

Synthesis of diverse dihydropyrimidine-related scaffolds by fluorous benzaldehyde-based Biginelli reaction and post-condensation modifications

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Letter

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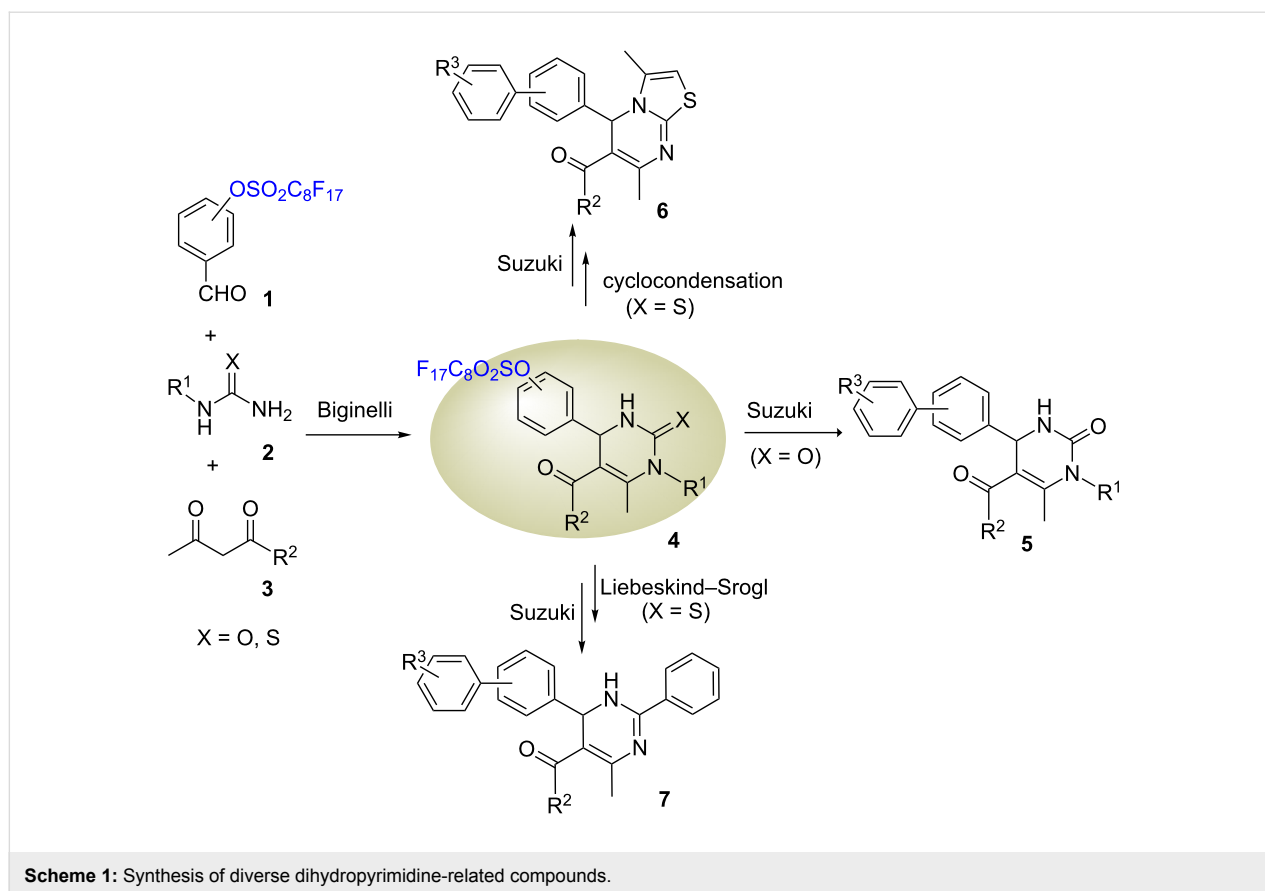
Abstract

Dihydropyrimidinones and dihydropyrimidinethiones generated from the Biginelli reactions of perfluorooctanesulfonyl-attached benzaldehydes are used as common intermediates for post-condensation modifications such as cycloaddition, Liebeskind–Srogl reaction and Suzuki coupling to form biaryl-substituted dihydropyrimidinone, dihydropyrimidine, and thiazolopyrimidine compounds. The high efficiency of the diversity-oriented synthesis is achieved by conducting a multicomponent reaction for improved atom economy, under microwave heating for fast reaction, and with fluorous solid-phase extractions (F-SPE) for ease of purification.

Introduction

Dihydropyrimidinone and dihydropyrimidine derivatives have broad biologically activities. Many synthetic samples have been studied as antibacterial, antiviral, antihypertensive, and anticancer agents [1], and the natural products containing these heterocyclic moieties have been studied as new leads for AIDS therapies [2]. The Biginelli reaction of a β -keto ester, an aldehyde, and urea is considered as one of the most efficient ways to synthesize dihydropyrimidinones [3]. This acid-catalyzed reaction can be conducted under conventional or microwave

heating [4,5]. Reported in this paper is a diversity-oriented synthesis of biaryl-substituted dihydropyrimidinone **5**, thiazolopyrimidine **6**, and dihydropyrimidine **7** compounds (Scheme 1). The perfluorooctanesulfonyl-attached benzaldehydes **1** were used as a key component for the Biginelli reactions [6]. The Biginelli products **4** were used as a common intermediate for post-condensation reactions including cycloaddition, Liebeskind–Srogl reaction and Suzuki coupling to form three different heterocyclic skeletons. The high efficiency of the



diversity-oriented synthesis was achieved by conducting fast, microwave-heated reactions and simple fluorosolid-phase extractions (F-SPE) for purification [7]. The perfluorooctanesulfonyl group served as a phase tag for F-SPE and also as a convertible linker for the Suzuki coupling to introduce biaryl functionality to the heterocyclic skeletons [8–12].

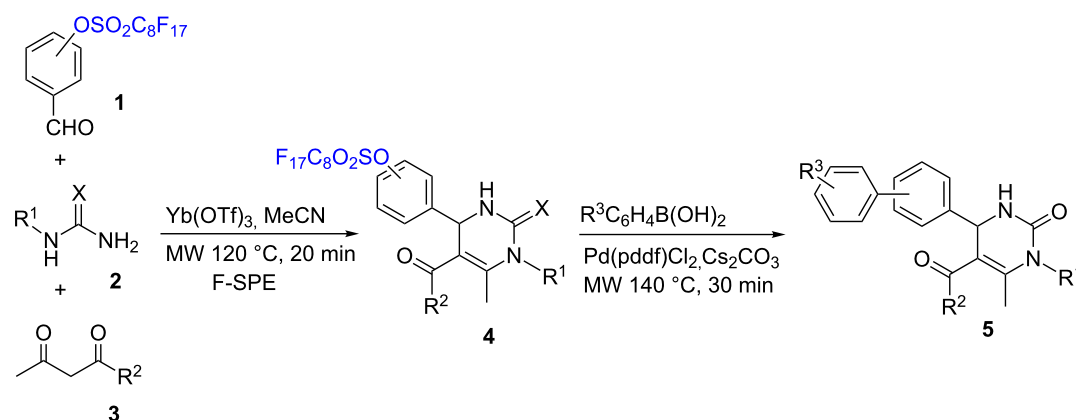
Result and Discussion

Fluorous benzaldehydes **1** were prepared by the reaction of phenols with perfluorooctanesulfonyl fluoride, by following the reported procedure [13]. Compounds **1** were used as a limiting agent to react with urea/thiourea **2** and acetylacetone **3** for the Biginelli reactions. The reactions were promoted by $\text{Yb}(\text{OTf})_3$ as a catalyst [14,15], acetonitrile as a solvent, and under microwave irradiation at 120 °C for 20 min. This optimized condition was developed after other solvents, including water, EtOH and toluene, and different microwave reaction temperatures (100–130 °C) and times (10–20 min) were explored. The Biginelli products were separated from the reaction mixtures by F-SPE eluted with fluorophobic 80:20 MeOH/H₂O and then fluorophilic 100% MeOH or acetone [7]. The fluorosolid Biginelli products were collected from the MeOH fraction to give dihydropyrimidinones **4a–d** and dihydropyrimidinethiones **4e,f** in 85–95% yields (Table 1). The Biginelli products **4a–e** were

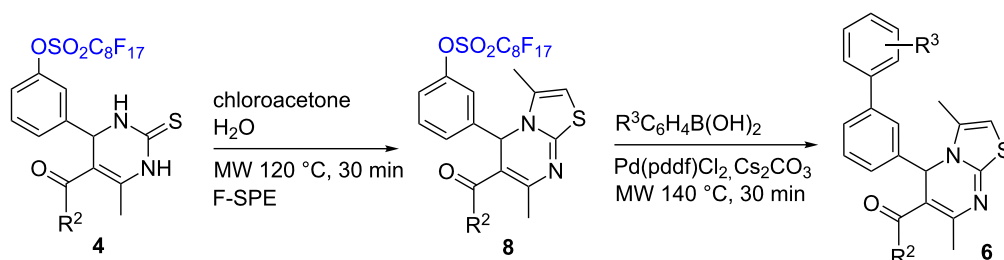
used for Suzuki coupling reactions to remove the fluorosolid linker and introduce the biaryl functional group. The coupling reactions were promoted by microwave heating at 140 °C for 30 min with $\text{Pd}(\text{pddf})\text{Cl}_2$ as a catalyst, Cs_2CO_3 as a base, and 4:4:1 acetone/toluene/H₂O as a solvent [13]. Dihydropyrimidinones **4a–d** gave the expected products **5a–h** in 51–68% yield after F-SPE and flash chromatography purification. However, no reactions occurred with the dihydropyrimidinethiones **4e,f** under these reaction conditions.

Since dihydropyrimidinethiones **4e,f** failed to give Suzuki coupling products, our next effort was to convert them to thiazolopyrimidine through cyclocondensation with chloroacetone [16,17]. The reaction was performed in water under microwave heating at 120 °C for 30 min to afford thiazolopyrimidines **8a** and **8b** in 89% and 85% yields, respectively, after F-SPE. Suzuki reactions of **8a** and **8b** with four boronic acids yielded 5-biaryl-5H-thiazolo[3,2-a]pyrimidines **6a–h** in 55–64% yields after F-SPE and flash chromatography purifications (Table 2).

Dihydropyrimidinethione **4f** was used for the Liebeskind–Srogl coupling reaction with a phenylboronic acid to convert to 2-aryl-1,6-dihydropyrimidine **9** [18–20]. The reaction was performed following a literature procedure [21] and was

Table 1: Biginelli reactions followed by Suzuki reactions of dihydropyrimidinones and dihydropyrimidinethiones.

R ¹	R ²	X	F-Sulfonyl position	4 (yield)	R ³	5 (yield)
CH ₃	CH ₃	O	<i>meta</i>	4a (91%)	<i>p</i> -OCH ₃	5a (67%)
					H	5b (56%)
CH ₃	OCH ₃	O	<i>meta</i>	4b (95%)	<i>p</i> -OCH ₃	5c (57%)
					H	5d (51%)
CH ₃	CH ₃	O	<i>para</i>	4c (90%)	<i>p</i> -OCH ₃	5e (68%)
					H	5f (62%)
CH ₃	OCH ₃	O	<i>para</i>	4d (88%)	<i>p</i> -OCH ₃	5g (58%)
					H	5h (60%)
H	CH ₃	S	<i>meta</i>	4e (89%)	H	-
H	OCH ₃	S	<i>meta</i>	4f (85%)	H	-

Table 2: Synthesis of biaryl-substituted thiazolopyrimidines.

4	R ²	8	R ³	6 (yield)
4e	CH ₃	8a (89%)	H	6a (61%)
			<i>p</i> -OCH ₃	6b (64%)
			<i>m</i> -Cl	6c (56%)
			<i>p</i> -CH ₃	6d (62%)
4f	OCH ₃	8b (85%)	H	6e (58%)
			<i>p</i> -OCH ₃	6f (55%)
			<i>m</i> -Cl	6g (63%)
			<i>p</i> -CH ₃	6h (55%)

catalyzed by Pd(PPh₃)₄ and copper(I) thiophene-2-carboxylate (CuTC) under microwave heating at 100 °C for 25 min to afford aryl-substituted dihydropyrimidine **9** in 76% yield. This compound was then subjected to Suzuki coupling reactions with four boronic acids to yield 2-aryl-6-biaryl substituted dihydropyrimidines **7a–d** after F-SPE and flash chromatography purifications (Table 3).

Conclusion

We have developed a new application of perfluorooctanesulfonyl-attached benzaldehydes for the diversity-oriented synthesis of heterocyclic scaffolds. The intermediates obtained from the Biginelli reaction were used for post-condensation modifications to afford biaryl-substituted dihydropyrimidinone, dihydropyrimidine, and thiazolopyrimidine compounds. A set of reaction and separation techniques such as multicomponent reactions, microwave heating, and F-SPE was employed to increase the synthetic efficiency. The fluorosulfonyl group not only served as a phase tag for F-SPE separation, but also as a cleavable linker for the Suzuki coupling reactions.

Experimental

Typical Biginelli reaction procedure: Synthesis of 5-acetyl-4-(4-(perfluorooctylsulfonyloxy)phenyl)-1,6-dimethyl-3,4-dihydropyrimidin-2(1*H*)-one (**4c**)

A solution of *p*-perfluorooctanesulfonyl benzaldehyde **1** (1.2 g, 2.0 mmol), methylurea **2** (0.18 g, 2.4 mmol), methyl acetoacetate **3** (0.35 g, 3.0 mmol) and Yb(OTf)₃ (124 mg, 0.2 mmol) in 2 mL of acetonitrile was heated in a Biotage Initiator microwave synthesizer at 120 °C for 20 min. The resulting mixture was purified by F-SPE eluted with 40 mL of 80:20 MeOH/

H₂O and then 40 mL of acetone. The acetone fraction was concentrated to give **4c** (1.3 g) in 90% yield.

Typical Suzuki reaction procedure: Synthesis of 5-acetyl-4-(4'-methoxy-[1,1'-biphenyl]-3-yl)-1,6-dimethyl-3,4-dihydropyrimidin-2(1*H*)-one (**5a**)

A solution of **4a** (75 mg, 0.1 mmol), 4-methoxyphenylboronic acid (23 mg, 0.15 mmol), Cs₂CO₃ (81 mg, 0.25 mmol) and Pd(dppf)Cl₂ (16 mg, 0.02 mmol) in 3 mL of 4:1:4 acetone/H₂O/toluene was heated in a Biotage Initiator microwave synthesizer at 140 °C for 30 min. The resulting mixture was purified by flash chromatography to give **5a** (24 mg) in 67% yield.

Typical procedure for cyclocondensation of **4e,f**. Synthesis of methyl 3,7-dimethyl-5-(3-(perfluorooctylsulfonyloxy)phenyl)-5*H*-thiazolo[3,2-*a*]pyrimidine-6-carboxylate (**8b**)

A solution of 3,4-dihydropyrimidinethione **4f** (0.76 g, 1 mmol), chloroacetone (185 mg, 1.5 mmol) in 2 mL water was heated in Biotage Initiator microwave synthesizer at 120 °C for 30 min. The resulting mixture was purified by F-SPE eluted with 30 mL of 80:20 MeOH/H₂O and then 30 mL of acetone. The acetone fraction was concentrated to give **8b** (0.67 g) in 85% yield.

Typical Liebeskind–Srogl reaction procedure. Synthesis of methyl 4-methyl-6-(3-(perfluorooctylsulfonyloxy)phenyl)-2-phenyl-1,6-dihydropyrimidine-5-carboxylate (**9**)

A solution of 3,4-dihydropyrimidinethione **4f** (152 mg, 0.20 mmol), phenylboronic acid (82 mg, 0.3 mmol), CuTC (95 mg, 0.6 mmol), and Pd(PPh₃)₄ (3 mol %) in 2 mL THF was heated in Biotage Initiator microwave synthesizer at 100 °C for 25

Table 3: Synthesis of 2-aryl-6-biaryl-substituted dihydropyrimidines.

R ³	7 (yield)
H	7a (45%)
<i>p</i> -OCH ₃	7b (48%)
<i>m</i> -Cl	7c (31%)
<i>p</i> -CH ₃	7d (48%)

min. The mixture was purified by F-SPE eluted with 30 mL of 80:20 MeOH/H₂O and then 30 mL of acetone. The acetone fraction was concentrated to give **9** (0.85 g) in 76% yield.

20. Prokopcová, H.; Kappe, C. O. *J. Org. Chem.* **2007**, *72*, 4440–4448. doi:10.1021/jo070408f

21. Prokopcová, H.; Kappe, C. O. *Angew. Chem., Int. Ed.* **2009**, *48*, 2276–2286. doi:10.1002/anie.200802842

Supporting Information

Supporting Information File 1

LC-MS, ¹H NMR and ¹³C NMR data and spectra for compounds **4c**, **5a**, **6b**, **7b**, **8b**, **9**.

[<http://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-7-150-S1.pdf>]

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References

- Kappe, C. O. *Acc. Chem. Res.* **2000**, *33*, 879–888. doi:10.1021/ar000048h
- Kappe, C. O. *Eur. J. Med. Chem.* **2000**, *35*, 1043–1052. doi:10.1016/S0223-5234(00)01189-2
- Biginelli, P. *Gazz. Chim. Ital.* **1893**, *23*, 360–416.
- Kappe, C. O.; Stadler, A. *Org. React.* **2004**, *63*, 1–116. doi:10.1002/0471264180.or063.01
- Kappe, C. O. The Biginelli Reaction. In *Multicomponent Reactions*; Zhu, J.; Bienaymé, H., Eds.; Wiley-VCH: Weinheim, Germany, 2005; pp 95–120.
- Studer, A.; Jeger, P.; Wipf, P.; Curran, D. P. *J. Org. Chem.* **1997**, *62*, 2917–2924.
- Zhang, W.; Curran, D. P. *Tetrahedron* **2006**, *62*, 11837–11865. doi:10.1016/j.tet.2006.08.051
- Larhed, M.; Hallberg, A. *J. Org. Chem.* **1996**, *61*, 9582–9584. doi:10.1021/jo9612990
- Larhed, M.; Hoshino, M.; Hadida, S.; Curran, D. P.; Hallberg, A. *J. Org. Chem.* **1997**, *62*, 5583–5587. doi:10.1021/jo970362y
- Kadam, A.; Zhang, Z.; Zhang, W. *Curr. Org. Synth.* **2011**, *8*, 295–309. doi:10.2174/157017911794697259
- Zhang, W. *Chem. Rev.* **2009**, *109*, 749–795. doi:10.1021/cr800412s
- Zhang, W. *Comb. Chem. High Throughput Screening* **2007**, *10*, 219–229.
- Zhang, W.; Chen, C. H.-T.; Lu, Y.; Nagashima, T. *Org. Lett.* **2004**, *6*, 1473–1476. doi:10.1021/ol0496428
- Kappe, C. O.; Dallinger, D. *Nat. Rev. Drug Discovery* **2006**, *5*, 51–63. doi:10.1038/nrd1926
- Dallinger, D.; Kappe, C. O. *Nat. Protoc.* **2007**, *2*, 1713–1721. doi:10.1038/nprot.2007.224
- Wang, X.-C.; Quan, Z.-J.; Zhang, Z.; Liu, Y.-J.; Ji, P.-Y. *Lett. Org. Chem.* **2007**, *4*, 370–373. doi:10.2174/157017807781212139
- Quan, Z.-J.; Zhang, Z.; Wang, J.-K.; Wang, X.-C.; Liu, Y.-J.; Ji, P.-Y. *Heteroat. Chem.* **2008**, *19*, 149–152. doi:10.1002/hc.20386
- Liebeskind, L. S.; Srogl, J. *J. Am. Chem. Soc.* **2000**, *122*, 11260–11261. doi:10.1021/ja005613q
- Villalobos, J. M.; Srogl, J.; Liebeskind, L. S. *J. Am. Chem. Soc.* **2007**, *129*, 15734–15735. doi:10.1021/ja074931n

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