleophilic lithium amide from *n*-propylamine, the reduction of phosphine oxide **9** by trichlorosilane/triethylamine, and the *N*-acylation of **10** with cinnamoyl chloride in three steps (Scheme 1). We also analyzed amide compound **7** by HPLC analysis using a chiral stationary phase column with a CD detector and found that the C(aryl)–N(amide) bond axial chirality exists in amide compound **7**. We attempted the optical resolution of racemic compound  $(\pm)$ -**7** and obtained  $(+)$ -**7** and  $(-)$ -**7** using a semi-preparative chiral HPLC on 50 milligram scales. We also investigated the racemization process associated with the axial chirality of compound **7** (see Supporting Information File 1). The racemization barrier ( $\Delta G^\ddagger_{rac}$ ) of  $(-)$ -**7** in *n*-dodecane was determined to be 25.0 kcal/mol at 25 °C, as calculated using the Arrhenius and Eyring equations [33–35]. Therefore, the half-life of racemization of ligand  $(-)$ -**7** at 25 °C in *n*-dodecane is approximately 1.3 days, which is faster compared to ligand **6**, which has a half-life of about 3.7 days [31].

We next investigated the ability of optically active amides  $(aR)$ - $(-)$ -**6** and  $(-)$ -**7** as chiral ligands for the Pd-catalyzed asym-



metric allylic amination of allylic acetate, such as a 1,3-diphenyl-2-propenyl acetate (**12**) with isatin (**11a**). We began the investigation under conditions using 5 mol % of  $[\text{Pd}(\text{C}_3\text{H}_5)\text{Cl}]_2$  (Pd = 10 mol %) and 12 mol % of chiral ligands (Table 1).

The reaction with (a*R*)-(–)-**6** as the chiral ligand and  $\text{K}_2\text{CO}_3$  as the base in  $\text{CHCl}_3$  gave the desired product (*S*)-**13a** in 72% yield with 87% ee (Table 1, entry 1). In contrast, the reaction with (–)-**7** afforded (*S*)-**13a** in significantly lower yield, albeit with an enantioselectivity similar to that of the reaction with **6** (Table 1, entry 2). This result clarifies that (–)-**7**, with a racemization half-life of only approximately 1.3 days, also has a chiral induction ability. However, improvement is required in terms of the reactivity of the catalytic reaction. Subsequently, we investigated the effect of the base using (a*R*)-(–)-**6** by testing various bases. The reaction in the presence of  $\text{Na}_2\text{CO}_3$  delivered the product in 99% yield, although the enantioselectivity slightly decreased compared to the reaction using  $\text{K}_2\text{CO}_3$  (see Table 1, entry 1 vs entry 3). The use of  $\text{Cs}_2\text{CO}_3$  resulted in a significant drop in the yield (Table 1, entry 4), whereas  $\text{NaOAc}$  improved the yield but slightly lowered the enantioselectivity (Table 1, entry 5). Other potassium salts such as  $\text{K}_3\text{PO}_4$  led to a low yield

**Table 1:** Optimization of conditions for the Pd-catalyzed asymmetric allylic amination of acetate **12** with isatin (**11a**).<sup>a</sup>

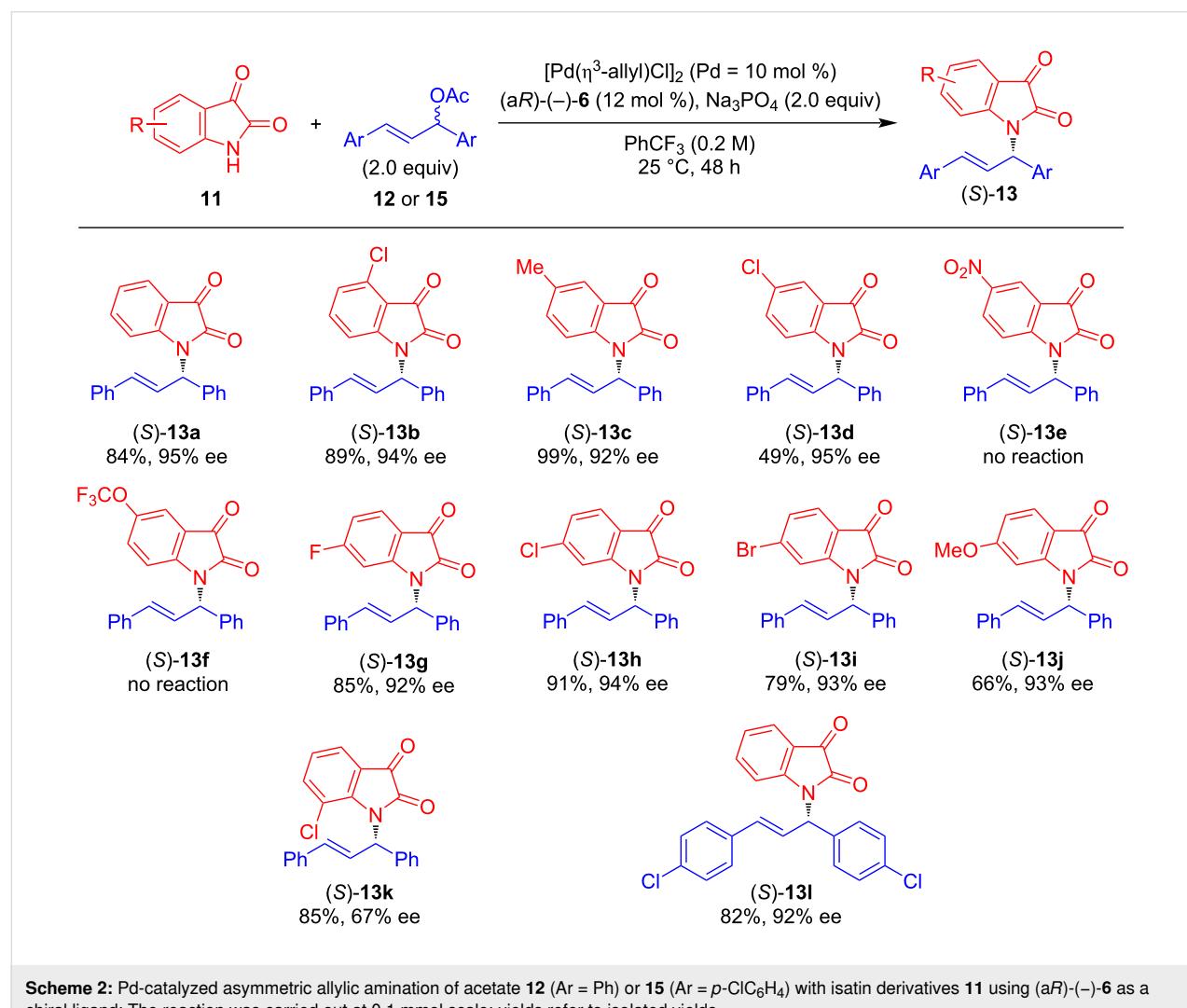
Entry	Base	Solvent	Yield (%) <sup>b</sup>	ee (%) <sup>c</sup>
1	$\text{K}_2\text{CO}_3$	$\text{CHCl}_3$	72	87
2 <sup>d</sup>	$\text{K}_2\text{CO}_3$	$\text{CHCl}_3$	3	84
3	$\text{Na}_2\text{CO}_3$	$\text{CHCl}_3$	99	85
4	$\text{Cs}_2\text{CO}_3$	$\text{CHCl}_3$	19	86
5	$\text{NaOAc}$	$\text{CHCl}_3$	89	86
6	$\text{K}_3\text{PO}_4$	$\text{CHCl}_3$	12	86
7	$\text{Na}_3\text{PO}_4$	$\text{CHCl}_3$	60	88
8	$\text{Na}_3\text{PO}_4$	$\text{CH}_2\text{Cl}_2$	88	92
9	$\text{Na}_3\text{PO}_4$	$\text{CH}_3\text{CN}$	75	93
10	$\text{Na}_3\text{PO}_4$	THF	74	93
11	$\text{Na}_3\text{PO}_4$	DMF	trace	–
12	$\text{Na}_3\text{PO}_4$	$\text{PhCF}_3$	84	95
13 <sup>e</sup>	$\text{Na}_3\text{PO}_4$	$\text{PhCF}_3$	50	86
14 <sup>f</sup>	$\text{Na}_3\text{PO}_4$	$\text{PhCF}_3$	80	94

<sup>a</sup>The reaction was carried out at 0.1 mmol scale. <sup>b</sup>Isolated yield. <sup>c</sup>Determined by chiral HPLC analysis using a chiral column. Absolute configuration was assigned by comparison of HPLC analysis with reported data [26]. <sup>d</sup>This reaction was carried out using (–)-**7** instead of (a*R*)-(–)-**6** as a chiral ligand. <sup>e</sup>This reaction was carried out using 1,3-diphenylallyl pivalate (**14**) instead of acetate **12**. <sup>f</sup>This reaction was carried out at a 1.0 mmol scale.

of the product (Table 1, entry 6). Meanwhile, when  $\text{Na}_3\text{PO}_4$  was tested, the yield decreased, but the enantioselectivity improved to 88% ee (Table 1, entry 7). With  $\text{Na}_3\text{PO}_4$  as the optimum base, which showed the highest enantioselectivity, we conducted a solvent screening. The reaction in  $\text{CH}_2\text{Cl}_2$  resulted in better yield and enantioselectivity than in  $\text{CHCl}_3$  (Table 1, entry 8). The coordinating solvents,  $\text{CH}_3\text{CN}$  and  $\text{THF}$ , further improved the enantioselectivity to 93% ee (Table 1, entries 9 and 10). In contrast, the reaction barely proceeded when  $\text{DMF}$  was used (Table 1, entry 11). The reaction in  $\text{PhCF}_3$  afforded the target product in a good yield with the highest enantioselectivity compared to other solvents (Table 1, entry 12). Furthermore, when *(E)*-1,3-diphenyl-2-propenyl pivalate (**14**) was tested as the allyl ester, the desired product (*S*)-**13a** was obtained with a yield of 50% and an enantioselectivity of 86% ee (Table 1, entry 13). Additionally, the scale-up reaction using 1 mmol of isatin (**11a**) as the nucleophile under the optimal conditions (Table 1, entry 12) afforded the desired product

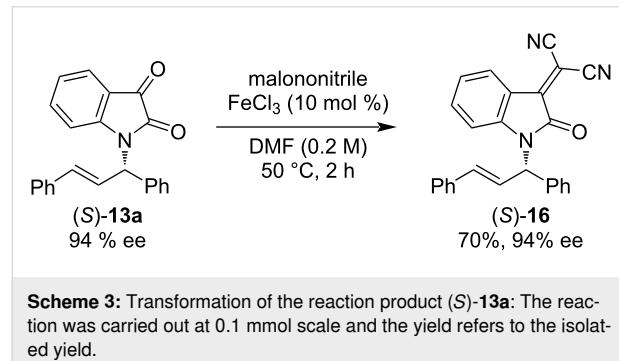
(*S*)-**13a** with nearly the same yield and enantioselectivity as the 0.1 mmol scale reaction (entry 14).

Next, we investigated the substrate scope of the palladium-catalyzed asymmetric allylic amination of 1,3-diphenyl-2-propenyl acetate (**12**) with isatin derivatives **11** as nucleophiles under the optimized conditions using (*aR*)-*(–)–6* as the ligand and  $\text{Na}_3\text{PO}_4$  as the base in  $\text{PhCF}_3$  as the solvent (Scheme 2). An isatin derivative bearing a chloro group at the 4-position afforded the desired product (*S*)-**13b** with good yield and enantioselectivity. Similarly, an isatin derivative with a methyl group as an electron-donating group at the 5-position gave (*S*)-**13c** in good yield, although with slightly decreased enantioselectivity. The introduction of the chloro group at the same position led to a moderate yield for (*S*)-**13d**, while the enantioselectivity remained high. In contrast, the reaction with the isatin derivative bearing a nitro group at the 5-position did not proceed, and (*S*)-**13e** was not produced. Likewise, no reaction occurred with



**Scheme 2:** Pd-catalyzed asymmetric allylic amination of acetate **12** ( $\text{Ar} = \text{Ph}$ ) or **15** ( $\text{Ar} = p\text{-ClC}_6\text{H}_4$ ) with isatin derivatives **11** using (*aR*)-*(–)–6* as a chiral ligand: The reaction was carried out at 0.1 mmol scale; yields refer to isolated yields.

a trifluoromethoxy-substituted derivative, resulting in no formation of *(S)*-13f. Reactions using isatin derivatives bearing halogen substituents at the 6-position proceeded efficiently, affording *(S)*-13g–i in good yields with high enantioselectivities. Conversely, the isatin derivative bearing a methoxy group at the 6-position led to a decreased yield for *(S)*-13j, though the enantioselectivity remained high. Additionally, we tested the reaction using an isatin derivative with a chloro group at the 7-position and obtained *(S)*-13k in good yield with moderate enantioselectivity. Furthermore, when *(E)*-1,3-di(*p*-chlorophenyl)-2-propenyl acetate (**15**) was utilized as an allylic acetate, the desired product *(S)*-13l was obtained in high yield with excellent enantioselectivity. We confirmed that the product **13** from the Pd-catalyzed asymmetric allylic amination of allyl esters with isatin using (aR)-(-)-6 possesses an *S*-configuration. This stereochemical outcome follows the same reaction mechanism as the Pd-catalyzed asymmetric allylic substitution of allyl esters with indoles using (aR)-(-)-6 [31]. To explore further applications of this product, we treated *(S)*-13a (94% ee) with malononitrile in the presence of FeCl<sub>3</sub> as a catalyst [36] and obtained the corresponding malononitrile derivative *(S)*-16 without any loss of optical purity (Scheme 3).



## Conclusion

In this study, *N*-propyl-*N*-cinnamoylamide **7** was synthesized in three steps from phosphine oxide **8**. Chiral HPLC analysis confirmed its axial chirality at the C(aryl)–N(amide) bond. The optical resolution of  $(\pm)$ -**7** yielded  $(+)$ -**7** and  $(-)$ -**7**. The racemization barrier of  $(-)$ -**7** in *n*-dodecane was determined to be 25.0 kcal/mol at 25 °C, with a half-life of approximately 1.3 days. The chiral amides (aR)-(-)-6 and  $(-)$ -**7** were evaluated as ligands in Pd-catalyzed asymmetric allylic amination, and while  $(-)$ -**7** exhibited promising enantioselectivity, its yield was lower than (aR)-(-)-6. Further optimization of reaction conditions led to improved yields and enantioselectivities up to 95% ee. Moreover, the reaction was successfully scaled up to 1 mmol. The substrate scope was investigated using various isatin derivatives, yielding high enantioselectivities (up to 95% ee) for most, except for those bearing certain electron-

withdrawing groups. Additionally, we demonstrated the further conversion of *(S)*-13a into the malononitrile derivative *(S)*-16 without loss of optical purity.

## Supporting Information

Data of thermal racemization of **7**, DFT calculations for investigating racemization mechanism of **7**, general methods and materials, experimental procedures and characterization data, <sup>1</sup>H, <sup>13</sup>C and <sup>31</sup>P NMR spectra for **9** and **10**, <sup>1</sup>H, <sup>13</sup>C and <sup>31</sup>P NMR spectra and HPLC charts for  $(\pm)$ -**7**,  $(+)$ -**7** and  $(-)$ -**7**, <sup>1</sup>H and <sup>13</sup>C NMR spectra and HPLC charts for *(S)*-13a–k (except *(S)*-13e) and *(S)*-16.

### Supporting Information File 1

Experimental section and compounds characterization.  
[<https://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-21-83-S1.pdf>]

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## Author Contributions

Natsume Akimoto: investigation. Kaho Takaya: investigation. Yoshio Kasashima: investigation. Kohei Watanabe: investigation. Yasushi Yoshida: investigation. Takashi Mino: conceptualization; supervision.

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## Data Availability Statement

All data that supports the findings of this study is available in the published article and/or the supporting information of this article.

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